

The thermodynamics package*

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Abstract

A package, `thermodynamics`, is defined that makes typesetting quantities found in thermodynamics texts relatively simple. The commands are flexible and intended to be relatively intuitive. It handles several sets of notation for total, specific, and molar quantities; allows changes between symbols (e.g., A vs. F for Helmholtz free energy); and greatly simplifies the typesetting of symbols and partial derivatives commonly encountered in mixture thermodynamics. Changes of one's notes from one textbook to another can be achieved relatively easily by changing package options.

1 Introduction

The purpose of this package is to simplify the typesetting of equations in thermodynamics, particularly chemical engineering thermodynamics, which are often cumbersome to enter. For example, consider the following equation:

$$d\underline{U} = \left(\frac{\partial \underline{U}}{\partial \underline{S}}\right)_{\underline{V}, \underline{n}} d\underline{S} + \left(\frac{\partial \underline{U}}{\partial \underline{V}}\right)_{\underline{S}, \underline{n}} d\underline{V} + \sum_{i=1}^C \left(\frac{\partial \underline{U}}{\partial n_i}\right)_{\underline{S}, \underline{V}, n_{j \neq i}} dn_i. \quad (1)$$

This equation is pretty basic, and equations like it occur all the time in thermodynamics. Without this package, you might typeset it like this:

```
\[ d\underline{U} =
  \left(\frac{\partial \underline{U}}{\partial \underline{S}}\right)_{\underline{V}, \underline{n}}
  d\underline{S}
+ \left(\frac{\partial \underline{U}}{\partial \underline{V}}\right)_{\underline{S}, \underline{n}}
  d\underline{V}
+ \sum_{i=1}^C \left(\frac{\partial \underline{U}}{\partial n_i}\right)_{\underline{S}, \underline{V}, n_{j \neq i}}
  dn_i. \]
```

This is a lot of code, and even then the output is slightly clunky:

$$d\underline{U} = \left(\frac{\partial \underline{U}}{\partial \underline{S}}\right)_{\underline{V}, \underline{n}} d\underline{S} + \left(\frac{\partial \underline{U}}{\partial \underline{V}}\right)_{\underline{S}, \underline{n}} d\underline{V} + \sum_{i=1}^C \left(\frac{\partial \underline{U}}{\partial n_i}\right)_{\underline{S}, \underline{V}, n_{j \neq i}} dn_i.$$

*This document corresponds to `thermodynamics` v2.03, dated 2026/01/05.

It is also frustratingly difficult to change one's notes or handouts from one textbook that uses, say, n_1 to denote moles of component 1 to another textbook that uses N_1 for the same quantity, or perhaps denotes the total internal energy as U or U^t rather than \underline{U} . For example, if you wanted it to be

$$dU = \left(\frac{\partial U}{\partial S} \right)_{V, N_1, \dots, N_C} dS + \left(\frac{\partial U}{\partial V} \right)_{S, N_1, \dots, N_C} dV + \sum_{i=1}^C \left(\frac{\partial U}{\partial N_i} \right)_{S, V, N_1, \dots, [N_i], \dots, N_C} dN_i.$$

without changing any of your code—to update it across all handouts, exams, and homework sets after changing textbooks, say—you would be out of luck (or in for a lot of work).

With this package, you could reduce the code to typeset this equation to

```
\begin{equation}
  d\Ut = \Partial*\{\Ut\}{\St}{\Vt, \allNs} d\St
  + \Partial*\{\Ut\}{\Vt}{\St, \allNs} d\Vt
  + \sumall_i \Partial*\{\Ut\}{\Nt_i}{\St, \Vt, \allNsbut{i}} d\Nt_i
\end{equation}
```

and it will render similarly to Equation (1), including the shortened underscores and negative kerning. If you later decide to change the notation such that extensive properties are not underlined, you can do that without changing any of your code (just change a package option). Similarly, if you want \vec{n} replaced by n_1, \dots, n_C , you can do that with a package option, too.

The package handles second derivatives, too. For example,

```
\[ \Partial*\{\Hm\}{T}{P} = T\Partial*\{\Sm\}{T}{P}
    = -T\PartialSecond*\{\Gm\}{T}{P} = \cP \]
```

renders (using the default options)

$$\left(\frac{\partial H}{\partial T} \right)_P = T \left(\frac{\partial S}{\partial T} \right)_P = -T \left(\frac{\partial^2 G}{\partial T^2} \right)_P = C_P.$$

Similarly, macros are defined for mixed second partial derivatives that allow things like

```
\[ \Partial*\{\Gpm_i\}{P}{T, \allNs}
    = \PartialMixSecond*\{\Gt\}{P}{\Nt_i}{T, \allNsbut{i}}
    = \PartialMixSecond*\{\Gt\}{\Nt_i}{P}{T, \allNsbut{i}}
    = \Partial*\{\Vt\}{\Nt_i}{T, P, \allNsbut{i}} = \Vpm_i \],
```

which renders

$$\left(\frac{\partial \overline{G_i}}{\partial P} \right)_{T, \vec{n}} = \left(\frac{\partial^2 \underline{G}}{\partial P \partial n_i} \right)_{T, n_{j \neq i}} = \left(\frac{\partial^2 \underline{G}}{\partial n_i \partial P} \right)_{T, n_{j \neq i}} = \left(\frac{\partial \underline{V}}{\partial n_i} \right)_{T, P, n_{j \neq i}} = \overline{V_i}$$

using the defaults.

2 Using the Package

There are three categories of macros defined in this package: macros that produce symbols (or groups of them), macros that typeset derivatives, and macros that are used internally that the user need not know about. There are also several environments that allow the user to change notation temporarily.

2.1 Predefined Symbols

The macros used to produce symbols fall into five categories: extensive properties, molar properties, specific properties (i.e., per unit mass), partial molar properties, and shortcut macros (e.g., macros for the heat capacities, saturation pressure, and so forth). The macros corresponding to extensive, molar, and specific properties are

`\Ht` shown in Table 1. Examples using the enthalpy are

```
\Hm      \[ \Ht \quad \Hm \quad \Hs \quad \Hpm_i. \]
\Hs
```

`\Hpm` Using the default package options, the above renders as

$$\underline{H} \quad H \quad \hat{H} \quad \overline{H}_i.$$

In addition, the properties in Table 2 are defined for convenience. A more comprehensive list, including how those properties are represented in the various textbook-related options supported by this package, are shown in Appendix A.

`\Ut` How these symbols are rendered can be customized by package options. As long
`\Um` as the user consistently uses `\Ut` to render the total internal energy, `\Um` to render the molar internal energy, and so forth, switching notation from, say, \underline{U} to U^t for extensive properties is trivial.

`\cP` The heat capacities (see Table 2) are generally assumed to be molar (e.g., `\cP`
`\cV` is interpreted to be the molar heat capacity). To get the specific heat capacities,
`\cPt` the macros `\cPs` and `\cVs` are provided, which by default render as \hat{C}_P and \hat{C}_V ,
`\cVt` respectively. There are also extensive versions, so `\cPt` and `\cVt` will render as \underline{C}_P
`\cPs` and \underline{C}_V , respectively. Note that `\cP` and friends require you to surround `\text` with
`\cVs` brackets. For example, `\cP_i^\text{A}` will not work, nor will `\cP^\text{A}_i`; you
`\cPpm` need to use `\cP_i^\text{A}` and `\cP^\text{A}_i`. Using `\cP^\text{IG}_i` will work
`\cVpm` as expected.

You can also get partial molar heat capacities via `\cPpm` and `\cVpm`, though the latter's mathematical definition is a bit hard to wrap one's head around:

$$\begin{aligned} \text{\[\cVpm}_i &= \text{\Partial*{\cVt}{\Nt_i}{T,P,\allNsbut{i}}}} \\ &= \text{\frac{\partial}{\partial \Nt_i} \left[-T \text{\PartialSecond{Ft}{T}{\Vt,\allNs}} \right]_{T,P,\allNsbut{i}}} \text{\]} \end{aligned}$$

yields

$$\overline{C}_{V,i} = \left(\frac{\partial \underline{C}_V}{\partial n_i} \right)_{T,P,n_{j \neq i}} = \frac{\partial}{\partial n_i} \left[-T \left(\frac{\partial^2 A}{\partial T^2} \right)_{\underline{V},\vec{n}} \right]_{T,P,n_{j \neq i}}.$$

2.2 Partial Molar Properties

`\Upm` Partial molar quantities are defined with the suffix `pm`. For example, `\Upm` refers to the
`\Vpm` partial molar internal energy. There are two options for how to enter partial molar quantities: as commands or as super/subscripts. For example,

$$\text{\[\Upm{i} \quad \Upm[IG]{i} \quad \Vpm_i \quad \Vpm^\text{IG}_i \text{\]}}$$

will typeset as

$$\overline{U}_i \quad \overline{U}_i^{\text{IG}} \quad \overline{V}_i \quad \overline{V}_i^{\text{IG}}.$$

There are also partial molar heat capacities available via the macros `\cPpm` and `\cVpm`.

Important: The `\text` command defined by the `amstext` package is usually robust

Table 1. Commands defined in this package to represent extensive thermodynamic quantities and their molar and specific analogs. These macros should be used even if the symbol the user wishes to use does not match the command used (e.g., `\Ft` for total Helmholtz free energy even if it ends up being set as \underline{A}).

Property	Total	Molar	Specific	Partial Molar	Excess				Residual (Departure)			
					T	M	S	PM	T	M	S	PM
Heat	<code>\Qt</code>	<code>\Qm</code>	<code>\Qs</code>									
Work	<code>\Wt</code>	<code>\Wm</code>	<code>\Ws</code>									
Total energy	<code>\Et</code>	<code>\Em</code>	<code>\Es</code>	<code>\Epm</code>	<code>\EEt</code>	<code>\EE</code>	<code>\EEs</code>	<code>\EEpm</code>	<code>\ERt</code>	<code>\ER</code>	<code>\ERs</code>	<code>\ERpm</code>
Internal energy	<code>\Ut</code>	<code>\Um</code>	<code>\Us</code>	<code>\Upm</code>	<code>\UEt</code>	<code>\UE</code>	<code>\UES</code>	<code>\UEpm</code>	<code>\URt</code>	<code>\UR</code>	<code>\URs</code>	<code>\URpm</code>
Enthalpy	<code>\Ht</code>	<code>\Hm</code>	<code>\Hs</code>	<code>\Hpm</code>	<code>\HEt</code>	<code>\HE</code>	<code>\HEs</code>	<code>\HEpm</code>	<code>\HRt</code>	<code>\HR</code>	<code>\HRs</code>	<code>\HRpm</code>
Entropy	<code>\St</code>	<code>\Sm</code>	<code>\Ss</code>	<code>\Spm</code>	<code>\SEt</code>	<code>\SE</code>	<code>\SEs</code>	<code>\SEpm</code>	<code>\SRt</code>	<code>\SR</code>	<code>\SRs</code>	<code>\SRpm</code>
Volume	<code>\Vt</code>	<code>\Vm</code>	<code>\Vs</code>	<code>\Vpm</code>	<code>\VEt</code>	<code>\VE</code>	<code>\VEs</code>	<code>\VEpm</code>	<code>\VRt</code>	<code>\VR</code>	<code>\VRs</code>	<code>\VRpm</code>
Helmholtz free energy	<code>\Ft</code>	<code>\Fm</code>	<code>\Fs</code>	<code>\Fpm</code>	<code>\FEt</code>	<code>\FE</code>	<code>\FEs</code>	<code>\FEpm</code>	<code>\FRt</code>	<code>\FR</code>	<code>\FRs</code>	<code>\FRpm</code>
Gibbs free energy	<code>\Gt</code>	<code>\Gm</code>	<code>\Gs</code>	<code>\Gpm</code>	<code>\GET</code>	<code>\GE</code>	<code>\GES</code>	<code>\GEpm</code>	<code>\GRt</code>	<code>\GR</code>	<code>\GRs</code>	<code>\GRpm</code>
Surface area	<code>\At</code>	<code>\Am</code>	<code>\As</code>	<code>\Apm</code>								
Grand potential ^a	<code>\Lt</code>	<code>\Lm</code>	<code>\Ls</code>	<code>\Lpm</code>	<code>\LEt</code>	<code>\LE</code>	<code>\LEs</code>	<code>\LEpm</code>	<code>\LRt</code>	<code>\LR</code>	<code>\LRs</code>	<code>\LRpm</code>
Moles	<code>\Nt</code>											

^aThe grand potential, $\underline{\Omega}(T, \underline{V}, \mu_1, \dots, \mu_C) = \underline{U} - T\underline{S} - \sum_{i=1}^C \mu_i n_i$, is also called the Landau free energy by some authors.

enough that something like `\Um^{\text{L}}` will work as expected, without additional braces. This does *not* work for partial molar properties; for example, `\Hpm^{\text{L}}_i` will produce an error, as will `\Hpm_i^{\text{L}}`. The expression `\Hpm_i^{\text{L}}` will work as expected.

`\partialmolar` New partial molar properties can be defined for any “simple” symbol using the `\partialmolar` macro. “Simple” means it has no subscripts or superscripts. For example, the macro for the partial molar Gibbs free energy is effectively defined via `\Gpm` the macro

```
\NewDocumentCommand{\Gpm}{}{\partialmolar{G}}.
```

A list of pre-defined macros for total, molar, specific, and partial molar quantities commonly used in thermodynamics is included in Table 1.

2.3 Defining New Properties

`\NewExtensiveProperty` Users can create new properties using a family of commands. Typically, a user would want to define at least the total, molar, and specific properties, which can be accomplished by the `\NewExtensiveProperty` command. For example,

```
\NewExtensiveProperty{J}{K}
```

would define the commands `\Jt`, `\Jm`, and `\Js`, which would produce (using the default package options) the symbols \underline{K} , K , and \hat{K} , respectively.

`\NewPartialMolarProperty` Partial molar properties can be created either with the `\partialmolar` macro directly as described above or with `\NewPartialMolarProperty`, which has the same argument style as `\NewExtensiveProperty`.

`\NewExcessProperty` Similarly, one can define commands for the total, molar, and specific excess properties using `\NewExcessProperty` in a similar manner, and similar commands for the residual properties with `\NewResidualProperty`.

`\NewThermodynamicProperty` It is common that a user wants the total, molar, specific, and partial molar commands for a new symbol, as well as excess and residual (departure) properties for each

Table 2. Convenience macros and their default symbols. These are generally “smart”: for example, `\cPi` renders as $C_{P,i}$, as expected, and `\cPi°` renders as $C_{P,i}^{\circ}$, also as expected. You can also reverse it: `\cP°` becomes $C_{P,i}^{\circ}$.

Name	Macro	Sym.	Definition	Base Symbol Macro
Isobaric heat capacity	<code>\cP^a</code>	C_P	$T \left(\frac{\partial S}{\partial T} \right)_P$	<code>\heatcapacitysymbol</code>
Isochoric heat capacity	<code>\cV^a</code>	C_V	$T \left(\frac{\partial S}{\partial T} \right)_V$	<code>\heatcapacitysymbol</code>
Isothermal compressibility	<code>\kappa_T</code>	κ_T	$-\frac{1}{V} \left(\frac{\partial V}{\partial P} \right)_T$	<code>\compressibilitysymbol</code>
Isentropic compressibility	<code>\kappa_S</code>	κ_S	$-\frac{1}{V} \left(\frac{\partial V}{\partial P} \right)_S$	<code>\compressibilitysymbol</code>
Isobaric expansivity	<code>\alpha_P</code>	α_P	$\frac{1}{V} \left(\frac{\partial V}{\partial T} \right)_P$	<code>\expansivitysymbol</code>
Isentropic expansivity	<code>\alpha_S</code>	α_S	$\frac{1}{V} \left(\frac{\partial V}{\partial T} \right)_S$	<code>\expansivitysymbol</code>
Joule–Thomson coeff.	<code>\mu_{JT}</code>	μ_{JT}	$\left(\frac{\partial T}{\partial P} \right)_H$	<code>\JTsymbol</code>
Pure fugacity	<code>\fpure</code>	f	ϕP	
Saturation fugacity	<code>\fsat</code>	f^{sat}	$\phi^{\text{sat}} P^{\text{sat}}$	<code>\sat</code>
Mixture fugacity	<code>\fmix</code>	\hat{f}	$\hat{f}_i = x_i \hat{\phi}_i P$	
Pure fugacity coefficient	<code>\phipure</code>	ϕ	$\phi_i = \exp \left(\frac{1}{RT} \int_0^P V_i(T, p) - \frac{RT}{p} dp \right)$	
Saturation fugacity coeff.	<code>\phisat</code>	ϕ^{sat}	$\phi(T, P^{\text{sat}})$	<code>\sat</code>
Mixture fugacity coeff.	<code>\phimix</code>	$\hat{\phi}$	$\hat{\phi}_i = \exp \left(\frac{1}{RT} \int_0^P \bar{V}_i(T, p, \vec{x}) - \frac{RT}{p} dp \right)$	
Absolute activity	<code>\actabs</code>	λ	$\lambda_i = \exp(\mu_i/RT)$	
Relative activity	<code>\actrel</code>	a	$a_i = \exp \left(\frac{\mu_i - \mu_i^{\circ}}{RT} \right) = \frac{\hat{f}_i(T, P, \vec{x})}{f_i^{\circ}(T)}$	
Henry’s constant (rational)	<code>\Henryrat</code>	h	$\gamma_i^{\infty} f_i$	
Henry’s constant (molal)	<code>\Henrymol</code>	\mathcal{H}	$M_s \gamma_i^{\infty} f_i$	
Rational activity coeff.	<code>\gamarat</code>	γ^*	γ/γ^{∞}	
Molal activity coeff.	<code>\gammamol</code>	γ^{\square}	$x_s \gamma/\gamma^{\infty}$	
Saturation pressure	<code>\Psat^b</code>	P^{sat}		<code>\sat</code>
Standard state	<code>\std^c</code>			
Standard pressure	<code>\Pstd</code>	P°		<code>\std</code>
Standard molality	<code>\Cstd</code>	C°		<code>\std</code>
Standard fugacity	<code>\fstd</code>	f°	$f_i^{\circ} = f_i(T, P^{\circ}, x^{\circ})$	<code>\std</code>
Standard chemical potential	<code>\mustd</code>	μ°	$\mu_i^{\circ} = \mu(T, P^{\circ}, x^{\circ})$	<code>\std</code>
Change on mixing	<code>\Deltamix^d</code>	ΔM_{mix}	$M - \sum_{i=1}^C x_i M_i$	<code>\mixing</code>
Change on reaction	<code>\Deltarxn^d</code>	ΔM_{rxn}	$\sum_{i=1}^C \nu_i M_i$	<code>\reaction</code>
Change on melting	<code>\Delta_{tafus}^d</code>	ΔM^{fus}	$M^L - M^S$	<code>\fusion</code>
Change on boiling	<code>\Delta_{tavap}^d</code>	ΔM^{vap}	$M^V - M^L$	<code>\vaporization</code>
Change on subliming	<code>\Delta_{tasub}^d</code>	ΔM^{sub}	$M^V - M^S$	<code>\sublimation</code>
Electric potential	<code>\Epot</code>	E	$-n_e \mathcal{F} \Delta G_{\text{rxn}}$	
Extent of reaction	<code>\xrxn</code>	ξ		

^aExtensive and specific (per-unit-mass) versions are available as `\cPt` and `\cPs`, respectively, with similar macros for the isochoric heat capacity.

^b`\Pvap` is a synonym for `\Psat`; if you want it to produce P^{vap} , you should redefine the `\sat` macro

^cTypical usage: `\Deltarxn\Gm^{\circ}\std = \sum_{i=1}^C \nu_i \Delta G_{f,i}^{\circ}` yielding $\Delta G_{\text{rxn}}^{\circ} = \sum_{i=1}^C \Delta G_{f,i}^{\circ}$

^dThe usual usage would be something like `\Deltamix\Vm^{\circ}\IGM = 0$`.

case. Users can define such properties—common examples are \underline{B} and \underline{M} to represent uncommon or arbitrary properties—by using `\NewThermodynamicProperty`, which calls all four of the aforementioned declarations on the same command/symbol combinations. For example,

```
\NewThermodynamicProperty{B}{B}
```

defines the commands `\Bt`, `\Bm`, `\Bs`, and `\Bpm`, which define the total, molar, specific, and partial molar properties, respectively. It also defines `\BEt`, `\BE`, `\BEs`, and `\BEpm` for the corresponding excess properties, and `\BRt`, `\BR`, `\BRs`, and `\BRpm` for residual (departure) properties. These produce, respectively, \underline{B} , B , \hat{B} , \bar{B}_i , \underline{B}^E , B^E , \hat{B}^E , \bar{B}_i^E , \underline{B}^R , B^R , \hat{B}^R , and \bar{B}_i^R using the defaults.

2.4 Other Predefined Symbols and Modifiers

There are a number of predefined symbols and modifiers. While these symbols could be defined or used without these macros, such use is not recommended: changing package options will result in inconsistencies if these macros are not used.

2.4.1 Heat Capacities, Compressibilities, and Expansivities

`\cP` The isobaric and isochoric heat capacities are produced with `\cP` and `\cV`, respectively. Four other measurable quantities are defined: the isothermal and isentropic compressibilities, `\kappaT` and `\kappaS`, respectively; and the isobaric and isentropic volume expansivities, `\alphaP` and `\alphaS`, respectively. Some textbooks use β instead of α for the volume expansivity to differentiate it from the *linear* expansivity; this can be changed by redefining `\expansivitysymbol`, which is done automatically by some of the package options that create notation specific to a particular textbook.

2.4.2 Joule–Thomson Coefficient

`\muJT` The Joule–Thomson coefficient is produced with `\muJT`, which by default is rendered μ_{JT} . Some books call this coefficient α_H ; this is handled automatically for books of which the package author is aware.

2.4.3 Chemical Potentials

`\mu` The chemical potential is always produced by the macro `\mu`. Some textbook-related options use μ instead of μ for chemical potential; if so, the macro `\mu` will always produce μ .

2.4.4 Fugacities and Fugacity Coefficients

`\fpure` Different textbooks use different variations on the symbol f for fugacity, so it is recommended to use the macro `\fpure` to denote the pure-component fugacity and `\fmix` to denote the mixture fugacity. Similarly, the pure-component fugacity coefficient `\phipure` should be generated with `\phipure`, and that in a mixture should be `\phimix`.

For example, the following markup is an example of a common equation in mixture thermodynamics:

$$\begin{aligned} \left[\text{\fmix}_j = x_j \text{\phimix}_j P = x_j \gamma_j \text{\fpure}_j \right. \\ \left. = x_j \gamma_j \text{\phipure}_j P. \right] \end{aligned}$$

With the default package options, this produces

$$\hat{f}_j = x_j \hat{\phi}_j P = x_j \gamma_j f_j = x_j \gamma_j \phi_j P.$$

With the Thompson package option, however, the same markup produces

$$\hat{f}_j = x_j \hat{\phi}_j P = x_j \gamma_j f_j^\bullet = x_j \gamma_j \phi_j^\bullet P.$$

Similarly, the Prausnitz package option causes it to generate

$$f_j = x_j \phi_j P = x_j \gamma_j f_{\text{pure},j} = x_j \gamma_j \phi_{\text{pure},j} P,$$

and the Sandler option causes it to generate

$$\bar{f}_j = x_j \bar{\phi}_j P = x_j \gamma_j f_j = x_j \gamma_j \phi_j P.$$

2.4.5 Activity Coefficients and Henry's Constants

`\gamma` The activity coefficient based on the Lewis–Randall rule can be generated with `\gamma`, `\gammaait` as usual. The symbol `\gammaait` is provided just in case `\gamma` has been redefined by a package option and you want to use the original symbol for some reason. The `\gammarat` Henry's Law activity coefficients should be produced with `\gammarat` (rational basis) and `\gammamol` (molal basis). There are also macros to generate the Henry's law constants for both the rational basis (`\Henryrat`) and the molal basis (`\Henrymol`). These are interrelated:

$$\begin{aligned} \text{\texttt{[\texttt{\fmix_i} = \texttt{x_i} \texttt{\gamma_i} \texttt{\fpure_i} = \texttt{x_i} \texttt{\gammarat_i} \texttt{\Henryrat_i} \\ = \texttt{C_i} \texttt{\gammamol_i} \texttt{\Henrymol_i} \texttt{]}} \end{aligned}$$

produces

$$\hat{f}_i = x_i \gamma_i f_i = x_i \gamma_i^* h_i = C_i \gamma_i^\square \mathcal{H}_i$$

using the default options. These symbols can be customized either directly or by using package options. For example, the `TesterModell` package option changes the equation above to

$$\hat{f}_i = x_i \gamma_i f_i = x_i \gamma_i^{**} f_i^{**} = C_i \gamma_i^* f_i^*$$

without any changes in markup.

Note that some books (e.g., `ElliottLira`) prefer the atmospheric chemistry convention for the molal basis, namely

$$\hat{f}_i = x_i \gamma_i f_i = x_i \gamma_i^* h_i = C_i \gamma_i^\square / K_{H,i}$$

(that is, using $1/K_{H,i}$ in place of \mathcal{H}_i). Unfortunately, this makes it impossible to change symbols without any edits to markup, so this convention is *not* supported by this package.

2.4.6 Saturation Properties

`\Psat` The saturation pressure is generated with `\Psat`. The macro `\Pvap` is an alias for `\Psat`. The fugacity and fugacity coefficient at saturation are accessed via `\fsat` and `\phisat`, respectively. Package options can be used to change some of these to match the notation of specific textbooks.

`\sat` The `\sat` macro is used “behind the scenes” as part of `\Psat`, `\fsat`, and `\phisat`, which produce P^{sat} , f^{sat} , and ϕ^{sat} , respectively (using the defaults). If you wanted to redefine them to be P^{vap} , f^{vap} , and ϕ^{vap} , you could simply redefine `\sat` with `\RenewDocumentCommand{\sat}{*}{\text{vap}}`. This is done automatically using package option `Sandler`.

2.4.7 Standard States and Chemical Reactions

The symbol \circ (`\circ`) is used by default for standard states. This is intended to be easy to change should the user want to replace P° with P^\ominus , say. This is accessed via the `\std` macro. The macro `\Pstd` is defined as `P^\std` for convenience to denote standard pressure, `\Cstd` is defined as `C^\std` to denote standard molality, `\fstd` is defined for standard fugacities, and `\mustd` is defined for standard chemical potentials so as to ease implementation across textbooks.

The usual usage would be something like this:

```
\[ \Deltarxn\Gm = \sumall_i \nu_i \mu_i = \sumall_i \left[ \nu_i \mustd_i
+ \nu_i RT \log\left(\frac{fmix_i}{fstd_i}\right) \right]
= \Deltarxn\Gm^\std + RT \log\left[\prodall_i a_i^{\nu_i} \right] \]
```

which produces

$$\Delta G_{\text{rxn}} = \sum_{i=1}^C \nu_i \mu_i = \sum_{i=1}^C \left[\nu_i \mu_i^\circ + \nu_i RT \log \left(\frac{\hat{f}_i}{f_i^\circ} \right) \right] = \Delta G_{\text{rxn}}^\circ + RT \log \left[\prod_{i=1}^C a_i^{\nu_i} \right]$$

with the default settings.

The extent of reaction is generated by the macro `\xxrxn`, which by default is printed as ξ .

2.4.8 Changes on Mixing, Reaction, Fusion, Vaporization, and Sublimation

Mixing properties are handled via the `\Deltamix` macro, and are used as in the following example:

```
\[ \Deltamix\Gm = \Gm - \sumall_i x_i \Gm_i
= \Deltamix\Hm - T\Deltamix\Sm \]
```

which yields, using the default options,

$$\Delta G_{\text{mix}} = G - \sum_{i=1}^C x_i G_i = \Delta H_{\text{mix}} - T \Delta S_{\text{mix}}$$

Some textbooks (Sandler, Thompson) choose to typeset these with the word “mix” before the symbol, which is handled automatically by this package. The macro `\mixing` determines how the change in mixing label is rendered; the default is `\text{mix}`.

The commands `\Deltafus`, `\Deltasub`, and `\Deltavap` typeset changes due to fusion (melting), sublimation (subliming), and vaporization (boiling), respectively.

Their use is straightforward, viz.,

```
\[ \Deltasub\Hm = \Hm^V - \Hm^S = \Hm^V - \Hm^L + (\Hm^L - \Hm^S)
= \Deltafus\Hm + \Deltavap\Hm \]
```

yielding

$$\Delta H^{\text{sub}} = H^V - H^S = H^V - H^L + (H^L - H^S) = \Delta H^{\text{fus}} + \Delta H^{\text{vap}}$$

with the default options. Note that some textbooks (e.g., Sandler) typeset these quantities quite differently; this is handled automatically. Other textbooks (e.g., Koretsky) typeset them as subscripts; this is also handled automatically.

The macros `\Deltaf` and `\Deltarxn` are intended to typeset the enthalpy or free energy of formation and reaction, respectively. For example,

```
\[ \Deltarxn\Hm^\std = \sumall_i \nu_i \Deltaf\Hm_i^\std \]
```


results in

$$\Delta H_{\text{rxn}}^{\circ} = \sum_{i=1}^C \nu_i \Delta H_{f,i}^{\circ}.$$

It is not anticipated that this command will be combined with something like a heat capacity, which already has a (potentially double) subscript, but as there is no “formation” heat capacity, that should not present a problem.

2.5 Residual and Excess Properties

Additional macros are defined that make it easy to typeset the residual (also called “departure”) and excess total, molar, specific, and partial molar properties. These macros follow the same pattern: `\UR`, `\URt`, `\URs`, and `\URpm` typeset the molar, total, specific, and partial molar residual internal energies, respectively, and by default expand to U^R , \underline{U}^R , \hat{U}^R , and \underline{U}_i^R (the last is called as `\URpm{i}` or `\URpm_i`). Similarly, `\UEt`, `\UE`, `\UEt`, `\UES`, and `\UEpm` typeset the corresponding excess properties. The first character of the macros for other properties follow the same pattern as in Table 1.

The R and E characters are generated by the macros `\residual` and `\excess`, respectively. These macros can be redefined; for example, if you want `\SE`, which normally produces S^E , to give you S^{EX} —and let’s be honest, who doesn’t want that?¹—then you can redefine it with

```
\RenewDocumentCommand{\excess}{}{\{EX\}},
```

or possibly

```
\RenewDocumentCommand{\excess}{}{\{\mathrm{EX}\}}
```

or even

```
\RenewDocumentCommand{\excess}{}{\text{EX}},
```

which cause `\SE` to expand to S^{EX} , S^{EX} , and S^{EX} , respectively.

It is generally possible to use superscripts with the excess or residual properties; in the event this fails, the `\excess` and `\residual` macros can be used directly, viz.,

```
\begin{gather*}
\HE^{\std} = \Hm^{\excess,\std} = \HE(T,\Pstd)
= \HR(T,\Pstd) - \HR^{\IS}(T,\Pstd) = \HR^{\std} - \HR^{\IS,\std} \\
\RenewDocumentCommand{\excess}{}{\{EX\}}
\HE^{\std} = \Hm^{\excess,\std} = \HE(T,\Pstd)
= \HR(T,\Pstd) - \HR^{\IS}(T,\Pstd)
\end{gather*},
```

which yields

$$H^{E,\circ} = H^{E,\circ} = H^E(T, P^\circ) = H^R(T, P^\circ) - H^{R,IS}(T, P^\circ) = H^{R,\circ} - H^{R,IS,\circ}$$

$$H^{EX,\circ} = H^{EX,\circ} = H^{EX}(T, P^\circ) = H^R(T, P^\circ) - H^{R,IS}(T, P^\circ)$$

using the default options.

2.6 Partial Derivatives

`\Partial` Partial derivatives are easily rendered using the `\Partial` command. There is a `\Partial*` starred form (`\Partial*`) that additionally adjusts the spacing after the closing symbol to remove some of the space, anticipating that the following binary operator will overhang the subscripts. Compare the following:

`\[\Partial{\Hm}{T}{P} = \cP \quad \Partial*{\Hm}{T}{P} = \cP \]`,

which yields

$$\left(\frac{\partial H}{\partial T}\right)_P = C_P \quad \left(\frac{\partial H}{\partial T}\right)_P = C_P.$$

`\Partialinline` Inline first derivatives² can be entered the same way; compare:

`\[\Partial*{\Hm}{T}{P} = \Partialinline{\Hm}{T}{P}`
`= T\Partial{\Sm}{T}{P} \]`,

which results in

$$\left(\frac{\partial H}{\partial T}\right)_P = (\partial H/\partial T)_P = T (\partial S/\partial T)_P.$$

There is no need for an inline starred form, as the subscripts do not extend far enough below the baseline.

2.6.1 Second-Order Partial Derivatives

`\PartialSecond` Second partial derivatives and mixed-second partial derivatives are typeset with the `\PartialSecond` and `\PartialMixSecond`, respectively. Like the first-order variety, these also have starred versions that remove the space immediately following the closing symbols, anticipating that the equals sign or other binary operator following the derivative will overhang the elements held constant. For example,

`\[\Vpm_i = \Partial*{\Vt}{\Nt_i}{T,P,\allNsbut{i}}`
`= \PartialMixSecond*{\Gt}{\Nt_i}{P}{T,\allNsbut{i}}`
`= \PartialMixSecond*{\Gt}{P}{\Nt_i}{T,\allNsbut{i}}`
`= \Partial{\Gpm_i}{P}{T,\allNs} \]`

looks like

$$\overline{V}_i = \left(\frac{\partial \underline{V}}{\partial n_i}\right)_{T,P,n_{j \neq i}} = \left(\frac{\partial^2 \underline{G}}{\partial n_i \partial P}\right)_{T,n_{j \neq i}} = \left(\frac{\partial^2 \underline{G}}{\partial P \partial n_i}\right)_{T,n_{j \neq i}} = \left(\frac{\partial \overline{G}_i}{\partial P}\right)_{T,\vec{n}}.$$

`\PartialSecondinline` Inline versions² of second-order derivatives are handled with `\PartialSecondinline` and `\PartialMixSecondinline`, viz.,

```
\begin{equation}
\overline{Vpm}_i = \Partialinline{\Vt}{\Nt_i}{T,P,\allNsbut{i}}
= \PartialMixSecondinline{\Gt}{\Nt_i}{P}{T,\allNsbut{i}}
= \PartialMixSecondinline{\Gt}{P}{\Nt_i}{T,\allNsbut{i}}
= \Partialinline{\Gpm_i}{P}{T,\allNs} \]
\end{equation}
```

¹You knew that joke was coming.

²The “inline” versions of partial derivatives are “sticky”: if you issue `\Partialinline` or its second-order equivalents anywhere in a line, all subsequent `\Partial` and `\Partial[Mix]Second` macros on the same line (technically, anywhere in the same TeX “group”) will expand inline as well. To prevent this, enclose your `\Partialinline` and associated arguments in its own group (i.e., `\Partialinline{...}`).

looks like

$$\overline{V}_i = \left(\frac{\partial \underline{V}}{\partial n_i} \right)_{T,P,n_{j \neq i}} = \left(\frac{\partial^2 \underline{G}}{\partial n_i \partial P} \right)_{T,n_{j \neq i}} = \left(\frac{\partial^2 \underline{G}}{\partial P \partial n_i} \right)_{T,n_{j \neq i}} = \left(\frac{\partial \overline{G}_i}{\partial P} \right)_{T,\vec{n}} \quad (2)$$

with the default options.

2.6.2 Delimiter Sizing

There are instances (such as the line above) when `\Partial` causes parentheses that are slightly too tall but do not need to be—particularly when partial molar properties, specific quantities, or fugacities are involved. The macro `\PartialBigg` uses `amsmath`’s `\Biggl` and `\Biggr` macros in place of `\left` and `\right` to size the delimiters accordingly; `\Partialbigg` uses `\biggl` and `\biggr` in a similar fashion. For example, compare the following:

```
\[ \Vpm_i = \Partial*{\Vt}{\Nt_i}{T,P,\allNsbut{i}}
      = \Partial*{\Gpm_i}{P}{T,\allNs}
      = \PartialBigg*{\Gpm_i}{P}{T,\allNs}
      = RT\Partial*{\log\fmix_i}{P}{T,\allNs}
      = RT\PartialBigg*{\log\fmix_i}{P}{T,\allNs}
      = RT\Partialbigg{\log\fmix_i}{P}{T,\allNs} \]
```

which typesets as

$$\overline{V}_i = \left(\frac{\partial \underline{V}}{\partial n_i} \right)_{T,P,n_{j \neq i}} = \left(\frac{\partial \overline{G}_i}{\partial P} \right)_{T,\vec{n}} = \left(\frac{\partial \overline{G}_i}{\partial P} \right)_{T,\vec{n}} = RT \left(\frac{\partial \log \hat{f}_i}{\partial P} \right)_{T,\vec{n}} = RT \left(\frac{\partial \log \hat{f}_i}{\partial P} \right)_{T,\vec{n}} = RT \left(\frac{\partial \log \hat{f}_i}{\partial P} \right)_{T,\vec{n}}.$$

Note that a similar effect—possibly with other side effects—can be achieved with `amsmath`’s `\smash` command, which has the effect of removing all vertical space associated with a particular character. Observe:

```
\[ \Vpm_i = \Partial*{\Vt}{\Nt_i}{T,P,\allNsbut{i}}
      = \Partial{\smash{\Gpm_i}}{P}{T,\allNs} \]
```

produces

$$\overline{V}_i = \left(\frac{\partial \underline{V}}{\partial n_i} \right)_{T,P,n_{j \neq i}} = \left(\frac{\partial \overline{G}_i}{\partial P} \right)_{T,\vec{n}}.$$

Similarly, there are times when `\Partialinline` causes parentheses that are too big for inline text, and they do not need to be—particularly for symbols with overlines, underlines, or other decorations. In this case, the macros `\Partialinlinetext`, `\PartialSecondinlinetext`, and `\PartialMixSecondinlinetext` come in handy; using these macros like so,

```
\[ \Vpm_i = \Partialinlinetext{\Vt}{\Nt_i}{T,P,\allNsbut{i}}
      = \PartialMixSecondinlinetext{\Gt}{\Nt_i}{P}{T,\allNsbut{i}}
      = \PartialMixSecondinlinetext{\Gt}{P}{\Nt_i}{T,\allNsbut{i}}
      = \Partialinlinetext{\Gpm_i}{P}{T,\allNs} \]
```

Equation (2) looks like

$$\overline{V}_i = (\partial \underline{V} / \partial n_i)_{T,P,n_{j \neq i}} = (\partial^2 \underline{G} / \partial n_i \partial P)_{T,n_{j \neq i}} = (\partial^2 \underline{G} / \partial P \partial n_i)_{T,n_{j \neq i}} = (\partial \overline{G}_i / \partial P)_{T,\vec{n}}.$$

2.6.3 Higher-Order Derivatives

It is possible to “fake” higher-order derivatives via some trickery. For example,

```
\[ \cPpm_i = T\Partial*\{\Spm_i\}{T}\{P,\allXs\}
      = T\PartialMixSecond*\{\St\}{T}\{Nt_i\}{P,\allNsbut{i}}
      = -T\Partial{\^3\Gt}\{T^2\partial\Nt_i\}{P,\allNsbut{i}} \]
```

gives

$$\overline{C_{P,i}} = T \left(\frac{\partial \overline{S_i}}{\partial T} \right)_{P,\vec{x}} = T \left(\frac{\partial^2 \overline{S}}{\partial T \partial n_i} \right)_{P,n_{j \neq i}} = -T \left(\frac{\partial^3 \overline{G}}{\partial T^2 \partial n_i} \right)_{P,n_{j \neq i}},$$

which is probably pretty close to what you wanted. Using this trickery with the package option `nosubscripts` will not work, and the use of third- and higher-order derivatives with this package should generally be considered unsupported.

2.7 Holding Constant the Number of Moles of Several Species

```
\allNs It is common in thermodynamics to use notation such as
\allNsbut
\allmus
\allmusbut
\allXs
\allXsbut or perhaps
\allYs
\allYsbut
\allMs
\allMsbut or
\allWs
\allWsbut
```

$$\overline{V}_k = \left(\frac{\partial \overline{V}}{\partial n_k} \right)_{T,P,n_{j \neq k}},$$

$$\overline{V}_k = \left(\frac{\partial \overline{V}}{\partial n_k} \right)_{T,P,n_1,\dots,[n_k],\dots,n_C}$$

$$\overline{V}_k = \left(\frac{\partial \overline{V}}{\partial n_k} \right)_{T,P,n_j[k]},$$

to mean partial derivatives that hold the number of moles of each species constant *except* the one being changed. Similarly, a property determined with all mole fractions held constant might be written

$$C_P = \left(\frac{\partial H}{\partial T} \right)_{P,\vec{x}},$$

or perhaps

$$C_P = \left(\frac{\partial H}{\partial T} \right)_{P,x_1,\dots,x_C}.$$

There are several macros that standardize such constructs. The `\allNs` macro expands to something meaning the number of moles of all species; by default, this is \vec{n} (package option `moles-index`), but can be changed to n_1, \dots, n_C using the package option `moles-range`. Similarly, the macros `\allmus` and `\allmusbut` do the same but with n replaced by μ , and `\allMsbut` is the same with m instead of n . There are analogous macros for mole fractions, namely `\allXs` and `\allXsbut` for x and `\allYs` and `\allYsbut` for y , as well as `\allWs` and `\allWsbut` for mass fractions—these implicitly assume that all mole or mass fractions *except* the last are used as variables. These macros all take an optional argument; for example,

```
\[ \Partial{\Ht}\{Nt_1\}{T,P,\allNsbut{1}} =
   \Partial{\Ht}\{Nt_1\}{T,P,\allNsbut[m]{1}} = \Hpm_1 \]
```

typesets as

$$\left(\frac{\partial H}{\partial n_1}\right)_{T,P,n_{j \neq 1}} = \left(\frac{\partial H}{\partial n_1}\right)_{T,P,n_{m \neq 1}} = \overline{H}_1.$$

Similarly,

$$\begin{aligned} & \left[\frac{\partial H_m}{\partial x_i} \right]_{T,P,\text{allXsbut}\{i\}} \\ &= \left[\frac{\partial H_m}{\partial x_i} \right]_{T,P,\text{allXsbut}\{m\}} \\ &= H_{pm_i} - H_{pm_ncomponents} \end{aligned}$$

becomes

$$\left(\frac{\partial H}{\partial x_i}\right)_{T,P,x_{j \neq i,C}} = \left(\frac{\partial H}{\partial x_i}\right)_{T,P,x_{m \neq i,C}} = \overline{H}_i - \overline{H}_C.$$

Users must supply their own redefinitions if they wish to hold something other than `\ncomponents` constant in addition to the argument for mole and mass fractions. Using the `moles-range` package option, for which `\allXsbut{k}` expands to $x_1, \dots, [x_k], \dots, x_{C-1}$ rather than $x_{j \neq k,C}$, the optional argument is ignored.

The optional argument to `\allNs` and similar commands is ignored when using the default options; it is relevant for package options that redefine `\allNs` to make N_i , for example; in this case, one can enter `\allNs[j]` to make TeX render N_j instead of N_i . This is useful if you are using i somewhere else in the equation.

`\allbut` Users can define new “all but” macros using the `\allbut` and `\allbutlastand`
`\allbutlastand` commands. For example,

```
\NewDocumentCommand{\allNsbut}{0{jj} m}{\allbut[#1]{#2}{\Nt}}
\NewDocumentCommand{\allXsbut}{0{jj} m}{\allbutlastand[#1]{#2}{x}}
```

are the definitions of `\allNsbut` and `\allXsbut`, respectively.

2.8 Jacobians

`\Jacobian` The Jacobian determinant is often denoted with Leibnitz-like notation, viz.,
`\Jacobiandet` $\left[\frac{\partial(K,L)}{\partial(X,Y)} \right] = \text{\Jacobiandet}\{K,L\}\{X,Y\}$,

which produces (assuming the `amsmath` package has been loaded)

$$\frac{\partial(K,L)}{\partial(X,Y)} = \begin{vmatrix} \left(\frac{\partial K}{\partial X}\right)_Y & \left(\frac{\partial K}{\partial Y}\right)_X \\ \left(\frac{\partial L}{\partial X}\right)_Y & \left(\frac{\partial L}{\partial Y}\right)_X \end{vmatrix}.$$

There are two optional arguments to `\Jacobiandet`. The first will be pre-pended before every element of the matrix (typically `\textstyle` or `\displaystyle`); the second is the extra spacing added between rows (default is 1.25 ex for text-style fractions and 2.75 ex for display-style fractions). More than two variables can be specified, viz.,

```
\left[ \frac{\partial(K,L,M)}{\partial(X,Y,Z)} \right]
= \Jacobiandet[\displaystyle][3ex]\{K,L,M\}\{X,Y,Z\}
```

will produce

$$\frac{\partial(K, L, M)}{\partial(X, Y, Z)} = \begin{vmatrix} \left(\frac{\partial K}{\partial X}\right)_{Y,Z} & \left(\frac{\partial K}{\partial Y}\right)_{X,Z} & \left(\frac{\partial K}{\partial Z}\right)_{X,Y} \\ \left(\frac{\partial L}{\partial X}\right)_{Y,Z} & \left(\frac{\partial L}{\partial Y}\right)_{X,Z} & \left(\frac{\partial L}{\partial Z}\right)_{X,Y} \\ \left(\frac{\partial M}{\partial X}\right)_{Y,Z} & \left(\frac{\partial M}{\partial Y}\right)_{X,Z} & \left(\frac{\partial M}{\partial Z}\right)_{X,Y} \end{vmatrix}.$$

The `\Jacobiandet` macro will understand implied multicomponent Jacobians, too, namely,

```
\[ \Jacobian{f_1,\dots,f_m}{x_1,\dots,x_m} =
   \Jacobiandet{f_1,\dots,f_m}{x_1,\dots,x_m} \]
```

typesets as

$$\frac{\partial(f_1, \dots, f_m)}{\partial(x_1, \dots, x_m)} = \begin{vmatrix} \left(\frac{\partial f_1}{\partial x_1}\right)_{x_{j \neq 1}} & \dots & \left(\frac{\partial f_1}{\partial x_m}\right)_{x_{j \neq m}} \\ \vdots & & \vdots \\ \left(\frac{\partial f_m}{\partial x_1}\right)_{x_{j \neq 1}} & \dots & \left(\frac{\partial f_m}{\partial x_m}\right)_{x_{j \neq m}} \end{vmatrix}.$$

If the option `moles-range` or the `thermomolesrange` environment is used, the same code produces

$$\frac{\partial(f_1, \dots, f_m)}{\partial(x_1, \dots, x_m)} = \begin{vmatrix} \left(\frac{\partial f_1}{\partial x_1}\right)_{x_2, \dots, x_m} & \dots & \left(\frac{\partial f_1}{\partial x_m}\right)_{x_1, \dots, x_{m-1}} \\ \vdots & & \vdots \\ \left(\frac{\partial f_m}{\partial x_1}\right)_{x_2, \dots, x_m} & \dots & \left(\frac{\partial f_m}{\partial x_m}\right)_{x_1, \dots, x_{m-1}} \end{vmatrix}.$$

2.9 Sums and Products

`\sumall` It is common to require sums and products such as
`\sumallbutlast`
`\prodall`

$$\sum_{i=1}^C x_i = 1 \quad \text{or} \quad x_C = 1 - \sum_{i=1}^{C-1} x_i \quad \text{or} \quad \underline{G} = \sum_{j=1}^C \mu_j n_j \quad \text{and} \quad K = \prod_{k=1}^C a_k^{v_k}.$$

This package defines shortcuts to typeset such terms thus:

```
\[ \sumall_i x_i = 1 \quad \text{\text{or}} \quad \quad
   x_{\ncomponents} = 1 - \sumallbutlast_i x_i \quad \text{\text{or}} \quad \quad
   \underline{G} = \sumall_j \mu_j Nt_j \quad \text{\text{and}} \quad \quad
   K = \prodall_k a_k^{\nu_k} \quad \]
```

The symbol C can be changed by redefining the expandable macro `\ncomponents`. This is done automatically by some package options (e.g., `TesterModel1` changes it to n ; `Sandler` changes it to C ; `Thompson` changes it to c).

Table 3. Options controlling which symbols to use by default. The macros `\Et`, `\Ut`, `\Ft`, `\Gt`, `\Ht`, `\At`, and `\Nt` represent the total energy, internal energy, Helmholtz free energy, Gibbs free energy, enthalpy, surface area, and number of moles, respectively. Symbols are shown as they would appear with the (default) option `intensive-plain`.

Option	<code>\Et</code>	<code>\Ut</code>	<code>\Ft</code>	<code>\Gt</code>	<code>\Ht</code>	<code>\At</code>	<code>\Nt</code>
EUAGHan	\underline{E}	\underline{U}	\underline{A}	\underline{G}	\underline{H}	\underline{a}	n
EUAGHaN	\underline{E}	\underline{U}	\underline{A}	\underline{G}	\underline{H}	\underline{a}	N
EUHAGan	(synonym for EUAGHan)						
EUHAGaN	(synonym for EUAGHaN)						
EUFGHan	\underline{E}	\underline{U}	\underline{F}	\underline{G}	\underline{H}	\underline{A}	n
EUFGHaN	\underline{E}	\underline{U}	\underline{F}	\underline{G}	\underline{H}	\underline{A}	N
EEFGHan	$\underline{\mathcal{E}}$	$\underline{\mathcal{E}}$	$\underline{\mathcal{F}}$	$\underline{\mathcal{G}}$	$\underline{\mathcal{H}}$	$\underline{\mathcal{A}}$	n
EEFGHaN	$\underline{\mathcal{E}}$	$\underline{\mathcal{E}}$	$\underline{\mathcal{F}}$	$\underline{\mathcal{G}}$	$\underline{\mathcal{H}}$	$\underline{\mathcal{A}}$	N
EEFGHan	$\underline{\mathcal{E}}$	$\underline{\mathcal{E}}$	$\underline{\mathcal{F}}$	$\underline{\mathcal{G}}$	$\underline{\mathcal{H}}$	\underline{a}	n
EEFGHaN	$\underline{\mathcal{E}}$	$\underline{\mathcal{E}}$	$\underline{\mathcal{F}}$	$\underline{\mathcal{G}}$	$\underline{\mathcal{H}}$	\underline{a}	N
EEAGHan	$\underline{\mathcal{E}}$	$\underline{\mathcal{E}}$	$\underline{\mathcal{A}}$	$\underline{\mathcal{G}}$	$\underline{\mathcal{H}}$	\underline{a}	N
EEAGHaN	$\underline{\mathcal{E}}$	$\underline{\mathcal{E}}$	$\underline{\mathcal{A}}$	$\underline{\mathcal{G}}$	$\underline{\mathcal{H}}$	$\underline{\mathcal{A}}$	n
EUAGHAN	\underline{E}	\underline{U}	\underline{A}	\underline{G}	\underline{H}	$\underline{\mathcal{A}}$	N
EUFGHan	\underline{E}	\underline{U}	\underline{F}	\underline{G}	\underline{H}	\underline{a}	n
EUFGHaN	\underline{E}	\underline{U}	\underline{F}	\underline{G}	\underline{H}	\underline{a}	N

3 Loading the Package

To load the package with the defaults enabled, load it the usual way:

```
\usepackage{thermodynamics}
```

The package options loaded by default are `EUAGHan`, `subscripts`, `parentheses`, `intensive-plain`, and `moles-index`. These define, respectively, the default symbols to use for total energy, internal energy, Helmholtz free energy, and so forth; the manner of writing partial derivatives; the delimiters around partial derivatives; the manner of denoting extensive, molar, and specific properties; and the manner of writing the number of moles of all or most species in partial derivatives. The default behavior can be altered by options in the following section.

3.1 Package Options

There are three categories of options: options that affect which symbols are used, options that affect how symbols are decorated, and options that affect how partial derivatives are displayed. These are presented in turn.

3.1.1 Options that Change Symbol Sets

There are several options that choose the set of symbols to use for total energy, internal energy, Helmholtz free energy, and so forth. These are summarized in Table 3. The default is `EUAGHan`.

Using `EUAGHan` (the default), we might use the following markup:

```
\[ \Ft = \Ut - T\St = -P\Vt + \sumall_i \mu_i \Nt_i + \sigma \At
\quad \Hm = \Um + P\Vm \quad \Et = \Ut + \frac{1}{2} mv^2 \]
```

Table 4. Notation sets that can be set using the options `intensive-plain` (the default), `extensive-plain`, `extensive-superscript`, and `intensive-lowercase`.

Option	<code>\Vt</code>	<code>\Vm</code>	<code>\Vs</code>	<code>\Vpm_i</code>
<code>intensive-plain</code>	\underline{V}	V	\hat{V}	\overline{V}_i
<code>extensive-plain</code>	V	\underline{V}	\hat{V}	\overline{V}_i
<code>extensive-superscript</code>	V^t	V	\hat{V}	\overline{V}_i
<code>intensive-lowercase</code>	V	v	\hat{v}	\overline{v}_i

which would look like

$$\underline{A} = \underline{U} - T\underline{S} = -P\underline{V} + \sum_{i=1}^C \mu_i n_i + \sigma \underline{a} \quad H = U + PV \quad \underline{E} = \underline{U} + \frac{1}{2}mv^2$$

Using the `EEFGHAN` option, the same markup would yield

$$\underline{F} = \underline{E} - T\underline{S} = -P\underline{V} + \sum_{i=1}^C \mu_i N_i + \sigma \underline{A} \quad H = U + PV \quad \underline{\mathcal{E}} = \underline{E} + \frac{1}{2}mv^2.$$

3.1.2 Options for Extensive vs. Molar Properties

There are four sets of notation that define how extensive properties are represented, as shown in Table 4. The default is `intensive-plain`, which (using the volume as an example) represents the total, molar, specific, and partial molar volumes, respectively, as \underline{V} , V , \hat{V} , and \overline{V}_j .

For example, the definition of the partial molar enthalpy would be different depending on which set of notation is used. The markup

```
\[ \Hpm_i = \Partial*{\Ht}{\Nt_i}{T,P,\allNsbu{i}}
= \Partial{\Nt\Hm}{\Nt_i}{T,P,\allNsbu{i}} \]
```

yields the following, depending on the package option loaded:

$$\begin{aligned} \overline{H}_i &= \left(\frac{\partial \underline{H}}{\partial n_i} \right)_{T,P,n_{j \neq i}} = \left(\frac{\partial nH}{\partial n_i} \right)_{T,P,n_{j \neq i}} && \text{intensive-plain} \\ \overline{H}_i &= \left(\frac{\partial H}{\partial n_i} \right)_{T,P,n_{j \neq i}} = \left(\frac{\partial n\underline{H}}{\partial n_i} \right)_{T,P,n_{j \neq i}} && \text{extensive-plain} \\ \overline{H}_i &= \left(\frac{\partial H^t}{\partial n_i} \right)_{T,P,n_{j \neq i}} = \left(\frac{\partial nH}{\partial n_i} \right)_{T,P,n_{j \neq i}} && \text{extensive-superscript} \\ \overline{h}_i &= \left(\frac{\partial H}{\partial n_i} \right)_{T,P,n_{j \neq i}} = \left(\frac{\partial nh}{\partial n_i} \right)_{T,P,n_{j \neq i}} && \text{intensive-lowercase} \end{aligned}$$

The use of `intensive-lowercase` is strongly discouraged.

Note that the number of moles can be changed from n to N via the options in the previous section.

`thermoextensiveplain` (*env.*) It is possible to change notation locally, though there are very, very few good reasons why you would want to do this in a regular document—normally, one would use the corresponding package options. The environment `thermoextensivesuperscript` (*env.*) means `\St` will become \underline{S} and `\Sm` will become S in the text. Similarly, inside `thermoextensiveplain`, `\St` will become S and `\Sm` will become \underline{S} . Inside `thermoextensivesuperscript`, `\St` will become S^t and `\Sm` will be S ; and inside `thermoextensivesubscript`, `\St` will be S and `\Sm` will be s .

3.1.3 Options Affecting Partial Molar Properties

There are two options that change how partial molar properties are typeset. For option `longpm` (the default), the subscript and superscript are considered part of the symbol, producing something like \overline{H}_i^E . The option `shortpm` makes it display as \overline{H}_i^E . The `shortpm` option is the default for several textbook-related options.

`thermolongpm` (*env.*) It is possible to change partial molar notation locally—this could be useful if you have a very, very long superscript or subscript and the default option looks silly, say. The environment `thermolongpm` sets the style equivalent to the `longpm` option, while the `thermoshortpm` environment sets the local style equivalent to the `shortpm` option. For example,

```
\[ \Spm_i = \Spm_i^{\IGM} + \SRpm_i
\begin{thermoshortpm}
= \Spm_i^{\IGM} + \SRpm_i
\end{thermoshortpm} \]
```

produces

$$\overline{S}_i = \overline{S}_i^{\text{IGM}} + \overline{S}_i^R = \overline{S}_i^{\text{IGM}} + \overline{S}_i^R,$$

assuming the default options are loaded.

3.1.4 Options Affecting Partial Derivatives

There are several options that change how partial derivatives are rendered. First are the options that affect the delimiters. We will use the following code as an example:

```
\[ \Partial*\{\Vm\}{T}{P} = \PartialMixSecond{\Gm}{T}{P}{}
= \PartialMixSecond{\Gm}{P}{T}{}
= -\Partial{\Sm}{P}{T}. \]
```

Using the `parentheses` option (the default), this gives

$$\left(\frac{\partial V}{\partial T}\right)_P = \left(\frac{\partial^2 G}{\partial T \partial P}\right) = \left(\frac{\partial^2 G}{\partial P \partial T}\right) = -\left(\frac{\partial S}{\partial P}\right)_T.$$

The option `brackets` changes the output to

$$\left[\frac{\partial V}{\partial T}\right]_P = \left[\frac{\partial^2 G}{\partial T \partial P}\right] = \left[\frac{\partial^2 G}{\partial P \partial T}\right] = -\left[\frac{\partial S}{\partial P}\right]_T.$$

The option `bar` changes the output to

$$\frac{\partial V}{\partial T}\Big|_P = \frac{\partial^2 G}{\partial T \partial P} = \frac{\partial^2 G}{\partial P \partial T} = -\frac{\partial S}{\partial P}\Big|_T.$$

The option `plain-derivatives` eliminates all delimiters; this forces the `nosubscripts` option. The output in this case is

$$\frac{\partial V(T,P)}{\partial T} = \frac{\partial^2 G(T,P)}{\partial T \partial P} = \frac{\partial^2 G(T,P)}{\partial P \partial T} = - \frac{\partial S(T,P)}{\partial P}.$$

Accompanying the `plain-derivatives` option is the `nosubscripts` option, which overrides the default option `subscripts`. This option makes partial derivatives such as

$$\left(\frac{\partial V}{\partial P} \right)_T \quad (\text{subscripts option}),$$

and instead renders them

$$\left(\frac{\partial V(T,P)}{\partial P} \right) \quad (\text{nosubscripts option}).$$

Combined with `plain-derivatives`, this would give

$$\frac{\partial V(T,P)}{\partial P} \quad (\text{nosubscripts and plain-derivatives options}).$$

The variables are sorted into an order defined by an internal constant, meaning T will always be listed before P . The order by default is in the order that terms appear in the fundamental equations, that is,

$$\begin{aligned} d\underline{U} &= T d\underline{S} - P d\underline{V} + \mu d\underline{n} \\ d\underline{H} &= T d\underline{S} + \underline{V} dP + \mu d\underline{n} \\ d\underline{A} &= -\underline{S} dT - P d\underline{V} + \mu d\underline{n} \\ &\vdots \\ d\underline{\Omega} &= -\underline{S} dT - P d\underline{V} - n d\mu, \end{aligned}$$

with the exception that subscripted variables are (currently) not sortable. If `\Nt_i` or some similar construct appears as a variable and `\allNsbut{i}` appears in the held-constant list, the package will assume that the argument list *should* contain all of the mole numbers. Symbols without subscripts that are not in the fundamental equation or one of its variants are sorted in alphabetical order.

`thermoparentheses (env.)` If you want to use parentheses *locally*, even though your overall document uses
`thermobrackets (env.)` another delimiter, the `thermoparentheses` environment will do that. Similarly,
`thermobraces (env.)` `thermobrackets` will temporarily switch to brackets, `thermobar` will temporarily
`thermobar (env.)` switch to a tailing vertical bar, and `thermoplain` will remove delimiters altogether.
`thermoplain (env.)` The environments `thermosubscripts` and `thermoN0subscripts` force the use or
`thermosubscripts (env.)` disuse of subscripts, respectively.
`thermoN0subscripts (env.)`

3.1.5 Options Regarding the Number of Moles

`\allNs` The default option `moles-index` defines the macro `\allNs` to expand to \vec{n} and
`\allNsbut` the macro `\allNsbut{i}` to expand to $n_{j \neq i}$. You can change the dummy index:
`\allNsbut[k]{i}` expands to $n_{k \neq i}$ by default. This is typically not necessary, however:
if you type `\allNsbut{j}`, the package will figure out that you want $n_{k \neq j}$ rather than
 $n_{j \neq j}$. The time to use the optional argument is in situations such as

$$\left(\frac{\partial \mu_j}{\partial n_k} \right)_{n_{i \neq k}},$$

Table 5. Illustration of the moles-index and moles-range options and their effects on `\allNs` and `\allNsbut`.

Macro ^a	moles-index	moles-range
<code>\allNs</code>	\vec{n}	n_1, \dots, n_C
<code>\allNsbut{1}</code>	$n_{j \neq 1}$	n_2, \dots, n_C
<code>\allNsbut{i}</code>	$n_{j \neq i}$	$n_1, \dots, [n_i], \dots, n_C$
<code>\allNsbut{j}</code>	$n_{k \neq j}$	$n_1, \dots, [n_j], \dots, n_C$
<code>\allNsbut{\ncomponents}</code>	$n_{j \neq C}$	n_1, \dots, n_{C-1}
<code>\allNsbut[k]{i}</code>	$n_{k \neq i}$	$n_1, \dots, [n_i], \dots, n_C$
<code>\allXs</code>	\vec{x}	x_1, \dots, x_C
<code>\allXsbut{1}</code>	$x_{j \neq 1, C}$	x_2, \dots, x_{C-1}
<code>\allXsbut{i}</code>	$x_{j \neq i, C}$	$x_1, \dots, [x_i], \dots, x_{C-1}$
<code>\allXsbut{j}</code>	$x_{k \neq j, C}$	$x_1, \dots, [x_j], \dots, x_{C-1}$
<code>\allXsbut{\ncomponents-1}</code>	$x_{j \neq C-1, C}$	x_1, \dots, x_{C-2}
<code>\allXsbut[k]{\ncomponents-1}</code>	$x_{k \neq C-1, C}$	x_1, \dots, x_{C-2}
<code>\allXsbut{\ncomponents}</code> ^b	$x_{j \neq C}$	x_1, \dots, x_{C-1}

^aYou may use C directly instead of `\ncomponents` here, but then it will not change to another symbol if you want to switch to an option that redefines `\ncomponents` later.

^bThis would typically be used to denote something like $G(T, P, n, x_1, \dots, x_{C-1})$ rather than in a subscript, but it looks silly if we do not handle this case this way.

which is incorrect if the dummy index j is used in place of the i .

You can change these to expand to ranges using the `moles-range` option, which renders `\allNs` as n_1, \dots, n_C and `\allNsbut{i}` as $n_1, \dots, [n_i], \dots, n_C$. The optional argument is ignored in this set of notation. Examples of these options are shown in Table 5.

`thermomolesrange (env.)` The environment `thermomolesrange` temporarily redefines `\allNs` and other range-oriented macros as though the `moles-range` package option had been invoked.

`\ncomponents` You can change the symbol for the number of components (default: C) by redefining the macro `\ncomponents`. This is done for you by some package options that define notation for particular textbooks.

3.1.6 Other Options

`\dbar` The default for path-dependent one-forms (often called “inexact differentials”) is `\dbar`, which looks like \vec{d} . This can be changed, if desired, to a delta (δ) with the `delta` option to the package.

It should be noted that the `\dbar` macro is context-dependent: changing the typeface to something not supported will probably ruin it, as the kerning is very font-specific. This package currently supports Computer Modern, Times, Palatino, Bitstream Charter, Garamond, and Utopia, but other typefaces may require a manual redefinition.

3.1.7 Options for Particular Textbooks

There are several options that load package options and/or redefine particular commands to match the notation in a particular textbook. A table showing how various properties are typeset using each of these packages is included in Appendix A. So far, the following textbooks are supported:

- Bejan** Notation used by Bejan, *Advanced Engineering Thermodynamics*, Third Edition. Wiley: Hoboken, 2006.
- CBK** Notation used by Çengel, Boles, and Kanoğlu, *Thermodynamics: An Engineering Approach*, Ninth Edition. McGraw Hill: Singapore, 2020. Their symbols for specific and total volume, which appear to be from the font ITC Benguiat Gothic (and reasonably approximated by Krub Italic), are not directly supported.³
- ElliottLira** Notation used by Elliott and Lira, *Introductory Chemical Engineering Thermodynamics*, Second Edition. Prentice Hall: Upper Saddle River, 2012. Note: these authors use Henry's Law (molal basis) in the form $y_i P = m_i / K_{H,i}$, which is inconsistent with their notation for the rational basis, so that macro is left set to its default.
- KlotzRosenberg** Notation used by Klotz and Rosenberg, *Chemical Thermodynamics: Basic Concepts and Methods*, Seventh Edition. Wiley: Hoboken, 2008.
- Koretsky** Notation used by Koretsky, *Engineering and Chemical Thermodynamics*, Second Edition. Wiley: New Caledonia, 2013.
- ModellReid** Alias for TesterModell (the first and second editions were by Modell and Reid; Reid died prior to the third edition).
- MSBB** Notation used by Moran, Shapiro, Boettner, and Bailey, *Fundamentals of Engineering Thermodynamics*, Eighth Edition. Wiley: Kendallville, 2014.
- Prausnitz** Notation used by Prausnitz, Lichtenthaler, and de Azevedo, *Molecular Thermodynamics of Fluid-Phase Equilibria*, Third Edition, Pearson, 1998.
- Sandler** Notation used by Sandler, *Chemical, Biochemical, and Engineering Thermodynamics*, Fifth Edition. Wiley: Hoboken, 2017.
- SVNAS** Notation used by Smith, Van Ness, Abbott, and Swihart, *Introduction to Chemical Engineering Thermodynamics*, Ninth Edition. McGraw-Hill: Boston, 2021.
- TesterModell** Notation used by Tester and Modell, *Thermodynamics and Its Applications*, Third Edition, Prentice Hall: Upper Saddle River, 1997.
- Thompson** Notation used by Thompson, *A Unified Introduction to Chemical Engineering Thermodynamics*, Stillwater Press: Orono, 2000. Note that he uses c , c , n_C , and n for the number of components in various places in the book; I chose c for the definition of `\ncomponents`, but it is impossible to be completely consistent with his notation.

There may well be some inconsistencies between the notation in these books and the symbols used here. I will fix such inconsistencies as I become aware of them.

³ If you have ITC Benguiat Gothic installed and want to use it for the volume symbol, you can use it through Xe_{La}TeX; try this:

```
\usepackage{fontspec}
\setsansfont{ITC Benguiat Gothic}
\ExplSyntaxOn
\tl_gset:Nn \g__thermodynamics_volume_symbol {{\text{\sffamily V}}}
\ExplSyntaxOff
```

Important: Some textbooks use symbols other than μ for chemical potential and γ for activity coefficients. For example, KlotzRosenberg uses μ for the chemical potential and γ for the activity coefficient. In such cases, the macro `\mu` will produce whatever symbol is used for the chemical potential, and `\gamma` will produce the activity coefficient. The original symbols will still be available as `\gammaait` and `\muit`, respectively.

A Notation Across Textbooks

Some of the textbooks supported by this package use multiple symbols for the same quantity—for example, Sandler’s textbook uses both P^{sat} and P^{vap} for vapor (saturation) pressure, and Elliott and Lira use both ΔH° and ΔH_R° for the standard enthalpy of reaction. This package may only support one of the two in such cases.

Table 6 gives a description of each symbol supported by the package, the corresponding command, and the way that command is rendered using each of the textbook-related options in Section 3.1.7. Packages consisting of two names are shortened to one name in the headers for space considerations.

Table 6. How symbols are typeset for various textbooks. Entries in gray indicate quantities that are not directly or indirectly defined in a particular textbook and are therefore inherited from the defaults. *Note: Options with two authors, such as ElliottLira, only include one name so as to fit on one page.*

Quantity	Macro	Default	Bejan	CBK	Elliott	Klotz	Koretsky	MSBB	Prausnitz	Sandler	SVNAS	Modell	Thompson
# components	\ncomponents	C	n	C	C	C	m	C	m	C	C	n	c
activity (absolute)	\actabs_i	λ_i	λ_i	λ_i	λ_i	λ_i	λ_i	λ_i	λ_i	λ_i	λ_i	λ_i	λ_i
activity (relative)	\actrel_i	a_i	a_i	a_i	a_i	a_i	a_i	a_i	a_i^a	a_i	a_i	a_i	$\widehat{a_i}$
activity coefficient (ordinary)	\gamma_i	γ_i	γ_i	γ_i	γ_i	γ_i	γ_i	γ_i	γ_i	γ_i	γ_i	γ_i	γ_i
activity coefficient (Henry, rational)	\gammarat_i	γ_i^*	γ_i^*	γ_i^*	γ_i^*	γ_i	$\gamma_i^{\text{Henry's}}$	γ_i^*	γ_i^*	γ_i^*	γ_i^*	γ_i^{**}	γ_i^*
activity coefficient (Henry, molal)	\gammamol_i	γ_i^\square	γ_i^\square	γ_i^\square	γ_i^\square	γ_i	γ_i^m	γ_i^\square	γ_i^\square	γ_i^\square	γ_i^\square	γ_i^*	γ_i^\square
adiabatic compressibility	See isentropic compressibility												
adiabatic expansivity	See isentropic expansivity												
area (total)	\At	\underline{a}	A	A	\underline{a}	a	A	A	A	a	a^t	\underline{a}	\underline{A}
chemical potential	\mu	μ	μ	μ	μ	μ	μ	μ	μ	μ	μ	μ	μ
electric potential	\Epot	E	E	E	E	\mathcal{E}^b	E	E	\mathcal{E}^b	E	E	\exists	E
electric potential (standard)	\Epot*\std	E°	E°	E°	E°	\mathcal{E}°	E°	E°	\mathcal{E}^0	E°	E°	\exists°	E°
energy (total)	\Et	\underline{E}	E	E	\underline{E}	E	E	E	E	E	E^t	\underline{E}	\underline{E}
energy (molar)	\Em	E	\bar{e}	\bar{e}	E	E_m	e	\bar{e}	e	\underline{E}	E	E	E
energy (specific)	\Es	\hat{E}	e	e	E	\hat{E}	\hat{e}	e	\hat{e}	\hat{E}	E	\hat{E}	E
enthalpy (total)	\Ht	\underline{H}	H	H	\underline{H}	H	H	H	H	H	H^t	\underline{H}	\underline{H}
enthalpy (molar)	\Hm	H	\bar{h}	\bar{h}	H	H_m	h	\bar{h}	h	\underline{H}	H	H	H
enthalpy (specific)	\Hs	\hat{H}	h	h	H	\hat{H}	\hat{h}	h	\hat{h}	\hat{H}	H	\hat{H}	\hat{H}
enthalpy (partial molar) of i	\Hpm_i	\bar{H}_i	\bar{h}_i	\bar{h}_i	\bar{H}_i	H_{mi}	\bar{H}_i	\bar{H}_i	\bar{h}_i	\bar{H}_i	\bar{H}_i	\bar{H}_i	\bar{H}_i
enthalpy of reaction	\Deltarxn\Hm	ΔH_{rxn}	$\Delta \bar{h}_{\text{rxn}}$	\bar{h}_R	ΔH	ΔH_m	Δh_{rxn}	$\Delta \bar{h}_{\text{rxn}}$	Δh_{rxn}	$\Delta_{\text{rxn}} \underline{H}$	ΔH	ΔH_{rx}	ΔH_R
enthalpy of fusion	\Delta tafus\Hm	ΔH^{fus}	\bar{h}_{sf}	\bar{h}_{sf}	ΔH^{fus}	ΔH_m	Δh_{fus}	\bar{h}^{fus}	Δh^{fus}	$\Delta_{\text{fus}} \underline{H}$	ΔH^{sl}	ΔH^{fus}	ΔH^{SL}
enthalpy of vaporization	\Delta tavap\Hm	ΔH^{vap}	\bar{h}_{sf}	\bar{h}_{fg}	ΔH^{vap}	ΔH_m	Δh_{vap}	\bar{h}^{vap}	Δh^{vap}	$\Delta_{\text{vap}} \underline{H}$	ΔH^{vl}	ΔH^{vap}	ΔH^{LV}

continued on next page. . .

^aThe symbol \underline{a} requires the dutchcal fonts to be installed, though the package itself is not loaded. This symbol is an approximation to the one the book actually uses.

^bThe symbol \mathcal{E} requires the emf package. If that package is not loaded, \mathcal{E} is used instead.

Quantity	Macro	Default	Bejan	CBK	Elliott	Klotz	Koretsky	MSBB	Prausnitz	Sandler	SVNAS	Modell	Thompson
enthalpy of sublimation	$\backslash\text{Deltasub}\backslash\text{Hm}$	ΔH^{sub}	\bar{h}_{sg}	\bar{h}_{sg}	ΔH^{sub}	ΔH_{m}	Δh_{sub}	\bar{h}^{sub}	Δh^{sub}	$\Delta_{\text{sub}}H$	ΔH^{ps}	ΔH^{sub}	ΔH^{SV}
entropy (total)	$\backslash\text{St}$	\underline{S}	S	S	\underline{S}	S	S	S	S	S	S^t	\underline{S}	\underline{S}
entropy (molar)	$\backslash\text{Sm}$	S	\bar{s}	\bar{s}	S	S_{m}	s	\bar{s}	s	\underline{S}	S	S	S
entropy (specific)	$\backslash\text{Ss}$	\hat{S}	s	s	S	\hat{S}	\hat{s}	s	\hat{s}	\hat{S}	S	\hat{S}	\hat{S}
entropy (partial molar) of i	$\backslash\text{Spm}_i$	\bar{S}_i	\bar{S}_i	\bar{S}_i	\bar{S}_i	S_{mi}	\bar{S}_i	\bar{S}_i	\bar{S}_i	\bar{S}_i	\bar{S}_i	\bar{S}_i	\bar{S}_i
extent of reaction	$\backslash\text{rxn}$	ξ	ζ	ξ	ξ	ξ	ξ	ε	ξ	X	ε	ξ	ξ
fugacity of pure substance	$\backslash\text{fpure}$	f	f	f	f	f	f	f	f_{pure}	f	f	f	f
fugacity of pure i	$\backslash\text{fpure}_i$	f_i	f_i^{\bullet}	f_i	f_i	f_i^{\bullet}	f_i	f_i	$f_{\text{pure } i}$	f_i	f_i	f_i	f_i^{\bullet}
fugacity of i in mixture	$\backslash\text{fmix}_i$	\hat{f}_i	f_i	\hat{f}_i	\hat{f}_i	f_i	\hat{f}_i	\bar{f}_i	f_i	\bar{f}_i	\hat{f}_i	\hat{f}_i	\bar{f}_i
fugacity at saturation	$\backslash\text{fsat}$	f^{sat}	f^{sat}	f^{sat}	f^{sat}	f^{sat}	f^{sat}	f^{sat}	$f_{\text{pure}}^{\text{sat}}$	f^{sat}	f^{sat}	f^{sat}	f^{sat}
fugacity coefficient of pure substance	$\backslash\text{hipure}$	ϕ	ϕ	ϕ	ϕ	γ	φ	ϕ	φ_{pure}	ϕ	ϕ	ϕ	ϕ
fugacity coefficient of pure i	$\backslash\text{hipure}_i$	ϕ_i	ϕ_i^{\bullet}	ϕ_i	ϕ_i	γ_i^{\bullet}	φ_i	ϕ_i	$\varphi_{\text{pure } i}$	ϕ_i	ϕ_i	ϕ_i	ϕ_i^{\bullet}
fugacity coefficient of i in mixture	$\backslash\text{phimix}_i$	$\hat{\phi}_i$	ϕ_i	$\hat{\phi}_i$	$\hat{\phi}_i$	γ_i	$\hat{\phi}_i$	$\hat{\phi}_i$	φ_i	$\hat{\phi}_i$	$\hat{\phi}_i$	$\hat{\phi}_i$	$\hat{\phi}_i$
fugacity coefficient at saturation	$\backslash\text{phisat}$	ϕ^{sat}	ϕ^{sat}	ϕ^{sat}	ϕ^{sat}	γ^{sat}	φ^{sat}	ϕ^{sat}	φ^{s}	ϕ^{sat}	ϕ^{sat}	ϕ^{sat}	ϕ^{sat}
Gibbs free energy (total)	$\backslash\text{Gt}$	\underline{G}	G	G	\underline{G}	G	G	G	G	G	G^t	\underline{G}	\underline{G}
Gibbs free energy (molar)	$\backslash\text{Gm}$	G	\bar{g}	\bar{g}	G	G_{m}	g	\bar{g}	g	\underline{G}	G	G	G
Gibbs free energy (specific)	$\backslash\text{Gs}$	\hat{G}	g	g	G	\hat{G}	\hat{g}	g	\hat{g}	\hat{G}	G	\hat{G}	\hat{G}
Gibbs free energy (partial molar)	$\backslash\text{Gpm}_i$	\bar{G}_i	\bar{G}_i	\bar{G}_i	\bar{G}_i	G_{mi}	\bar{G}_i	\bar{G}_i	\bar{G}_i	\bar{G}_i	\bar{G}_i	\bar{G}_i	\bar{G}_i
Gibbs free energy of mixing	$\backslash\text{Deltamix}\backslash\text{Gt}$	ΔG_{mix}	ΔG_{mix}	ΔG_{mix}	ΔG_{mix}	ΔG_{mix}	ΔG_{mix}	ΔG_{mix}	ΔG_{mixing}	$\Delta_{\text{mix}}G$	ΔG^t	ΔG_{mix}	$\Delta_{\text{MIX}}G$
Gibbs free energy of reaction	$\backslash\text{Deltarxn}\backslash\text{Gm}$	ΔG_{rxn}	$\Delta \bar{g}_{\text{rxn}}$	\bar{g}_R	ΔG	ΔG_{m}	Δg_{rxn}	$\Delta \bar{g}_{\text{rxn}}$	Δg_{rxn}	$\Delta_{\text{rxn}}G$	ΔG	ΔG_{rx}	ΔG_R
grand potential (total)	$\backslash\text{Lt}$	$\underline{\Omega}$	F_{μ}	Ω	$\underline{\Omega}$	Ω	Ω	Ω	Ω	Ω	$\underline{\Omega}$	$\underline{\Omega}$	$\underline{\Omega}$
grand potential (molar)	$\backslash\text{Lm}$	Ω	\bar{f}_{μ}	$\bar{\omega}$	Ω	Ω_{m}	ω	$\bar{\omega}$	ω	$\underline{\Omega}$	Ω	Ω	Ω
grand potential (specific)	$\backslash\text{Ls}$	$\hat{\Omega}$	f_{μ}	ω	$\hat{\Omega}$	$\hat{\Omega}$	$\hat{\omega}$	ω	$\hat{\omega}$	$\hat{\Omega}$	Ω	$\hat{\Omega}$	$\hat{\Omega}$
heat (total)	$\backslash\text{Qt}$	\underline{Q}	Q	Q	\underline{Q}	Q	Q	Q	Q	Q	Q^t	\underline{Q}	\underline{Q}
heat (molar)	$\backslash\text{Qm}$	Q	\bar{q}	\bar{q}	Q	Q_{m}	q	\bar{q}	q	\underline{Q}	Q	Q	Q
heat (specific)	$\backslash\text{Qs}$	\hat{Q}	q	q	Q	\hat{Q}	\hat{q}	q	\hat{q}	\hat{Q}	Q	\hat{Q}	\hat{Q}
heat capacity (isobaric, total)	$\backslash\text{cPt}$	\underline{C}_P	C_P	C_P	\underline{C}_P	C_P	C_P	C_P	C_P	NC_P	nC_P	\underline{C}_P	\underline{C}_P

continued on next page...

Quantity	Macro	Default	Bejan	CBK	Elliott	Klotz	Koretsky	MSBB	Prausnitz	Sandler	SVNAS	Modell	Thompson
heat capacity (isobaric, molar)	\cP	C_P	\bar{c}_P	\bar{c}_P	C_P	C_{Pm}	c_P	\bar{c}_P	c_P	C_P	C_P	C_p	C_P
heat capacity of i (isobaric, molar)	\cP_i	$C_{P,i}$	$\bar{c}_{P,i}$	$\bar{c}_{P,i}$	$C_{P,i}$	C_{Pmi}	$C_{P,i}$	$\bar{c}_{P,i}$	$c_{P,i}$	$C_{P,i}$	$C_{P,i}$	C_{pi}	$C_{P,i}$
heat capacity (isobaric, specific)	\cPs	\hat{C}_P	c_P	c_P	C_P	\hat{C}_P	\hat{c}_P	c_P	\hat{c}_P	\hat{C}_P	C_P	\hat{C}_p	\hat{C}_p
heat capacity (isobaric, partial molar)	\cPpm_i	$\overline{C}_{P,i}$	$\bar{c}_{P,i}$	$\bar{c}_{P,i}$	$\overline{C}_{P,i}$	C_{Pmi}	$\overline{C}_{P,i}$	$\bar{C}_{P,i}$	$\bar{c}_{P,i}$	$\overline{C}_{P,i}$	$\overline{C}_{P,i}$	\overline{C}_{pi}	$\overline{C}_{P,i}$
heat capacity (isochoric, total)	\cVt	\underline{C}_V	C_V	C_V^3	\underline{C}_V	C_V	C_V	C_v	C_v	NC_V	nC_V	\underline{C}_v	\underline{C}_V
heat capacity (isochoric, molar)	\cV	C_V	\bar{c}_v	\bar{c}_v^3	C_V	C_{Vm}	c_V	\bar{c}_v	c_v	C_V	C_V	C_v	C_V
heat capacity (isochoric, specific)	\cVs	\hat{C}_V	c_v	c_v^3	C_V	\hat{C}_V	\hat{c}_v	c_v	\hat{c}_p	\hat{C}_V	C_V	\hat{C}_v	\hat{C}_V
heat capacity (isochoric, partial molar)	\cVpm_i	$\overline{C}_{V,i}$	$\bar{c}_{v,i}$	$\bar{c}_{v,i}^3$	$\overline{C}_{V,i}$	C_{Vmi}	\overline{C}_V	$\bar{C}_{v,i}$	$\bar{c}_{v,i}$	$\overline{C}_{V,i}$	$\overline{C}_{V,i}$	\overline{C}_{vi}	$\overline{C}_{V,i}$
Helmholtz free energy (total)	\Ft	\underline{A}	F	A	\underline{A}	A	A	Ψ	A	A	A^t	\underline{A}	\underline{A}
Helmholtz free energy (molar)	\Fm	A	\bar{f}	\bar{a}	A	A_m	a	$\bar{\psi}$	a	\underline{A}	A	A	A
Helmholtz free energy (specific)	\Fs	\hat{A}	f	a	A	\hat{A}	\hat{a}	ψ	\hat{a}	\hat{A}	A	\hat{A}	\hat{A}
Helmholtz free energy (partial molar)	\Fpm_i	\overline{A}_i	\bar{f}_i	\bar{a}_i	\overline{A}_i	A_{mi}	\overline{A}_i	$\overline{\Psi}_i$	\bar{a}_i	\overline{A}_i	\overline{A}_i	\overline{A}_i	\overline{A}_i
Henry's constant (rational basis)	\Henryrat_i	h_i	h_i	h_i	h_i	k_i	\mathcal{H}_i	h_i	H_i	H_i	\mathcal{H}_i	f^{**}	$k_{H,i}$
Henry's constant (molal basis)	\Henrymol_i	\mathcal{H}_i	\mathcal{H}_i	\mathcal{H}_i	\mathcal{H}_i	k_i''	\mathcal{H}_i	\mathcal{H}_i	\mathcal{H}_i	\mathcal{H}_i	\mathcal{H}_i	f^*	$k_{H,i}$
ideal gas enthalpy	\Hm^\IG	H^{IG}	\bar{h}^{IG}	\bar{h}^{IG}	H^{ig}	H_m^{IG}	$h^{\text{ideal gas}}$	\bar{h}^*	h^{id}	H^{IG}	H^{ig}	H^0	H^{IG}
ideal gas mixture enthalpy	\Hm^\IGM	H^{IGM}	\bar{h}^{IG}	\bar{h}^{IGM}	H^{ig}	H_m^{IGM}	h^{ideal}	\bar{h}^*	h^{id}	H^{IGM}	H^{ig}	H^0	H^{IG}
ideal solution enthalpy	\Hm^\IS	H^{IS}	\bar{h}^{IS}	\bar{h}^{IS}	H^{is}	H_m^{IS}	h^{ideal}	\bar{h}^{IS}	$h^{(\text{ideal})}$	H^{IM}	H^{id}	H^{ID}	H^{IS}
inexact differential	\dbar	d	δ	δ	d	d	δ	δ	d	d	d	δ	δ
internal energy (total)	\Ut	\underline{U}	U	U	\underline{U}	U	U	U	U	U	U^t	\underline{U}	\underline{U}
internal energy (molar)	\Um	\underline{U}	\bar{u}	\bar{u}	U	U_m	u	\bar{u}	u	\underline{U}	U	U	U
internal energy (specific)	\Us	\hat{U}	u	u	U	\hat{U}	\hat{u}	u	\hat{u}	\hat{U}	U	u	\hat{U}
internal energy (partial molar)	\Upm_i	\overline{U}_i	\bar{u}_i	\bar{u}_i	\overline{U}_i	U_{mi}	\overline{U}_i	\overline{U}_i	\bar{u}_i	\overline{U}_i	\overline{U}_i	\overline{U}_i	\overline{U}_i
isothermal compressibility	\kappaT	κ_T	κ	α	κ_T	β	κ	κ	β	κ_T	κ	κ_T	κ_T
isentropic compressibility	\kappapS	κ_S	κ_S	α_S	κ_S	β_S	κ_S	α	β_S	κ_S	κ_S	κ_S	κ_S
isobaric expansivity	\alphaP	α_P	β	β	α_P	α	β	β	α	α	β	α_P	α_P
isentropic expansivity	\alphapS	α_S	β_S	β_S	α_S	α_S	β_S	β_S	α_S	α_S	β_S	α_S	α_S
Joule-Thomson coefficient	\muJT	μ_{JT}	μ_J	μ_{JT}	μ_{JT}	$\mu_{J.T.}$	μ_{JT}	μ_J	μ_{JT}	μ	μ	α_H	α_H

continued on next page...

Quantity	Macro	Default	Bejan	CBK	Elliott	Klotz	Koretsky	MSBB	Prausnitz	Sandler	SVNAS	Modell	Thompson
moles	\Nt	n	N	N	n	n	n	n	n	N	n	N	n
saturation pressure	\Psat	p^{sat}	p^{sat}	p^v	p^{sat}	p	p^{sat}	p^{sat}	p^s	p^{sat}	p^{sat}	p^{sat}	p^{sat}
saturation pressure of i	\Psat_i	p_i^{sat}	p_i^{sat}	p_i^v	p_i^{sat}	p_i^\bullet	p_i^{sat}	p_i^{sat}	p_i^s	p_i^{sat}	p_i^{sat}	p_i^{sat}	p_i^{sat}
standard molality	\Cstd_i	C_i°	C_i°	C_i°	m_i°	m_i°	C_i°	C_i°	C_i^0	1 molal	m°	m^+	m_i°
standard chemical potential	\mustd	μ°	μ°	μ°	μ°	μ°	μ^0	\bar{g}°	μ^0	μ°	G°	μ^0	Γ
standard enthalpy	\Hm^\std	H°	\bar{h}°	\bar{h}°	H°	H_m°	h^0	\bar{h}°	h^0	\underline{H}°	H°	H^0	H°
standard enthalpy of formation	\DeltafHm^\std	ΔH_f°	\bar{h}_f°	\bar{h}_f°	ΔH_f°	$\Delta_f H_m^\circ$	Δh_f^0	\bar{h}_f°	Δh_f^0	$\Delta_f \underline{H}^\circ$	ΔH_{f298}°	ΔH_f^0	ΔH_F°
standard enthalpy of formation of i	\DeltafHm^\std_i	$\Delta H_{f,i}^\circ$	$\bar{h}_{f,i}^\circ$	$\bar{h}_{f,i}^\circ$	$\Delta H_{f,i}^\circ$	$\Delta_f H_{mi}^\circ$	$(\Delta h_f^0)_i$	$(\bar{h}_f^\circ)_i$	$\Delta h_{f,i}^0$	$\Delta_f \underline{H}_i^\circ$	ΔH_{f298i}°	ΔH_{fi}^0	$\Delta H_{F,i}^\circ$
standard enthalpy of reaction	\DeltarxnHm^\std	$\Delta H_{\text{rxn}}^\circ$	$\bar{h}_{\text{rxn}}^\circ$	\bar{h}_R°	ΔH°	ΔH_m	Δh_{rxn}^0	$\Delta \bar{h}_{\text{rxn}}^\circ$	Δh_{rxn}^0	$\Delta_{\text{rxn}} \underline{H}^\circ$	ΔH°	ΔH_{rx}^0	ΔH_R°
standard free energy of formation	\DeltafGm^\std	ΔG_f°	\bar{g}°	\bar{g}_f°	ΔG_f°	$\Delta_f G_m^\circ$	Δg_f^0	\bar{g}_f°	Δg_f^0	$\Delta_f \underline{G}^\circ$	ΔG_{f298}°	ΔG_f^0	ΔG_F°
standard free energy of formation of i	\DeltafGm_i^\std	$\Delta G_{f,i}^\circ$	$\bar{g}_{f,i}^\circ$	$\bar{g}_{f,i}^\circ$	$\Delta G_{f,i}^\circ$	$\Delta_f G_{mi}^\circ$	$(\Delta g_f^0)_i$	$(\bar{g}_f^\circ)_i$	$\Delta g_{f,i}^0$	$\Delta_f \underline{G}_i^\circ$	ΔG_{f298i}°	ΔG_{fi}^0	$\Delta G_{F,i}^\circ$
standard free energy of reaction	\DeltarxnGm^\std	$\Delta G_{\text{rxn}}^\circ$	$\bar{g}_{\text{rxn}}^\circ$	\bar{g}_R°	ΔG°	ΔG_m	Δg_{rxn}^0	$\Delta \bar{g}_{\text{rxn}}^\circ$	Δg_{rxn}^0	$\Delta_{\text{rxn}} \underline{G}^\circ$	ΔG°	ΔG_{rx}^0	ΔG_R°
standard fugacity	\fstd	f°	f°	f°	f°	f°	f^0	f°	f^0	\bar{f}°	f°	f°	f°
standard Gibbs free energy	\Gm^\std	G°	\bar{g}°	\bar{g}°	G°	G_m°	g^0	\bar{g}°	g^0	\underline{G}°	G°	G^0	G°
standard Gibbs free energy of i	\Gm^\std_i	G_i°	\bar{g}_i°	\bar{g}_i°	G_i°	G_{mi}°	g_i^0	\bar{g}_i°	g_i^0	\underline{G}_i°	G_i°	G_i^0	G_i°
standard pressure	\Pstd	P°	P_0	P_0	P°	P°	P^0	p_{ref}	P^0	1 bar	P°	P^*	P°
vapor pressure	\Pvap	p^{sat}	p^{sat}	p^v	p^{std}	p	p^{sat}	p_{ref}	p^s	p^{vap}	p^{sat}	p^{sat}	p^{sat}
volume (total)	\Vt	\underline{V}	V	V^3	\underline{V}	V	V	V	V	V	V^t	\underline{V}	\underline{V}
volume (molar)	\Vm	V	\bar{v}	\bar{v}^3	V	V_m	v	\bar{v}	v	\underline{V}	V	V	V
volume (specific)	\Vs	\hat{V}	v	v^3	V	\hat{V}	\hat{v}	v	\hat{v}	\hat{V}	V	v	\hat{V}
volume (molar, residual)	\VR	V^R	\bar{v}^R	\bar{v}^R3	V^R	V_m^R	v^{dep}	\bar{v}^R	v^R	\underline{V}^r	V^R	V^R	V^R
volume (molar, excess)	\VE	V^E	\bar{v}^E	\bar{v}^E3	V^E	V_m^E	v^E	\bar{v}^E	v^E	$\underline{V}^{\text{ex}}$	V^E	V^{EX}	V^{EX}
volume (partial molar, excess)	\VEpm_i	\bar{V}_i^E	\bar{v}_i^E	\bar{v}_i^E3	\bar{V}_i^E	V_{mi}^E	\bar{V}_i^E	\bar{V}_i^E	\bar{v}_i^E	\bar{V}_i^{ex}	\bar{V}_i^E	\bar{V}_i^{EX}	\bar{V}_i^{EX}
work (total)	\Wt	\underline{W}	W	W	\underline{W}	W	W	W	W	W	W^t	\underline{W}	\underline{W}
work (molar)	\Wm	W	\bar{w}	\bar{w}	W	W_m	w	\bar{w}	w	\underline{W}	W	W	W
work (specific)	\Ws	\hat{W}	w	w	W	\hat{W}	\hat{w}	w	\hat{w}	\hat{W}	W	w	\hat{W}

B Implementation

We set up some non-standard token comparison variants; these are designed to catch both `\ncomponents=\ncomponents` and `\ncomponents=C` (using the default options); we have to define `\ncomponents` to be expandable to make these work at all.

```
1 \ExplSyntaxOn
2 \cs_generate_variant:Nn \tl_if_eq:nnTF { xxTF }
```

This package requires the `amstext` package, as `\text` is used to handle `\sat`, `\IS`, `\IG`, `\IGM`, `\Henrymol`, `\fusion`, `\reaction`, `\vaporization`, and `\sublimation` by default, as well as several other macros defined by package options.

```
3 \RequirePackage{amstext}
```

B.1 Symbols Controlled by Package Options

We set some symbols prior to declaring the package options. The default symbols follow package option `EUAGHan`, even though the macros follow the option `EUFGHAN`.

`\dbar` The way `\dbar` is defined depends on the typeface you are using. We try to determine, at `\begin{document}`, which typeface you chose based on the packages that are loaded and some of their internal definitions. The `thermodynamics` package currently supports Computer Modern (the default or through `lmodern`), Palatino (through `pxfonts` or `newpxmath`), Times (through `txfonts`, `mathptmx`, or `newtxmath`), Utopia (through `mathdesign`), Bitstream Charter (through `mathdesign`), and Garamond (through `mathdesign`). Definitions of `\dbar` (with `\newcommand*`, `\providecommand*`, `\NewDocumentCommand`, or `\ProvideDocumentCommand`) in the preamble will override the ones here.

```
4 \AtBeginDocument{
5   \ifpackageloaded{pxfonts}{%
6     \ProvideDocumentCommand{\dbar}{}
7       {\mkern5mu\mathchar'26\mkern-10mu d}
8   }{}
9   \ifpackageloaded{newpxmath}{%
10    \ProvideDocumentCommand{\dbar}{}
11      {\mkern5mu\mathchar'26\mkern-10mu d}
12  }{}
13  \ifpackageloaded{txfonts}{%
14    \ProvideDocumentCommand{\dbar}{}
15      {\mkern5mu\mathchar'26\mkern-11mu d}
16  }{}
17  \ifpackageloaded{mathptmx}{%
18    \ProvideDocumentCommand{\dbar}{}
19      {\mkern5mu\mathchar'26\mkern-10mu d}
20  }{}
21  \ifpackageloaded{newtxmath}{%
22    \ProvideDocumentCommand{\dbar}{}
23      {\mkern5mu\mathchar'26\mkern-11mu d}
24  }{}
25  \ifpackageloaded{mathdesign}{%
26    \tl_const:Nn \c_@@_charter_tl {mdbch}
27    \tl_const:Nn \c_@@_utopia_tl {mdput}
28    \tl_const:Nn \c_@@_garamond_tl {mdugm}
29    \tl_if_eq:NNT \MD@default@family \c_@@_utopia_tl
30    { \ProvideDocumentCommand{\dbar}{}

```

```

31      {\mkern5mu\mathchar'26\mkern-20mu d}
32    }
33    \tl_if_eq:NNT \MD@default@family \c_@@_charter_tl
34    { \ProvideDocumentCommand{\dbar}{}
35      {\mkern5mu\mathchar'26\mkern-15mu d}
36    }
37    \tl_if_eq:NNT \MD@default@family \c_@@_garamond_tl
38    { \ProvideDocumentCommand{\dbar}{}
39      {\mkern5mu\mathchar'26\mkern-17mu d}
40    }
41  }{}
42  % Defaults to Computer Modern
43  \ProvideDocumentCommand{\dbar}{}
44    {\mkern3mu\mathchar'26\mkern-12mu d}
45 }

```

Symbols are defined for the total energy, internal energy, Helmholtz free energy, Gibbs free energy, grand potential (Landau free energy), enthalpy, entropy, surface area, volume, number of moles, heat, and work; these are E , U , A , G , Ω , H , S , a , V , n , Q , and W , respectively. These commands should not be used on their own, but rather accessed through the macros `\Ut`, `\Um`, and `\Us` (using the internal energy as an example).

The default symbols are not intended to be easy to change—the intended mechanism is through package options. If you want to use a non-standard symbol that is not available through one of the package options, you can redefine the internal token lists inside `\ExplSyntaxOn... \ExplSyntaxOff`. For example,

```

\ExplSyntaxOn
  \tl_gset:Nn \g__thermodynamics_Helmholtz_symbol {H}
  \tl_gset:Nn \g__thermodynamics_enthalpy_symbol {h}
\ExplSyntaxOff

```

would define the ill-advised notation that I have nonetheless heard of that uses H for Helmholtz free energy and h for enthalpy. Note that the macros for temperature and pressure are only used inside the definitions of the compressibilities, expansivities, and heat capacities; there is no user-level macro for the temperature or the pressure, so it is up to the user to use consistent symbols for those properties.

```

46 \tl_new:N \g_@@_total_energy_symbol
47 \tl_new:N \g_@@_internal_energy_symbol
48 \tl_new:N \g_@@_Helmholtz_symbol
49 \tl_new:N \g_@@_Gibbs_symbol
50 \tl_new:N \g_@@_Landau_symbol
51 \tl_new:N \g_@@_enthalpy_symbol
52 \tl_new:N \g_@@_entropy_symbol
53 \tl_new:N \g_@@_area_symbol
54 \tl_new:N \g_@@_volume_symbol
55 \tl_new:N \g_@@_mole_symbol
56 \tl_new:N \g_@@_heat_symbol
57 \tl_new:N \g_@@_work_symbol
58 \tl_new:N \g_@@_temperature_symbol
59 \tl_new:N \g_@@_pressure_symbol
60
61 \tl_gset:Nn \g_@@_total_energy_symbol E
62 \tl_gset:Nn \g_@@_internal_energy_symbol U
63 \tl_gset:Nn \g_@@_Helmholtz_symbol A

```

```

64\tl_gset:Nn \g_@@_Gibbs_symbol G
65\cs_if_exist:NTF \Omegait
66{ \tl_gset:Nn \g_@@_Landau_symbol \Omegait }
67{ \tl_gset:Nn \g_@@_Landau_symbol \Omega }
68\tl_gset:Nn \g_@@_enthalpy_symbol H
69\tl_gset:Nn \g_@@_entropy_symbol S
70\tl_gset:Nn \g_@@_area_symbol a
71\tl_gset:Nn \g_@@_volume_symbol V
72\tl_gset:Nn \g_@@_mole_symbol n
73\tl_gset:Nn \g_@@_heat_symbol Q
74\tl_gset:Nn \g_@@_work_symbol W
75\tl_gset:Nn \g_@@_temperature_symbol T
76\tl_gset:Nn \g_@@_pressure_symbol P

```

We then define two functions and several lengths that we shall use when drawing rules above or below a symbol. The default is to use underlined symbols for extensive quantities, plain symbols for molar quantities, and carets for specific quantities, but this can be changed using package options.

```

77\cs_new:Nn \@@_underline:n
78{ \mkern1mu\underline{\mkern-1mu #1\mkern-4mu}\mkern4mu }
79\cs_new:Nn \@@_overline:n
80{ \mkern2mu\overline{\mkern-2mu #1\mkern-1mu}\mkern1mu }

```

`\PartialOpen` We define three commands to use to denote the beginning and end of partial derivatives. These symbols can be customized by package options. Default is parentheses, `\PartialClose` meaning that `\[\Partial{f}{x}{y} \]` renders as

$$\left(\frac{\partial f}{\partial x}\right)_y$$

with the defaults. The macro `\PartialEmptyClose` is used when the last argument to `\Partial` is empty, which is important for the bar option to the document class or inside the `thermobar` environment.

```

81\tl_new:N \l_@@_PartialOpen_tl
82\tl_new:N \l_@@_PartialEmptyClose_tl
83\tl_new:N \l_@@_PartialClose_tl
84
85\tl_set:Nn \l_@@_PartialOpen_tl { ( }
86\tl_set:Nn \l_@@_PartialClose_tl { ) }
87\tl_set:Nn \l_@@_PartialEmptyClose_tl { } }

```

B.2 Package Options

We declare a bunch of options for which sets of symbols to use. These are summarized in Table 3.

```

88\DeclareOption{EUAGHan}{}% the default
89\DeclareOption{EUAGHaN}{\tl_gset:Nn \g_@@_mole_symbol N}%
90\DeclareOption{EUHAGan}{\ExecuteOptions{EUAGHan}}
91\DeclareOption{EUHAGaN}{\ExecuteOptions{EUAGHaN}}
92\DeclareOption{EUFGHan}{}%
93 \tl_gset:Nn \g_@@_Helmholtz_symbol F
94 \tl_gset:Nn \g_@@_area_symbol A
95 }
96\DeclareOption{EUFGHAN}{}%
97 \tl_gset:Nn \g_@@_Helmholtz_symbol F

```

```

98 \tl_gset:Nn \g_@@_area_symbol A
99 \tl_gset:Nn \g_@@_mole_symbol N
100 }
101 \DeclareOption{EEFGHAn}{%
102 \tl_gset:Nn \g_@@_total_energy_symbol {\mathcal{E}}
103 \tl_gset:Nn \g_@@_internal_energy_symbol E
104 \tl_gset:Nn \g_@@_Helmholtz_symbol F
105 \tl_gset:Nn \g_@@_area_symbol A
106 }
107 \DeclareOption{EEFGHAN}{%
108 \tl_gset:Nn \g_@@_total_energy_symbol {\mathcal{E}}
109 \tl_gset:Nn \g_@@_internal_energy_symbol E
110 \tl_gset:Nn \g_@@_Helmholtz_symbol F
111 \tl_gset:Nn \g_@@_area_symbol A
112 \tl_gset:Nn \g_@@_mole_symbol N
113 }
114 \DeclareOption{EEFGHan}{%
115 \tl_gset:Nn \g_@@_total_energy_symbol {\mathcal{E}}
116 \tl_gset:Nn \g_@@_internal_energy_symbol E
117 \tl_gset:Nn \g_@@_Helmholtz_symbol F
118 }
119 \DeclareOption{EEFGHaN}{%
120 \tl_gset:Nn \g_@@_total_energy_symbol {\mathcal{E}}
121 \tl_gset:Nn \g_@@_internal_energy_symbol E
122 \tl_gset:Nn \g_@@_Helmholtz_symbol F
123 \tl_gset:Nn \g_@@_mole_symbol N
124 }
125 \DeclareOption{EEAGHan}{%
126 \tl_gset:Nn \g_@@_total_energy_symbol {\mathcal{E}}
127 \tl_gset:Nn \g_@@_internal_energy_symbol E
128 }
129 \DeclareOption{EEAGHaN}{%
130 \tl_gset:Nn \g_@@_total_energy_symbol {\mathcal{E}}
131 \tl_gset:Nn \g_@@_internal_energy_symbol E
132 \tl_gset:Nn \g_@@_mole_symbol N
133 }
134 \DeclareOption{EUAGHAN}{%
135 \tl_gset:Nn \g_@@_area_symbol {\mathcal{A}}
136 }
137 \DeclareOption{EUAGHAN}{%
138 \tl_gset:Nn \g_@@_area_symbol {\mathcal{A}}
139 \tl_gset:Nn \g_@@_mole_symbol N
140 }
141 \DeclareOption{EUFGHan}{%
142 \tl_gset:Nn \g_@@_Helmholtz_symbol F
143 }
144 \DeclareOption{EUFGHaN}{%
145 \tl_gset:Nn \g_@@_Helmholtz_symbol F
146 \tl_gset:Nn \g_@@_mole_symbol N
147 }

```

The delta option redefines `\dbar` to produce the symbol δ . The default is to use a d with a slash through it (\mathcal{d}) for inexact differentials unless the user overrides it with this option. The macro could also be redefined manually, of course.

```

148 \DeclareOption{delta}{\cs_set_eq:NN \dbar \delta }

```

Next, we define options for the set of notation. The default is `intensive-plain`,

which produces things like V for molar volume, \underline{V} for total volume, and \hat{V} for specific volume. The symbols themselves are produced via internal (non-user-facing) macros.

```

149 \cs_new:Nn \@@_extensive:n {#1}
150 \cs_new:Nn \@@_intensive:n {#1}
151 \cs_new:Nn \@@_specific:n {\hat{#1}}
152
153 \cs_new:Npn \@@_set_intensive_plain
154 {
155   \cs_set:Nn \@@_extensive:n {\@@_underline:n{##1}}
156   \cs_set:Nn \@@_intensive:n {##1}
157 }
158 \cs_new:Npn \@@_set_extensive_plain
159 {
160   \cs_set:Nn \@@_extensive:n {##1}
161   \cs_set:Nn \@@_intensive:n {\@@_underline:n{##1}}
162 }
163 \cs_new:Npn \@@_set_lowercase_pms
164 {
165   \RenewDocumentCommand{\partialmolar}{m}
166   {
167     \tl_set:Nn \l_@@_pm_symbol_tl {\text_lowercase:n {##1}}
168     \@@_generic_pm:
169   }
170 }
171 \cs_new:Npn \@@_set_intensive_lowercase
172 {
173   \cs_set:Nn \@@_extensive:n {\text_uppercase:n {##1}}
174   \cs_set:Nn \@@_intensive:n {\text_lowercase:n {##1}}
175   \cs_set:Nn \@@_specific:n {\hat{\text_lowercase:n {##1}}}
176 }
177 \cs_new:Npn \@@_set_extensive_superscripts
178 {
179   \cs_set:Nn \@@_extensive_one:n
180   { \c_math_superscript_token {##1\l_@@_super_separator_tl t} }
181   \cs_set:Nn \@@_extensive:n
182   {
183     \peek_catcode_remove:NTF \c_math_superscript_token
184     { ##1 \@@_extensive_one:n }
185     { ##1 \c_math_superscript_token t }
186   }
187   \cs_set:Nn \@@_intensive:n {##1}
188 }
189 \DeclareOption{extensive-plain}{\@@_set_extensive_plain}
190 \DeclareOption{intensive-plain}{\@@_set_intensive_plain} % the default
191 \DeclareOption{intensive-lowercase}{% PLEASE don't use this!
192   \@@_set_intensive_lowercase
193   \AtEndOfPackage{
194     \@@_set_lowercase_pms
195   }
196 }
197 \DeclareOption{extensive-superscript}{%
198   \@@_set_extensive_superscripts
199 %^^A \AtEndOfPackage{%
200 %^^A   \RenewDocumentCommand{\URt}{}{\g_@@_internal_energy_symbol
201 %^^A     \c_math_superscript_token{\residual,t}}
202 %^^A   \RenewDocumentCommand{\HRT}{}{\g_@@_enthalpy_symbol

```

```

203 %^^A \c_math_superscript_token{\residual,t}}
204 %^^A \RenewDocumentCommand{\FRt}{}{\g_@@_Helmholtz_symbol
205 %^^A \c_math_superscript_token{\residual,t}}
206 %^^A \RenewDocumentCommand{\GRt}{}{\g_@@_Gibbs_symbol
207 %^^A \c_math_superscript_token{\residual,t}}
208 %^^A \RenewDocumentCommand{\VRt}{}{\g_@@_volume_symbol
209 %^^A \c_math_superscript_token{\residual,t}}
210 %^^A \RenewDocumentCommand{\SRt}{}{\g_@@_entropy_symbol
211 %^^A \c_math_superscript_token{\residual,t}}
212 %^^A \RenewDocumentCommand{\UET}{}{\g_@@_internal_energy_symbol
213 %^^A \c_math_superscript_token{\excess,t}}
214 %^^A \RenewDocumentCommand{\HET}{}{\g_@@_enthalpy_symbol
215 %^^A \c_math_superscript_token{\excess,t}}
216 %^^A \RenewDocumentCommand{\FET}{}{\g_@@_Helmholtz_symbol
217 %^^A \c_math_superscript_token{\excess,t}}
218 %^^A \RenewDocumentCommand{\GET}{}{\g_@@_Gibbs_symbol
219 %^^A \c_math_superscript_token{\excess,t}}
220 %^^A \RenewDocumentCommand{\VET}{}{\g_@@_volume_symbol
221 %^^A \c_math_superscript_token{\excess,t}}
222 %^^A \RenewDocumentCommand{\SET}{}{\g_@@_entropy_symbol
223 %^^A \c_math_superscript_token{\excess,t}}
224 %^^A }
225 }

```

thermolongpm (*env.*) The longpm option (the default) makes partial molar quantities that look like \overline{U}_i^E . The thermoshortpm (*env.*) shortpm option (which is used by several textbooks) makes partial molar quantities look like \overline{U}_i^E . The environments thermolongpm and thermoshortpm do the same thing locally.

```

226 \bool_new:N \l_@@_longpm_bool
227 \DeclareOption{longpm}{\bool_set_true:N \l_@@_longpm_bool}
228 \cs_set_eq:NN \@@_overline_copy:n \@@_overline:n
229 \DeclareOption{shortpm}{
230   \bool_set_false:N \l_@@_longpm_bool
231   \AtEndOfPackage{
232     \RenewDocumentCommand{\cPpm}{}
233     {
234       \cs_set:Nn \@@_overline:n {##1}
235       \partialmolar{\cPpmshort}
236     }
237     \RenewDocumentCommand{\cVpm}{}
238     {
239       \cs_set:Nn \@@_overline:n {##1}
240       \partialmolar{\cVpmshort}
241     }
242   }
243 }
244 \NewDocumentEnvironment{thermolongpm}{}{%
245   \bool_set_true:N \l_@@_longpm_bool
246 }{}
247 \NewDocumentEnvironment{thermoshortpm}{}{%
248   \bool_set_false:N \l_@@_longpm_bool
249   \RenewSubscriptedSymbol{\cPpm}{\@@_overline:n \heatcapacitiesymbol}
250   {\g_@@_pressure_symbol}
251   \RenewSubscriptedSymbol{\cVpm}{\@@_overline:n \heatcapacitiesymbol}
252   {\g_@@_volume_symbol}
253 }{}

```

The next two options choose whether variables held constant are subscripted (the default) or placed next to the function. The difference is

$$\left(\frac{\partial U}{\partial S}\right)_V \quad \text{versus} \quad \left(\frac{\partial U(S,V)}{\partial S}\right)$$

for `subscripts` and `nosubscripts`, respectively.

```
254 \bool_new:N \l_@@_subscripted_bool
255 \bool_set_true:N \l_@@_subscripted_bool
256 \DeclareOption{subscripts}{\bool_set_true:N \l_@@_subscripted_bool}
257 \DeclareOption{nosubscripts}{\bool_set_false:N \l_@@_subscripted_bool}
```

These options change how `\Partial` and friends render derivatives. The default is parentheses, but other options include brackets, braces, a vertical bar on the right side, or plain (undecorated) derivatives.

```
258 \DeclareOption{parentheses}{}
259 \DeclareOption{brackets}{%
260   \tl_set:Nn \l_@@_PartialOpen_tl {[ ]
261   \tl_set:Nn \l_@@_PartialClose_tl { ] }
262   \tl_set:Nn \l_@@_PartialEmptyClose_tl { ] }
263 }
264 \DeclareOption{braces}{%
265   \tl_set:Nn \l_@@_PartialOpen_tl { { }
266   \tl_set:Nn \l_@@_PartialClose_tl { } }
267   \tl_set:Nn \l_@@_PartialEmptyClose_tl { } }
268 }
269 \DeclareOption{bar}{%
270   \tl_set:Nn \l_@@_PartialOpen_tl { . }
271   \tl_set:Nn \l_@@_PartialClose_tl { \rvert }
272   \tl_set:Nn \l_@@_PartialEmptyClose_tl { . }
273 }
274 \DeclareOption{plain-derivatives}{% This implies dU(S,V,N)/dS notation
275   \tl_set:Nn \l_@@_PartialOpen_tl { . }
276   \tl_set:Nn \l_@@_PartialClose_tl { . }
277   \tl_set:Nn \l_@@_PartialEmptyClose_tl { . }
278   \ExecuteOptions{nosubscripts}
279 }
```

B.3 The Number of Moles Macros

`\ncomponents` We define the number of components, default C , for use in the “all moles” and related macros. The command is expandable so we can perform comparisons to user-entered values.

```
280 \NewExpandableDocumentCommand \ncomponents {} {C}
```

`\allNs` Several macros define a shorthand for “moles of all species” (`\allNs`) and “moles of all species except” (`\allNsbut`), as well as similar quantities for masses (`\allMs`, `\allYs` `\allMsbut`) and chemical potentials (`\allmus`, `\allmusbut`), which occur frequently in mixture thermodynamics. The default is for `\allNs` to become \vec{n} and `\allNsbut{i}` to become $n_{j \neq i}$. The optional argument changes which index (default: j) to use in the left side of the inequality.⁴ Essentially identical commands are defined for chemical potentials and masses: `\allmus` and `\allmusbut` and `\allMs` and `\allMsbut`, respectively.

⁴The index j is automatically replaced with k if the user issues `\allNsbut{j}`.


```

281 \NewDocumentCommand{\allNs}{0{i}}{\allcomponents[#1]{\Nt}}
282 \NewDocumentCommand{\allXs}{0{i}}{\allcomponents[#1]{x}}
283 \NewDocumentCommand{\allYs}{0{i}}{\allcomponents[#1]{y}}
284 \NewDocumentCommand{\allmus}{0{i}}{\allcomponents[#1]{\mu}}
285 \NewDocumentCommand{\allMs}{0{i}}{\allcomponents[#1]{m}}
286 \NewDocumentCommand{\allWs}{0{i}}{\allcomponents[#1]{w}}

```

`\allNsbut` Similar commands are defined for mole fractions (`\allXs`, `\allYs`, etc.), but these
`\allXsbut` assume the last mole fraction is *not* one of the variables—that is, `\allXsbut` and
`\allYsbut` `\allYsbut` assume the argument *and* `\ncomponents` are held constant. For example,

```

\allMsbut \[ \Partial{\Gm}{T}{P,\allXs} = -\Sm \quad
\allWsbut \Partial{\Gm}{x_i}{T,P,\allXsbut{i}} \neq \Gpm_i \]

```

yields

$$\left(\frac{\partial G}{\partial T}\right)_{P,\vec{x}} = -S \quad \left(\frac{\partial G}{\partial x_i}\right)_{T,P,x_{j\neq i},C} \neq \overline{G}_i.$$

```

287 \NewDocumentCommand{\allNsbut}{0{j} m} {\allbut[#1]{#2}{\Nt}}
288 \NewDocumentCommand{\allXsbut}{0{j} m} {\allbutlastand[#1]{#2}{x}}
289 \NewDocumentCommand{\allYsbut}{0{j} m} {\allbutlastand[#1]{#2}{y}}
290 \NewDocumentCommand{\allmusbut}{0{j} m} {\allbut[#1]{#2}{\mu}}
291 \NewDocumentCommand{\allMsbut}{0{j} m} {\allbut[#1]{#2}{m}}
292 \NewDocumentCommand{\allWsbut}{0{j} m} {\allbutlastand[#1]{#2}{w}}

```

`\allbutlastand` The `\allcomponents`, `\allbut`, and `\allbutlastand` macros can be used to define
`\allbut` new entities; say, if you want to use z_i as a mole fraction, then use

```

\allcomponents \NewDocumentCommand{\allZsbut}{0{j} m}{\allbutlastand[#1]{#2}{z}}

```

Similarly, something meaning the concentrations of every species could be defined
via

```

\NewDocumentCommand{\allCs}{0{}}{\allcomponents{C}}

293 \NewDocumentCommand{\allcomponents}{0{ } m}{\vec{#2}}
294 \NewDocumentCommand{\allbut}{0{j} m m}
295 { \tl_if_eq:nnTF {#1} {#2}
296   { {#3}\c_math_subscript_token{k \neq #2} }
297   { {#3}\c_math_subscript_token{#1 \neq #2} }
298 }
299 \NewDocumentCommand{\allbutlastand}{0{j} m m}
300 { \tl_if_eq:xxTF {#2} {\ncomponents}
301   { {#3}\c_math_subscript_token{#1 \neq #2} }
302   { \tl_if_eq:nnTF {#1} {#2}
303     { {#3}\c_math_subscript_token{k \neq #2,\ncomponents} }
304     { {#3}\c_math_subscript_token{#1 \neq #2,\ncomponents} }
305   }
306 }

```

B.4 Package Options

We define two package options that change how to render `\allNs` and friends.

```

307 \DeclareOption{moles-index}{}
308 \DeclareOption{moles-range}{ \@@_set_moles_range }
309 \cs_new:Npn \@@_set_moles_range {%
310   \RenewDocumentCommand{\allcomponents}{0{ } m}
311   { {#2}\c_math_subscript_token 1,\dots,

```

```

312      {##2}\c_math_subscript_token{\ncomponents} }
313 \RenewDocumentCommand{\allbut}{0{j} m m}
314 { \tl_if_eq:nnTF {##2} {1}
315   { {##3}\c_math_subscript_token 2,\dots,
316     {##3}\c_math_subscript_token{\ncomponents} }
317   { \tl_if_eq:xxTF {##2} {\ncomponents}
318     { {##3}\c_math_subscript_token 1,\dots,
319       {##3}\c_math_subscript_token{\ncomponents-1} }
320     { {##3}\c_math_subscript_token 1,\dots,
321       [{##3}\c_math_subscript_token{##2}],
322       \dots,{##3}\c_math_subscript_token{\ncomponents} } }
323   }
324 }
325 \RenewDocumentCommand{\allbutlastand}{0{j} m m}
326 { \tl_if_eq:nnTF {##2} {1}
327   { {##3}\c_math_subscript_token 2,\dots,
328     {##3}\c_math_subscript_token{\ncomponents-1} }
329   { \tl_if_eq:xxTF {##2} {\ncomponents}
330     { {##3}\c_math_subscript_token 1,\dots,
331       {##3}\c_math_subscript_token{\ncomponents-1}
332     }
333     { \tl_if_eq:xxTF {##2} {\ncomponents-1}
334       { {##3}\c_math_subscript_token 1,\dots,
335         {##3}\c_math_subscript_token{\ncomponents-2} }
336       {
337         {##3}\c_math_subscript_token 1,\dots,
338         [{##3}\c_math_subscript_token{##2}],\dots,
339         {##3}\c_math_subscript_token{\ncomponents-1}
340       }
341     }
342   }
343 }
344 }

```

The remaining options define textbook-specific notation.

```

345 \DeclareOption{Bejan}{
346   \ExecuteOptions{EUF GHAN,intensive-lowercase,delta,shortpm}
347   \AtEndOfPackage{
348     \cs_set:Nn \@@_specific:n {\text_lowercase:n {#1}}
349     \cs_set:Nn \@@_intensive:n {\bar{\text_lowercase:n {#1}}}
350     \cs_set:Nn \@@_extensive:n {\text_uppercase:n {#1}}
351     \tl_gset:Nn \g_@@_volume_symbol v
352     \RenewExpandableDocumentCommand{\ncomponents}{}{n}
353     \RenewSubscriptedSymbol{\Lt}{\g_@@_Helmholtz_symbol}{\mu}
354     \RenewSubscriptedSymbol{\Lm}
355       {\bar{\text_lowercase:n \g_@@_Helmholtz_symbol}}{\mu}
356     \RenewSubscriptedSymbol{\Ls}
357       {\text_lowercase:n \g_@@_Helmholtz_symbol}{\mu}
358     \cs_new:Npn \Delta_vap_sym {} {}
359     \NewSubscriptedSymbol{\Delta_vap}{\Delta_vap_sym}{\fg}
360     \RenewDocumentCommand{\Deltavap}{m}{
361       \cs_set:Npn \Delta_vap_sym {} {#1}
362       \Delta_vap
363     }
364     \cs_new:Npn \Delta_fus_sym {} {}
365     \NewSubscriptedSymbol{\Delta_fus}{\Delta_fus_sym}{\sf}
366     \RenewDocumentCommand{\Deltafus}{m}{

```

```

367 \cs_set:Npn \Delta_fus_sym {} {#1}
368 \Delta_fus
369 }
370 \cs_new:Npn \Delta_sub_sym {} {}
371 \NewSubscriptedSymbol{\Delta_sub}{\Delta_sub_sym}{sg}
372 \RenewDocumentCommand{\Deltasub}{m}{
373 \cs_set:Npn \Delta_sub_sym {} {#1}
374 \Delta_sub
375 }
376 \RenewExpandableDocumentCommand{\heatcapacitiesymbol}{}{c}
377 \RenewSubscriptedSymbol{\cV}{\@@_intensive:n {\heatcapacitiesymbol}}
378 {\text_lowercase:n \g_@@_volume_symbol}
379 \RenewSubscriptedSymbol{\cVt}{\@@_extensive:n {\heatcapacitiesymbol}}
380 {\text_uppercase:n \g_@@_volume_symbol}
381 \RenewSubscriptedSymbol{\cVpm}{\heatcapacitiesymbol}
382 {\text_lowercase:n \g_@@_volume_symbol}
383 \RenewSubscriptedSymbol{\cPpm}{\heatcapacitiesymbol}
384 {\text_lowercase:n \g_@@_pressure_symbol}
385 \RenewExpandableDocumentCommand{\xrxn}{}{\zeta}
386 \RenewDocumentCommand{\fmix}{}{f}
387 \RenewDocumentCommand{\Qm}{}{Q}
388 \RenewDocumentCommand{\Qs}{}{Q}
389 \RenewExpandableDocumentCommand{\kappaT}{}{\compressibilitysymbol}
390 \RenewExpandableDocumentCommand{\expansivitysymbol}{}{\beta}
391 \RenewExpandableDocumentCommand{\alphaP}{}{\expansivitysymbol}
392 \RenewSubscriptedSymbol{\muJT}{\JTsymbol}{J}
393 \RenewDocumentCommand{\Deltaf}{m}
394 { \tl_set:Nn \l_@@_Deltaf_sym_tl {#1} \@@_Deltaf }
395 \RenewDocumentCommand{\Pstd}{}{P\c_math_subscript_token 0}
396 \RenewDocumentCommand{\Wm}{}{W}
397 \RenewDocumentCommand{\Ws}{}{W}
398 \cs_new:Nn \@@_fpure_one:n
399 {
400 f\c_math_subscript_token{#1}
401 \peek_catcode_remove:NF \c_math_superscript_token
402 { \c_math_superscript_token\bullet }
403 }
404 \RenewDocumentCommand{\fpure}{}
405 {
406 \peek_catcode_remove:NTF \c_math_subscript_token
407 { \@@_fpure_one:n }
408 { f }
409 }
410 \cs_new:Nn \@@_phipure_one:n
411 {
412 \phi\c_math_subscript_token{#1}
413 \peek_catcode:NF \c_math_superscript_token
414 { \c_math_superscript_token\bullet }
415 }
416 \RenewDocumentCommand{\phipure}{}
417 {
418 \peek_catcode_remove:NTF \c_math_subscript_token
419 {
420 \@@_phipure_one:n
421 }
422 {\phi}

```

```

423     }
424     \RenewDocumentCommand{\phimix}{}{\phi}
425 }
426 }
427 \DeclareOption{CBK}{
428   \ExecuteOptions{EUAGHAN,intensive-lowercase,delta,shortpm}
429   \AtEndOfPackage{
430     \cs_set:Nn \@@_overline:n {\widetilde #1}
431     \cs_set:Nn \@@_overline_copy:n {\widetilde #1}
432     \cs_set:Nn \@@_specific:n {\text_lowercase:n {#1}}
433     \cs_set:Nn \@@_intensive:n {\bar{\text_lowercase:n{#1}}}
434     \tl_gset:Nn \g_@@_area_symbol A
435     \RenewExpandableDocumentCommand{\reaction}{}{R}
436     \cs_new:Npn \Delta_vap_sym {} {}
437     \NewSubscriptedSymbol{\Delta_vap}{\Delta_vap_sym}{{fg}}
438     \RenewDocumentCommand{\Deltavap}{m}{
439       \cs_set:Npn \Delta_vap_sym {} {#1}
440       \Delta_vap
441     }
442     \cs_new:Npn \Delta_fus_sym {} {}
443     \NewSubscriptedSymbol{\Delta_fus}{\Delta_fus_sym}{{sf}}
444     \RenewDocumentCommand{\Deltafus}{m}{
445       \cs_set:Npn \Delta_fus_sym {} {#1}
446       \Delta_fus
447     }
448     \cs_new:Npn \Delta_sub_sym {} {}
449     \NewSubscriptedSymbol{\Delta_sub}{\Delta_sub_sym}{{sg}}
450     \RenewDocumentCommand{\Deltasub}{m}{
451       \cs_set:Npn \Delta_sub_sym {} {#1}
452       \Delta_sub
453     }
454     \tl_gset:Nn \g_@@_pressure_symbol p
455     \RenewExpandableDocumentCommand{\compressibilitysymbol}{}{\alpha}
456     \RenewExpandableDocumentCommand{\expansivitysymbol}{}{\beta}
457     \RenewExpandableDocumentCommand{\kappaT}{}{\compressibilitysymbol}
458     \RenewExpandableDocumentCommand{\alphaP}{}{\expansivitysymbol}
459     \RenewSuperscriptedSymbol{\Psat}{P}{v}
460     \RenewDocumentCommand{\Deltaf}{m}
461     { \tl_set:Nn \l_@@_Deltaf_sym_tl {#1}
462       \@@_Deltaf
463     }
464     \RenewDocumentCommand{\Pstd}{}{P\c_math_subscript_token 0}
465     \RenewDocumentCommand{\Lm}{}{\@@_intensive:n\omega}
466     \RenewDocumentCommand{\Ls}{}{\@@_specific:n\omega}
467   }
468 }
469 \DeclareOption{ElliottLira}{
470   \ExecuteOptions{shortpm}
471   \AtEndOfPackage{
472     \cs_gset_eq:NN \@@_specific:n \@@_intensive:n
473     %^^A\RenewDocumentCommand{\partialmolar}{m}{\@@_overline:n #1}
474     \RenewDocumentCommand{\allcomponents}{0{} m}{#2}
475     \RenewDocumentCommand{\Deltarxn}{m}{\Delta #1}
476     \RenewDocumentCommand{\fusion}{}{{fus}}
477     \RenewDocumentCommand{\sublimation}{}{{sub}}
478     \RenewDocumentCommand{\vaporization}{}{{vap}}

```

```

479 \RenewDocumentCommand{\sat}{}{{sat}}
480 %^^A\RenewSubscriptedSymbol{\Henrymol}{/K}{H} %^^A FIXME
481 \RenewDocumentCommand{\IG}{}{{ig}}
482 \RenewDocumentCommand{\IGM}{}{{ig}}
483 \RenewDocumentCommand{\IS}{}{{is}}
484 \RenewDocumentCommand{\Cstd}{}{m\c_math_superscript_token \std}
485 \RenewDocumentCommand{\muJT}{}{\mu\c_math_subscript_token{JT}}
486 }
487 }
488 \DeclareOption{KlotzRosenberg}{
489 \ExecuteOptions{delta}
490 \AtEndOfPackage{
491 \tl_set:Nn \l_@@_sub_separator_tl {}
492 \cs_set:Nn \@@_extensive:n {#1}
493 \cs_new:Nn \@@_intensive_one:n
494 {
495 \c_math_superscript_token {#1}
496 \peek_catcode_remove:NTF \c_math_subscript_token
497 { \@@_intensive_three:n }
498 { \c_math_subscript_token{\mathrm{m}} }
499 }
500 \cs_new:Nn \@@_intensive_three:n
501 { \c_math_subscript_token{\mathrm{m}}\l_@@_sub_separator_tl #1 }
502 \cs_new:Nn \@@_intensive_two:n
503 {
504 \c_math_subscript_token{\mathrm{m}}\l_@@_sub_separator_tl #1}
505 \peek_catcode:NF \c_math_superscript_token
506 { \c_math_superscript_token\bullet }
507 }
508 \cs_set:Nn \@@_intensive:n
509 {
510 #1
511 \peek_catcode_remove:NTF \c_math_superscript_token
512 { \@@_intensive_one:n }
513 { \peek_catcode_remove:NTF \c_math_subscript_token
514 { \@@_intensive_two:n }
515 { \c_math_subscript_token{\mathrm{m}} } }
516 }
517 }
518 \cs_new:Nn \@@_fpure_one:n
519 {
520 f\c_math_subscript_token{#1}
521 \peek_catcode_remove:NF \c_math_superscript_token
522 { \c_math_superscript_token\bullet }
523 }
524 \RenewDocumentCommand{\fpure}{}
525 {
526 \peek_catcode_remove:NTF \c_math_subscript_token
527 { \@@_fpure_one:n }
528 { f }
529 }
530 \cs_new:Nn \@@_phipure_one:n
531 {
532 \gamma\c_math_subscript_token{#1}
533 \peek_catcode_remove:NF \c_math_superscript_token
534 { \c_math_superscript_token\bullet }

```

```

535 }
536 \RenewDocumentCommand{\phipure}{}
537 {
538   \peek_catcode_remove:NTF \c_math_subscript_token
539   { \@@_phipure_one:n }
540   { \gamma }
541 }
542 \RenewDocumentCommand{\fmix}{}{f}
543 \RenewDocumentCommand{\phimix}{}{\gamma}
544 \RenewDocumentCommand{\phisat}{}{\gamma^{\text{sat}}}
545 \cs_if_exist:NTF \gammaup
546 { \RenewDocumentCommand{\gamma}{}{\gammaup} }
547 { \cs_if_exist:NT \upgamma
548   { \RenewDocumentCommand{\gamma}{}{\upgamma} }
549 }
550 \RenewDocumentCommand{\gamarat}{}{\gamma}
551 \RenewDocumentCommand{\gammamol}{}{\gamma}
552 \cs_if_exist:NTF \muup
553 { \RenewDocumentCommand{\mu}{}{\muup} }
554 { \cs_if_exist:NT \upmu
555   { \RenewDocumentCommand{\mu}{}{\upmu} }
556 }
557 \cs_if_exist:NTF \alphaup
558 { \RenewExpandableDocumentCommand{\expansivitysymbol}{}{\alphaup} }
559 { \cs_if_exist:NTF \upalpha
560   { \RenewExpandableDocumentCommand{\compressibilitysymbol}{}{\alphaup} }
561   { \RenewExpandableDocumentCommand{\compressibilitysymbol}{}{\alpha} }
562 }
563 \cs_if_exist:NTF \betaup
564 { \RenewExpandableDocumentCommand{\compressibilitysymbol}{}{\betaup} }
565 { \cs_if_exist:NTF \upbeta
566   { \RenewExpandableDocumentCommand{\compressibilitysymbol}{}{\betaup} }
567   { \RenewExpandableDocumentCommand{\compressibilitysymbol}{}{\beta} }
568 }
569 \cs_if_exist:NTF \xiup
570 { \RenewDocumentCommand{\xrxn}{}{\xiup} }
571 { \cs_if_exist:NT \upxi
572   { \RenewDocumentCommand{\xrxn}{}{\upxi} }
573 }
574 \RenewDocumentCommand{\kappaT}{}{\compressibilitysymbol}
575 \RenewDocumentCommand{\alphaP}{}{\expansivitysymbol}
576 \RenewSubscriptedSymbol{\muJT}{\JTsymbol}{\text{J.T.}}
577 \RenewDocumentCommand{\Deltarxn}{m}{\Delta #1}
578 \RenewDocumentCommand{\Deltaf}{}{\Delta\c_math_subscript_token\formation}
579 \RenewDocumentCommand{\Deltafus}{}{\Delta}
580 \RenewDocumentCommand{\Deltavap}{}{\Delta}
581 \RenewDocumentCommand{\Deltasub}{}{\Delta}
582 %^^A FIXME: should be \Delta G_{\text{mix}},\mathrm{m} not other way around
583 \RenewDocumentCommand{\Deltamix}{m}{
584   \tl_set:Nn \l_@@_sub_separator_tl {,}
585   \Delta #1\c_math_subscript_token{mixing}
586 }
587 \RenewDocumentCommand{\excess}{}{\mathrm{E}}
588 \RenewDocumentCommand{\residual}{}{\mathrm{R}}
589 \cs_set:Nn \@@_overline:n {#1}
590 \RenewDocumentCommand{\IS}{}{\mathrm{I}}

```

```

591 \RenewDocumentCommand{\Psat}{}
592 {
593   \peek_catcode:NTF \c_math_subscript_token
594   {p\c_math_superscript_token\bullet}
595   {p}
596 }
597 \RenewDocumentCommand{\Cstd}{}{m^\std}
598 \RenewDocumentCommand{\Henryrat}{}{k}
599 \RenewDocumentCommand{\Henrymol}{}{k''}
600
601 % Fix partial molar properties
602 \cs_set:Npn \@@_pm_case_one #1
603 {
604   \l_@@_pm_symbol_tl\c_math_subscript_token{\mathrm{m}#1}
605 }
606 \cs_set:Npn \@@_pm_case_two [#1]#2
607 {
608   \l_@@_pm_symbol_tl
609   \c_math_superscript_token{#1}\c_math_subscript_token{\mathrm{m}#2}
610 }
611 \cs_set:Npn \@@_pm_case_three_part_two #1
612 {
613   \l_@@_pm_symbol_tl
614   \c_math_superscript_token{\l_@@_pm_arg_tl}
615   \c_math_subscript_token{\mathrm{m}#1}
616 }
617 \cs_set:Npn \@@_pm_case_four #1
618 {
619   \l_@@_pm_symbol_tl\c_math_superscript_token{#1}
620   \c_math_subscript_token{\mathrm{m}\l_@@_pm_arg_tl}
621 }
622 \cs_set:Npn \@@_pm_case_five
623 {
624   \l_@@_pm_symbol_tl
625   \c_math_subscript_token{\mathrm{m}\l_@@_pm_arg_tl}
626 }
627
628 % fix heat capacities
629 \RenewSubscriptedSymbol{\cP}{\heatcapacitysymbol}
630 {\g_@@_pressure_symbol\mathrm{m}}
631 \RenewSubscriptedSymbol{\cV}{\heatcapacitysymbol}
632 {\g_@@_volume_symbol\mathrm{m}}
633 \RenewDocumentCommand{\cPpm}{}{\partialmolar{\cPt}}
634 \RenewDocumentCommand{\cVpm}{}{\partialmolar{\cVt}}
635
636 \cs_set:Npn \cP_two:n #1
637 { \heatcapacitysymbol
638   \c_math_subscript_token{\g_@@_pressure_symbol\mathrm{m}}
639   \l_@@_sub_separator_tl #1}
640 \peek_catcode:NF \c_math_superscript_token
641 { \c_math_superscript_token\bullet }
642 }
643 \cs_set:Npn \cV_two:n #1
644 { \heatcapacitysymbol
645   \c_math_subscript_token{\g_@@_volume_symbol\mathrm{m}}
646   \l_@@_sub_separator_tl #1}

```

```

647     \peek_catcode:NF \c_math_superscript_token
648     { \c_math_superscript_token\bullet }
649   }
650 }
651 \AtBeginDocument{
652   \@ifpackageloaded{emf}{\RenewDocumentCommand{\Epot}{}{\emf}}
653   {%
654     \PackageWarningNoLine{thermodynamics}
655       {Package~emf~not~loaded;~load~to~make~Epot~match~Klotz~
656        and~Rosenberg's~notation}%
657   }
658 }
659 }
660 \DeclareOption{Koretsky}{
661   \ExecuteOptions{EUAGHAn,brackets,intensive-lowercase,delta,shortpm}
662   \AtEndOfPackage{
663     \tl_gset:Nn \g_@@_area_symbol {A}
664     \RenewExpandableDocumentCommand{\std}{}{o}
665     \RenewExpandableDocumentCommand{\ncomponents}{}{m}
666     \RenewDocumentCommand{\partialmolar}{m}
667     {
668       \tl_set:Nn \l_@@_pm_symbol_tl {#1}
669       \@@_generic_pm:
670     }
671     \RenewDocumentCommand{\expansivitysymbol}{}{\beta}
672     \RenewDocumentCommand{\IS}{}{\text{ideal}}
673     \RenewDocumentCommand{\residual}{}{\text{dep}}
674     \RenewDocumentCommand{\IG}{}{\text{ideal~gas}}
675     \RenewDocumentCommand{\IGM}{}{\text{ideal}}
676     \RenewDocumentCommand{\Henryrat}{}{\mathcal{H}}
677     \RenewDocumentCommand{\gammarat}{}
678       {\gamma\c_math_superscript_token\text{Henry's}}
679     \RenewDocumentCommand{\gammamol}{}{\gamma\c_math_superscript_token{m}}
680     \RenewDocumentCommand{\phipure}{}{\varphi}
681     \RenewDocumentCommand{\phimix}{}{\hat{\varphi}}
682     \RenewDocumentCommand{\phisat}{}{\varphi\c_math_superscript_token\text{sat}}
683     \cs_new:Npn \Delta_fus_sym {} {}
684     \NewSubscriptedSymbol{\Delta_fus}{\Delta_fus_sym}{\fusion}
685     \RenewDocumentCommand{\Deltafus}{m}{
686       \cs_set:Npn \Delta_fus_sym {} { \Delta #1 }
687       \Delta_fus
688     }
689     \cs_new:Npn \Delta_vap_sym {} {}
690     \NewSubscriptedSymbol{\Delta_vap}{\Delta_vap_sym}{\vaporization}
691     \RenewDocumentCommand{\Deltavap}{m}{
692       \cs_set:Npn \Delta_vap_sym {} { \Delta #1 }
693       \Delta_vap
694     }
695     \cs_new:Npn \Delta_sub_sym {} {}
696     \NewSubscriptedSymbol{\Delta_sub}{\Delta_sub_sym}{\sublimation}
697     \RenewDocumentCommand{\Deltasub}{m}{
698       \cs_set:Npn \Delta_sub_sym {} { \Delta #1 }
699       \Delta_sub
700     }
701     \tl_new:N \l_@@_Deltaf_superscript_tl
702     \tl_new:N \l_@@_Deltaf_subscript_tl

```



```

703 \bool_new:N \l_@@_Deltaf_parentheses_bool
704 \tl_new:N \l_@@_Deltaf_entity_tl
705 \cs_set:Nn \@@_Deltaf_one:n
706 {
707   \tl_set:Nn \l_@@_Deltaf_superscript_tl {#1}
708   \peek_catcode_remove:NTF \c_math_subscript_token
709   {
710     \bool_set_true:N \l_@@_Deltaf_parentheses_bool
711     \@@_Deltaf_two:n
712   }
713   {
714     \bool_if:NTF \l_@@_Deltaf_parentheses_bool
715     {
716       (\Delta\l_@@_Deltaf_entity_tl\c_math_subscript_token\formation
717        \tl_if_empty:NF \l_@@_Deltaf_superscript_tl
718        { \c_math_superscript_token\l_@@_Deltaf_superscript_tl }
719       )
720       \tl_if_empty:NF \l_@@_Deltaf_subscript_tl
721       { \c_math_subscript_token\l_@@_Deltaf_subscript_tl }
722     }
723     {
724       \Delta\l_@@_Deltaf_entity_tl\c_math_subscript_token\formation
725       \tl_if_empty:NF \l_@@_Deltaf_superscript_tl
726       { \c_math_superscript_token\l_@@_Deltaf_superscript_tl }
727     }
728   }
729 }
730 \cs_set:Nn \@@_Deltaf_two:n
731 {
732   \tl_set:Nn \l_@@_Deltaf_subscript_tl {#1}
733   % check for case 4
734   \peek_catcode_remove:NTF \c_math_superscript_token
735   { \@@_Deltaf_one:n }
736   {
737     \bool_if:NTF \l_@@_Deltaf_parentheses_bool
738     {
739       (\Delta\l_@@_Deltaf_entity_tl\c_math_subscript_token\formation
740        \tl_if_empty:NF \l_@@_Deltaf_superscript_tl
741        { \c_math_superscript_token\l_@@_Deltaf_superscript_tl }
742       )
743       \tl_if_empty:NF \l_@@_Deltaf_subscript_tl
744       \c_math_subscript_token\l_@@_Deltaf_subscript_tl
745     }
746     {
747       \Delta\l_@@_Deltaf_entity_tl\c_math_subscript_token\formation
748       \tl_if_empty:NF \l_@@_Deltaf_superscript_tl
749       { \c_math_superscript_token\l_@@_Deltaf_superscript_tl }
750     }
751   }
752 }
753 \RenewDocumentCommand{\Deltaf}{m}
754 {
755   \tl_clear:N \l_@@_Deltaf_superscript_tl
756   \tl_clear:N \l_@@_Deltaf_subscript_tl
757   \tl_set:Nn \l_@@_Deltaf_entity_tl {#1}
758   \bool_set_false:N \l_@@_Deltaf_parentheses_bool

```

```

759 \peek_catcode_remove:NTF \c_math_superscript_token
760 {
761   \@@_Deltaf_one:n
762 }
763 {
764   \peek_catcode_remove:NTF \c_math_subscript_token
765   {
766     \bool_set_true:N \l_@@_Deltaf_parentheses_bool
767     \@@_Deltaf_two:n
768   }
769   {
770     \Delta #1\c_math_subscript_token \formation
771   }
772 }
773 }
774 \RenewDocumentCommand{\kappaT}{}{\compressibilitysymbol}
775 \RenewDocumentCommand{\alphaP}{}{\expansivitysymbol}
776 \RenewSubscriptedSymbol{\cV}{\@@_intensive:n \heatcapacitysymbol}
777   {\@@_intensive:n \g_@@_volume_symbol}
778 \RenewSubscriptedSymbol{\cVs}{\@@_specific:n \heatcapacitysymbol}
779   {\@@_intensive:n \g_@@_volume_symbol}
780 \RenewDocumentCommand{\Lm}{}{\omega}
781 \RenewDocumentCommand{\Ls}{}{\hat\omega}
782 }
783 }
784 \DeclareOption{MSBB}{
785   \ExecuteOptions{EUFGHAn,intensive-lowercase,delta,shortpm}
786   \AtEndOfPackage{
787     \RenewDocumentCommand{\IGM}{}{\ast} % FIXME: is this * or \circ?
788     \RenewDocumentCommand{\IG}{}{\ast}
789     \RenewDocumentCommand{\expansivitysymbol}{}{\beta}
790     \RenewDocumentCommand{\muJT}{}{\mu\c_math_subscript_token J}
791     \RenewDocumentCommand{\allcomponents}{O{} m}{#2}
792     \RenewDocumentCommand{\allbut}{O{j} m m}
793     {
794       \tl_if_eq:nnTF {#1} {#2}
795       { {#3}\c_math_subscript_token k }
796       { {#3}\c_math_subscript_token{#1} }
797     }
798     \tl_gset_eq:NN \g_@@_Helmholtz_symbol \psi
799     \RenewDocumentCommand{\Ft}{}{\Psi}
800     \RenewDocumentCommand{\Fpm}{}{\partialmolar{\Psi}}
801     \cs_set:Nn \@@_intensive:n {\@@_overline:n{\text_lowercase:n{#1}}}
802     \cs_set:Nn \@@_specific:n {\text_lowercase:n{#1}}
803     \RenewDocumentCommand{\fmix}{}{\bar f}
804     \RenewDocumentCommand{\phimix}{}{\bar\phi}
805     \RenewDocumentCommand{\phimix}{}{\bar\phi}
806     \tl_gset:Nn \g_@@_pressure_symbol p
807     \tl_gset:Nn \g_@@_volume_symbol v
808     \RenewDocumentCommand{\partialmolar}{m}
809     {
810       \tl_set:Nn \l_@@_pm_symbol_tl {\text_uppercase:n #1}
811       \@@_generic_pm:
812     }
813     \RenewDocumentCommand{\kappaT}{}{\compressibilitysymbol}
814     \RenewDocumentCommand{\alphaP}{}{\expansivitysymbol}

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```

815 \RenewDocumentCommand{\mustd}{}{\Gm^\std}
816 \RenewExpandableDocumentCommand{\formation}{}{\mathrm{f}}
817 \RenewDocumentCommand{\Pstd}{}
818   {\g_@@_pressure_symbol\c_math_subscript_token\text{ref}}
819 \RenewExpandableDocumentCommand{\xrxn}{}{\varepsilon}
820 \RenewExpandableDocumentCommand{\Lm}{}{\@@_intensive:n \omega}
821 \RenewExpandableDocumentCommand{\Ls}{}{\@@_specific:n \omega}
822 \RenewExpandableDocumentCommand{\heatcapacitysymbol}{}{C}
823 \tl_new:N \l_@@_Deltaf_superscript_tl
824 \tl_new:N \l_@@_Deltaf_subscript_tl
825 \bool_new:N \l_@@_Deltaf_parentheses_bool
826 \tl_new:N \l_@@_Deltaf_entity_tl
827 \cs_set:Nn \@@_Deltaf_one:n
828 {
829   \tl_set:Nn \l_@@_Deltaf_superscript_tl {#1}
830   \peek_catcode_remove:NTF \c_math_subscript_token
831   {
832     \bool_set_true:N \l_@@_Deltaf_parentheses_bool
833     \@@_Deltaf_two:n
834   }
835   {
836     \bool_if:NTF \l_@@_Deltaf_parentheses_bool
837     {
838       (\l_@@_Deltaf_entity_tl\c_math_subscript_token\formation
839        \tl_if_empty:NF \l_@@_Deltaf_superscript_tl
840        { \c_math_superscript_token\l_@@_Deltaf_superscript_tl }
841       )
842       \tl_if_empty:NF \l_@@_Deltaf_subscript_tl
843       { \c_math_subscript_token\l_@@_Deltaf_subscript_tl }
844     }
845     {
846       \l_@@_Deltaf_entity_tl\c_math_subscript_token{\formation}
847       \tl_if_empty:NF \l_@@_Deltaf_superscript_tl
848       { \c_math_superscript_token\l_@@_Deltaf_superscript_tl }
849     }
850   }
851 }
852 \cs_set:Nn \@@_Deltaf_two:n
853 {
854   \tl_set:Nn \l_@@_Deltaf_subscript_tl {#1}
855   % check for case 4
856   \peek_catcode_remove:NTF \c_math_superscript_token
857   { \@@_Deltaf_one:n }
858   {
859     \bool_if:NTF \l_@@_Deltaf_parentheses_bool
860     {
861       (\l_@@_Deltaf_entity_tl\c_math_subscript_token\formation
862        \tl_if_empty:NF \l_@@_Deltaf_superscript_tl
863        { \c_math_superscript_token\l_@@_Deltaf_superscript_tl }
864       )
865       \tl_if_empty:NF \l_@@_Deltaf_subscript_tl
866       { \c_math_subscript_token\l_@@_Deltaf_subscript_tl }
867     }
868     {
869       \l_@@_Deltaf_entity_tl\c_math_subscript_token\formation
870       \tl_if_empty:NF \l_@@_Deltaf_superscript_tl

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```

871         { \c_math_superscript_token\l_@@_Deltaf_superscript_tl }
872     }
873 }
874 }
875 \RenewDocumentCommand{\Deltaf}{m}
876 {
877     \tl_clear:N \l_@@_Deltaf_superscript_tl
878     \tl_clear:N \l_@@_Deltaf_subscript_tl
879     \tl_set:Nn \l_@@_Deltaf_entity_tl {#1}
880     \bool_set_false:N \l_@@_Deltaf_parentheses_bool
881     \peek_catcode_remove:NTF \c_math_superscript_token
882     {
883         \@@_Deltaf_one:n
884     }
885     {
886         \peek_catcode_remove:NTF \c_math_subscript_token
887         {
888             \bool_set_true:N \l_@@_Deltaf_parentheses_bool
889             \@@_Deltaf_two:n
890         }
891         {
892             \Delta #1\c_math_subscript_token \formation
893         }
894     }
895 }
896 }
897 }
898 \DeclareOption{Prausnitz}{
899     \ExecuteOptions{intensive-lowercase,shortpm}
900     \AtEndOfPackage{
901         \RenewExpandableDocumentCommand{\ncomponents}{}{m}
902         \RenewDocumentCommand{\fmix}{}{f}
903         \RenewDocumentCommand{\fsat}{}{\fpure\c_math_superscript_token\sat}
904         % TODO: this should pick up H_2 and make it into H_{2,1} (assuming
905         % the solvent is always 1...?)
906         \RenewDocumentCommand{\Henryrat}{}{H}
907         %^^A\RenewDocumentCommand{\residual}{}{{\mathcal{R}}}}
908         \RenewDocumentCommand{\allcomponents}{0{i} m}
909         { {#2}\c_math_subscript_token{#1} }
910         \RenewDocumentCommand{\allbut}{0{i} m m}
911         { \tl_if_eq:nnTF {#1} {#2}
912             { {#3}\c_math_subscript_token k }
913             { {#3}\c_math_subscript_token{#1} }
914         }
915         \tl_set:Nn \l_@@_sub_separator_tl {\,,}
916         \tl_gset:Nn \g_@@_pressure_symbol {p}
917         \tl_gset:Nn \g_@@_volume_symbol {v}
918         \RenewSubscriptedSymbol{\fpure}{f}{\text{pure}}
919         \RenewDocumentCommand{\phimix}{}{\varphi}
920         \RenewDocumentCommand{\phisat}{}{\varphi^{\sat}}
921         \RenewDocumentCommand{\sat}{}{s}
922         \RenewDocumentCommand{\mixing}{}{\text{mixing}}
923         \RenewDocumentCommand{\Lm}{}{\omega}
924         \RenewDocumentCommand{\Ls}{}{\@@_specific:n \omega}
925         \RenewSubscriptedSymbol{\hipure}{\varphi}{\text{pure}}
926         \RenewDocumentCommand{\IG}{}{\text{id}}

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```

927 \RenewDocumentCommand{\IGM}{}{}{\text{id}}
928 \RenewDocumentCommand{\IS}{}{}{\text{(ideal)}}
929 \RenewExpandableDocumentCommand{\compressibilitysymbol}{}{\beta}
930 \RenewDocumentCommand{\kappaT}{}{\compressibilitysymbol}
931 \RenewDocumentCommand{\alphaP}{}{\expansivitysymbol}
932 \RenewExpandableDocumentCommand{\std}{}{0}
933 \ifpackageloaded{emf}
934   {\RenewDocumentCommand{\Epot}{}{\emf}}
935   {\RenewDocumentCommand{\Epot}{}{\mathcal{E}}}
936 \DeclareMathAlphabet{\mathdutchcal}{U}{dutchcal}{m}{n}
937 \RenewExpandableDocumentCommand{\actrel}{}{\mathdutchcal{a}}
938 }
939 }
940 \DeclareOption{Sandler}{
941   \ExecuteOptions{EUAGHaN,extensive-plain,shortpm}
942   \AtEndOfPackage{
943     \RenewDocumentCommand{\Pvap}{}
944       {\P\c_math_superscript_token{\text{vap}}}
945     \RenewDocumentCommand{\excess}{}{\text{ex}}
946     \RenewDocumentCommand{\residual}{}{\text{r}}
947     \RenewExpandableDocumentCommand{\ncomponents}{}{\mathcal{C}}
948     \RenewDocumentCommand{\fmix}{}{\bar f}
949     \RenewDocumentCommand{\fstf}{}{\bar f\c_math_superscript_token\std}
950     \RenewDocumentCommand{\phimix}{}{\bar\phi}
951     \RenewDocumentCommand{\allcomponents}{0}{m}{\@@_underline:n{#2}}
952     \RenewDocumentCommand{\IG}{}{\text{IG}}
953     \RenewDocumentCommand{\IGM}{}{\text{IGM}}
954     \RenewDocumentCommand{\IS}{}{\text{IM}}
955     \RenewDocumentCommand{\Deltamix}{m}
956       {\Delta\c_math_subscript_token\mixing #1}
957     \RenewDocumentCommand{\Deltarxn}{m}
958       {\Delta\c_math_subscript_token\reaction #1}
959     \RenewDocumentCommand{\Deltasub}{m}
960       {\Delta\c_math_subscript_token\sublimation #1}
961     \RenewDocumentCommand{\Deltafus}{m}
962       {\Delta\c_math_subscript_token\fusion #1}
963     \RenewDocumentCommand{\Deltavap}{m}
964       {\Delta\c_math_subscript_token\vaporization #1}
965     \RenewDocumentCommand{\Pstd}{}{\text{1~bar}}
966     \peek_catcode_remove:NT \c_math_subscript_token {\use_none:n}
967   }
968   \RenewDocumentCommand{\Cstd}{}{\text{1~molal}}
969   \peek_catcode_remove:NT \c_math_subscript_token {\use_none:n}
970 }
971 \RenewDocumentCommand{\Henryrat}{}{H}
972 \RenewSubscriptedSymbol{\cV}
973   {\heatcapacitysymbol}{\g_@@_volume_symbol}
974 \RenewSubscriptedSymbol{\cP}
975   {\heatcapacitysymbol}{\g_@@_pressure_symbol}
976 \RenewSubscriptedSymbol{\cVt}
977   {\Nt\heatcapacitysymbol}{\g_@@_volume_symbol}
978 \RenewSubscriptedSymbol{\cPt}
979   {\Nt\heatcapacitysymbol}{\g_@@_pressure_symbol}
980 \RenewDocumentCommand{\formation}{}{\mathrm{f}}
981 \RenewDocumentCommand{\Deltaf}{}
982   {\Delta\c_math_subscript_token\formation}

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```

983 \RenewDocumentCommand{\alphaP}{}{\expansivitysymbol}
984 \RenewDocumentCommand{\muJT}{}{\mu}
985 \RenewDocumentCommand{\rxn}{}{X}
986 }
987 }
988 \DeclareOption{SVNAS}{
989 \ExecuteOptions{extensive-superscript,shortpm}
990 \AtEndOfPackage{
991 \RenewDocumentCommand{\allcomponents}{0}{m}{#2}
992 \RenewDocumentCommand{\allbut}{0}{j}{m}{m}
993 {
994 \tl_if_eq:nnTF {#1} {#2}
995 { {#3}\c_math_subscript_token k }
996 { {#3}\c_math_subscript_token{#1} }
997 }
998 \RenewDocumentCommand{\IG}{}{{ig}}
999 \RenewDocumentCommand{\IGM}{}{{ig}}
1000 \RenewDocumentCommand{\IS}{}{{id}}
1001 \RenewDocumentCommand{\IS}{}{{id}}
1002 \RenewDocumentCommand{\fusion}{}{{sl}}
1003 \RenewDocumentCommand{\vaporization}{}{{vl}}
1004 \RenewDocumentCommand{\sublimation}{}{{vs}}
1005 \RenewDocumentCommand{\expansivitysymbol}{}{\beta}
1006 \RenewDocumentCommand{\rxn}{}{\varepsilon}
1007 \RenewDocumentCommand{\Deltarxn}{m}{\Delta #1}
1008 \RenewDocumentCommand{\Deltamix}{m}{\Delta #1}
1009 \RenewSubscriptedSymbol{\cVt}
1010 {\Nt\heatcapacitysymbol}{\g_@@_volume_symbol}
1011 \RenewSubscriptedSymbol{\cPt}
1012 {\Nt\heatcapacitysymbol}{\g_@@_pressure_symbol}
1013 % FIXME
1014 \RenewDocumentCommand{\cP}{}
1015 {{\heatcapacitysymbol\c_math_subscript_token\g_@@_pressure_symbol}}
1016 \RenewDocumentCommand{\cV}{}
1017 {{\heatcapacitysymbol\c_math_subscript_token\g_@@_volume_symbol}}
1018 \RenewDocumentCommand{\kappaT}{}{\compressibilitysymbol}
1019 \RenewDocumentCommand{\alphaP}{}{\expansivitysymbol}
1020 \RenewDocumentCommand{\muJT}{}{\mu}
1021 \RenewDocumentCommand{\Cstd}{}{m\c_math_superscript_token\std}
1022 \peek_catcode_remove:NT \c_math_subscript_token {\use_none:n}
1023 }
1024 \RenewDocumentCommand{\mustd}{}{\Gm\c_math_superscript_token\std}
1025 \RenewDocumentCommand{\Henryrat}{}{\mathcal{H}}
1026 \RenewDocumentCommand{\formation}{}
1027 {
1028 \tl_set:Nn \l_@@_sub_separator_tl {}
1029 f298
1030 }
1031 \cs_set:Nn \@@_specific:n {#1}
1032 }
1033 }
1034 \DeclareOption{ModellReid}{\ExecuteOptions{TesterModell}}
1035 \DeclareOption{TesterModell}{
1036 \ExecuteOptions{EUAGHaN,delta,shortpm}
1037 \AtEndOfPackage{
1038 \cs_set:Nn \@@_specific:n {\text_lowercase:n{#1}}

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```

1039 \RenewExpandableDocumentCommand{\ncomponents}{n}
1040 \RenewDocumentCommand{\allcomponents}{0i m}
1041 {
1042   {#2}\c_math_subscript_token{#1}
1043 }
1044 \RenewDocumentCommand{\allbut}{0i m m}
1045 { \tl_if_eq:nnTF {#1} {#2}
1046   { {#3}\c_math_subscript_token{k}[#2] }
1047   { {#3}\c_math_subscript_token{#1}[#2] }
1048 }
1049 \RenewDocumentCommand{\allbutlastand}{0j m m}
1050 { \tl_if_eq:xxTF {#2} {\ncomponents}
1051   { {#3}\c_math_subscript_token{#1}\relax[#2] }
1052   { \tl_if_eq:nnTF {#1} {#2}
1053     { {#3}\c_math_subscript_token{k}[#2,\ncomponents] }
1054     { {#3}\c_math_subscript_token{#1}[#2,\ncomponents] }
1055   }
1056 }
1057 \RenewDocumentCommand{\IG}{o}
1058 \RenewDocumentCommand{\IGM}{o}
1059 \RenewDocumentCommand{\IS}{ID}
1060 \RenewDocumentCommand{\excess}{EX}
1061 \RenewDocumentCommand{\reaction}{rx}
1062 \RenewDocumentCommand{\Henryrat}{}
1063   {f\c_math_superscript_token{\ast\ast}}
1064 \RenewDocumentCommand{\Henrymol}{}{f\c_math_superscript_token\ast}
1065 \cs_if_exist:NTF \gammaup
1066 { \RenewDocumentCommand{\gamma}{}{\gammaup} }
1067 { \cs_if_exist:NT \upgamma
1068   { \RenewDocumentCommand{\gamma}{}{\upgamma} }
1069 }
1070 \RenewDocumentCommand{\gammarat}{}
1071   {\gamma\c_math_superscript_token{\ast\ast}}
1072 \RenewDocumentCommand{\gammamol}{}
1073   {\gamma\c_math_superscript_token\ast}
1074 \RenewExpandableDocumentCommand{\JTsymbol}{}{\alpha}
1075 \RenewSubscriptedSymbol{\muJT}{JTsymbol}{H}
1076 \cs_if_exist:NTF \xiup
1077 { \RenewDocumentCommand{\xrxn}{}{\xiup} }
1078 { \cs_if_exist:NT \upxi
1079   { \RenewDocumentCommand{\xrxn}{}{\upxi} }
1080 }
1081 \cs_if_exist:NTF \phiup
1082 {
1083   \RenewDocumentCommand{\hipure}{}{\phiup}
1084   \RenewDocumentCommand{\phimix}{}{\hat\phiup}
1085   \RenewDocumentCommand{\phisat}{}{\phiup\c_math_superscript_token\sat}
1086 }
1087 { \cs_if_exist:NT \upphi
1088   {
1089     \RenewDocumentCommand{\hipure}{}{\upphi}
1090     \RenewDocumentCommand{\phimix}{}{\hat\upphi}
1091     \RenewDocumentCommand{\phisat}{}{\upphi\c_math_superscript_token\sat}
1092   }
1093 }
1094 \RenewDocumentCommand{\Cstd}{}{m\c_math_superscript_token+

```

```

1095     \peek_catcode_remove:NT \c_math_subscript_token {\use_none:n}
1096   }
1097   \cs_if_exist:NTF \muup
1098   { \RenewDocumentCommand{\mu}{\muup} }
1099   { \cs_if_exist:NT \upmu
1100     { \RenewDocumentCommand{\mu}{\upmu} }
1101   }
1102   \RenewDocumentCommand{\std}{\o}
1103   \RenewDocumentCommand{\Pstd}{\P^{\ast}}
1104   \tl_gset:Nn \g_@@_area_symbol {\text{\large\(\mathit{a}\)}}
1105   \RenewDocumentCommand{\Epot}{\exists}
1106   \tl_set:Nn \l_@@_sub_separator_tl {}
1107   \tl_gset:Nn \g_@@_pressure_symbol {p}
1108   \RenewSubscriptedSymbol{\cVt}{\@@_extensive:n {\heatcapacitysymbol}}{v}
1109   \RenewSubscriptedSymbol{\cV}{\@@_intensive:n {\heatcapacitysymbol}}{v}
1110   \RenewSubscriptedSymbol{\cVs}{\@@_specific:n {\heatcapacitysymbol}}{v}
1111 }
1112 }
1113 \DeclareOption{Thompson}{
1114   \ExecuteOptions{EUAGHAn,delta,shortpm}
1115   \AtEndOfPackage
1116   {
1117     \RenewDocumentCommand{\excess}{\EX}
1118     \RenewDocumentCommand{\residual}{\R}
1119     \RenewDocumentCommand{\actrel}{\widehat{a}}
1120     \RenewSubscriptedSymbol{\Henryrat}{k}{H}
1121     \RenewSubscriptedSymbol{\Henrymol}{k}{H}
1122     \RenewDocumentCommand{\allcomponents}{O{j} m}
1123     {
1124       {#2}\c_math_subscript_token{#1}
1125     }
1126     \RenewDocumentCommand{\allNs}{O{j}}{\allcomponents[#1]{\Nt}}
1127     \RenewDocumentCommand{\allXs}{O{j}}{\allcomponents[#1]{x}}
1128     \RenewDocumentCommand{\allYs}{O{j}}{\allcomponents[#1]{y}}
1129     \RenewDocumentCommand{\allmus}{O{j}}{\allcomponents[#1]{\mu}}
1130     \RenewDocumentCommand{\allMs}{O{j}}{\allcomponents[#1]{m}}
1131     \RenewDocumentCommand{\allWs}{O{j}}{\allcomponents[#1]{w}}
1132     \RenewExpandableDocumentCommand{\ncomponents}{\c}
1133     \RenewDocumentCommand{\IS}{\IS}
1134     \RenewDocumentCommand{\IG}{\IG}
1135     \RenewDocumentCommand{\IGM}{\IG}
1136     \cs_new:Nn \@@_fpure_one:n
1137     {
1138       f\c_math_subscript_token{#1}
1139       \peek_catcode_remove:NF \c_math_superscript_token
1140       { \c_math_superscript_token\bullet }
1141     }
1142     \RenewDocumentCommand{\fpure}{\c}
1143     {
1144       \peek_catcode_remove:NTF \c_math_subscript_token
1145       { \@@_fpure_one:n }
1146       { f }
1147     }
1148     \cs_new:Nn \@@_intensive_two:n
1149     {
1150       \c_math_subscript_token{#1}

```



```

1151 \peek_catcode:NF \c_math_superscript_token
1152 {
1153   \c_math_superscript_token\bullet
1154 }
1155 }
1156 \cs_set:Nn \@@_intensive:n
1157 {
1158   #1
1159   \peek_catcode_remove:NT \c_math_subscript_token
1160   { \@@_intensive_two:n }
1161 }
1162 \cs_new:Nn \@@_phipure_one:n
1163 {
1164   \phi\c_math_subscript_token{#1}
1165   \peek_catcode:NF \c_math_superscript_token
1166   { \c_math_superscript_token\bullet }
1167 }
1168 \RenewDocumentCommand{\phipure}{}
1169 {
1170   \peek_catcode_remove:NTF \c_math_subscript_token
1171   {
1172     \@@_phipure_one:n
1173   }
1174   {\phi}
1175 }
1176 \RenewDocumentCommand{\phimix}{}{\widehat{\phi}}
1177 \RenewDocumentCommand{\fmix}{}{\widehat{f}}
1178 \RenewDocumentCommand{\mixing}{}{MIX}
1179 \RenewDocumentCommand{\muJT}{}{\alpha\c_math_subscript_token H}
1180 \RenewDocumentCommand{\Deltamix}{m}
1181   {\Delta\c_math_subscript_token\mixing #1}
1182 \RenewDocumentCommand{\reaction}{}{R}
1183 \RenewDocumentCommand{\fusion}{}{SL}
1184 \RenewDocumentCommand{\vaporization}{}{LV}
1185 \RenewDocumentCommand{\sublimation}{}{SV}
1186 \RenewDocumentCommand{\allbut}{O{j} m m}
1187 { \tl_if_eq:nnTF {#1} {#2}
1188   {
1189     {#3}\c_math_subscript_token k\neq{#3}\c_math_subscript_token{#2}
1190   }
1191   {
1192     {#3}\c_math_subscript_token{#1}\neq{#3}\c_math_subscript_token{#2}
1193   }
1194 }
1195 \RenewDocumentCommand{\Cstd}{}{m^{\std}}
1196 \RenewDocumentCommand{\mustd}{}{\Gamma}
1197 \RenewDocumentCommand{\formation}{}{F}
1198 %^^A \RenewDocumentCommand{\Deltarxn}{}{\Delta}
1199 }
1200 }

```

We execute the default options below.

```

1201 \ExecuteOptions{EUAGHan,subscripts,parentheses,intensive-plain,
1202   moles-index,longpm}
1203 \ProcessOptions

```

B.5 Variable Order

We next encode a routine to sort non-subscripted variables into a consistent order. It currently does not sort variables with subscripts.

```

1204 \tl_const:Nn \c_@@_sort_order_tl
1205     {\Et\Em\Es\Ut\Um\Us\Ht\Hm\Hs\Ft\Fm\Fs\Gt\Gm\Gs\Lt\Lm\Ls T\St\Sm\Ss
1206      P\Vt\Vm\Vs\mu\Nt mxyz\At\Am\As\sigma
1207      ABCDEFGHIJKLMNOPQRSTUVWXYZabcdefghijklmnopqrstuvwxyz}
1208 \clist_new:N \l_@@_in_list_clist
1209 \clist_new:N \l_@@_sorted_list_clist
1210 \clist_new:N \l_@@_remaining_list_clist
1211 \cs_new:Nn \@@_sort_clist:n
1212 {% Sort the list in the order of \c_@@_sort_order_tl
1213
1214  % Wipe out any remnants from the last sort
1215  \clist_clear:N \l_@@_in_list_clist
1216  \clist_clear:N \l_@@_sorted_list_clist
1217
1218  % Make a copy of the list
1219  \clist_set:Nn \l_@@_remaining_list_clist {#1}
1220
1221  % Make a list of everything that's in the known sort order list
1222  % and put everything else in the "not in sort order list" list.
1223  \tl_map_inline:Nn \c_@@_sort_order_tl
1224  {
1225    \clist_if_in:NnT \l_@@_remaining_list_clist {##1}
1226    { \clist_put_right:Nn \l_@@_in_list_clist {##1} }
1227
1228    \clist_remove_all:Nn \l_@@_remaining_list_clist {##1}
1229  }
1230
1231  % Then merge the lists back together again.
1232  \clist_if_empty:NF \l_@@_in_list_clist
1233  {
1234    \clist_put_right:Nn \l_@@_sorted_list_clist \l_@@_in_list_clist
1235  }
1236  \clist_if_empty:NF \l_@@_remaining_list_clist
1237  {
1238    \clist_put_right:Nn \l_@@_sorted_list_clist \l_@@_remaining_list_clist
1239  }
1240  \clist_use:Nn \l_@@_sorted_list_clist ,
1241 }
```

B.6 Commands for Partial Derivatives

The `\Partial` command and its second-order siblings are defined as below. They typeset partial derivatives of the first argument with respect to the second (and third, in the case of mixed second partial derivatives) arguments, holding the last argument constant.

The starred forms adjust the spacing after the partial derivative so the trailing binary operator (assumed to be the same width as an equals sign) overhangs the variables held constant. We thus set `operator_width` to be *just* greater than the width of an equals sign.

```

1242 \dim_new:N \l_@@_Partial_const_dim
1243 \dim_new:N \l_@@_operator_width_dim
```

```

1244 \dim_new:N \l_@@_adjust_width_dim
1245 \settowidth{\l_@@_operator_width_dim}{=}
1246 \dim_set:Nn \l_@@_adjust_width_dim {0.1\l_@@_operator_width_dim}
1247 \dim_add:Nn \l_@@_operator_width_dim \l_@@_adjust_width_dim

```

`\Partial` The command `\Partial` and its friends drastically simplify the creation of partial derivatives. The command `\Partial*` is the same as `\Partial` except that it adjusts the spacing so the (presumably) binary operator that follows it slightly overlaps the subscripts.

```

1248 \tl_new:N \l_@@_Partial_start_tl
1249 \tl_new:N \l_@@_Partial_end_tl
1250 \tl_new:N \l_@@_Partial_empty_end_tl
1251 \tl_new:N \l_@@_Partial_middle_tl
1252 \tl_set:Nn \l_@@_Partial_start_tl {\left\l_@@_PartialOpen_tl}
1253 \tl_set:Nn \l_@@_Partial_end_tl {\right\l_@@_PartialClose_tl}
1254 \tl_set:Nn \l_@@_Partial_empty_end_tl {\right\l_@@_PartialEmptyClose_tl}
1255 \tl_set:Nn \l_@@_Partial_middle_tl {\middle}
1256 \cs_set_eq:NN \@@_frac:nn \frac
1257 \NewDocumentCommand{\Partial}{s m m m}
1258 { \bool_if:nTF {#1}
1259   {% Starred form (recursive)
1260     \settowidth{\l_@@_Partial_const_dim}{\ensuremath{#4}}%
1261     \dim_add:Nn \l_@@_Partial_const_dim {-0.20\l_@@_Partial_const_dim}%
1262     \Partial{#2}{#3}{#4}%
1263     \bool_if:NT \l_@@_subscripted_bool
1264     { \dim_compare:nNnTF \l_@@_operator_width_dim
1265       < \l_@@_Partial_const_dim
1266       { \kern -\l_@@_operator_width_dim }
1267       { \kern -\l_@@_Partial_const_dim }
1268     }
1269   }
1270   {% Unstarred form
1271     \bool_if:NtF \l_@@_subscripted_bool
1272     {% Handle case of empty variables held constant
1273       \tl_if_eq:nnTF {#4} {}
1274       { \l_@@_Partial_start_tl
1275         \@@_frac:nn{\partial #2}{\partial #3}\l_@@_Partial_empty_end_tl
1276       }
1277       { \l_@@_Partial_start_tl\@@_frac:nn{\partial #2}
1278         {\partial #3}\l_@@_Partial_end_tl
1279         \c_math_subscript_token{#4}%
1280       }
1281     }
1282     {% Check whether #4 contains \allNsbut{i} and #3 is \Nt_i
1283       \tl_if_in:nnTF {#3} {\Nt}
1284       { \RenewDocumentCommand{\allbut}{O{j} m m}{\allcomponents{##3}}
1285         \l_@@_Partial_start_tl
1286         \@@_frac:nn{\partial #2(\@@_sort_clist:n{#4})}
1287         {\partial #3}\l_@@_Partial_end_tl
1288       }
1289       { \l_@@_Partial_start_tl
1290         \@@_frac:nn{\partial #2(\@@_sort_clist:n{#3,#4})}
1291         {\partial #3}\l_@@_Partial_end_tl
1292       }
1293     }
1294   }

```

1295 }

`\PartialBigg` The `\PartialBigg` macro (and its starred form) replace the `\left` and `\right` commands in `\Partial` with `amsmath`'s `\Biggl` and `\Biggr` variants. The starred form is inherited from `\Partial` without modification.

```
1296 \NewDocumentCommand{\PartialBigg}{}
1297 { \tl_set:Nn \l_@@_Partial_start_tl {\Biggl\l_@@_PartialOpen_tl}
1298   \tl_set:Nn \l_@@_Partial_end_tl {\Biggr\l_@@_PartialClose_tl}
1299   \tl_set:Nn \l_@@_Partial_Empty_end_tl
1300     {\Biggr\l_@@_PartialEmptyClose_tl}
1301   \Partial
1302 }
```

`\Partialbigg` The `\Partialbigg` macro does the same thing as `\PartialBigg`, except using `amsmath`'s `\biggl`/`\biggr` variants.

```
1303 \NewDocumentCommand{\Partialbigg}{}
1304 { \tl_set:Nn \l_@@_Partial_start_tl {\biggl\l_@@_PartialOpen_tl}
1305   \tl_set:Nn \l_@@_Partial_end_tl {\biggr\l_@@_PartialClose_tl}
1306   \tl_set:Nn \l_@@_Partial_empty_end_tl
1307     {\biggr\l_@@_PartialEmptyClose_tl}
1308   \Partial
1309 }
```

`\PartialSecond` The second partial derivatives are defined similarly to `\Partial`.

```
1310 \NewDocumentCommand{\PartialSecond}{s m m m}
1311 {
1312   \bool_if:nTF {#1}
1313   {% Starred form
1314     \settowidth{\l_@@_Partial_const_dim}{\ensuremath{#4}}%
1315     \dim_add:Nn \l_@@_Partial_const_dim {-0.20\l_@@_Partial_const_dim}
1316     \PartialSecond{#2}{#3}{#4}%
1317     \bool_if:nT \l_@@_subscripted_bool
1318     { \dim_compare:nNnTF {\l_@@_operator_width_dim}
1319       < {\l_@@_Partial_const_dim}
1320       { \kern -\l_@@_operator_width_dim }
1321       { \kern -\l_@@_Partial_const_dim }
1322     }
1323   }
1324   {% Unstarred form
1325     \bool_if:NTF \l_@@_subscripted_bool
1326     {% Handles case of empty variables held constant
1327       \tl_if_eq:nnTF {#4} {}
1328       { \l_@@_Partial_start_tl
1329         \@@_frac:nn{\partial\c_math_superscript_token 2 #2}
1330         {\partial #3\c_math_superscript_token 2}\l_@@_Partial_empty_end_tl
1331       }
1332       { \l_@@_Partial_start_tl
1333         \@@_frac:nn{\partial\c_math_superscript_token 2 #2}
1334         {\partial #3\c_math_superscript_token 2}\l_@@_Partial_end_tl
1335         \c_math_subscript_token{#4}%
1336       }
1337     }
1338     {% Check whether #4 contains \allNsbut{i} and #3 is \Nt_i
1339     \tl_if_in:nnTF {#2} {\Nt}
1340     { \RenewDocumentCommand{\allbut}{O{j} m m}{\allcomponents{##3}}
1341       \l_@@_Partial_start_tl
```

```

1342      \@@_frac:nn{\partial\c_math_superscript_token 2 #2(#4)}
1343      {\partial\c_math_superscript_token 2 #3}\l_@@_Partial_end_tl
1344    }
1345    { \l_@@_Partial_start_tl
1346      \@@_frac:nn{\partial\c_math_superscript_token 2
1347        #2(\@@_sort_clist:n{#3,#4})}
1348      {\partial #3\c_math_superscript_token 2}\l_@@_Partial_end_tl
1349    }
1350  }
1351 }
1352 }

```

`\PartialSecondBigg` The `\PartialSecondBigg` macro and its starred variant replace `\left` and `\right` with `amsmath`'s `\Biggl` and `\Biggr`.

```

1353 \NewDocumentCommand{\PartialSecondBigg}{}
1354 { \tl_set:Nn \l_@@_Partial_start_tl {\biggl\l_@@_PartialOpen_tl}
1355   \tl_set:Nn \l_@@_Partial_end_tl {\biggl\l_@@_PartialClose_tl}
1356   \tl_set:Nn \l_@@_Partial_empty_end_tl {\biggl\l_@@_PartialClose_tl}
1357   \PartialSecond
1358 }

```

`\PartialSecondbigg` The `\PartialSecondbigg` macro and its starred variant replace `\left` and `\right` with `amsmath`'s `\biggl` and `\biggr`.

```

1359 \NewDocumentCommand{\PartialSecondbigg}{}
1360 { \tl_set:Nn \l_@@_Partial_start_tl {\biggl\l_@@_PartialOpen_tl}
1361   \tl_set:Nn \l_@@_Partial_end_tl {\biggl\l_@@_PartialClose_tl}
1362   \tl_set:Nn \l_@@_Partial_empty_end_tl {\biggl\l_@@_PartialClose_tl}
1363   \PartialSecond
1364 }

```

`\PartialMixSecond` The macro `\PartialMixSecond` takes an extra argument, but is otherwise the same as its same-variable cousin.

```

1365 \bool_new:N \l_@@_has_x_or_y_bool
1366 \NewDocumentCommand{\PartialMixSecond}{s m m m m}
1367 {
1368   \bool_if:nTF {#1}
1369   {% Starred version
1370     \settowidth{\l_@@_Partial_const_dim}{\ensuremath{#4}}%
1371     \dim_add:Nn \l_@@_Partial_const_dim {-0.20\l_@@_Partial_const_dim}
1372     \PartialMixSecond{#2}{#3}{#4}{#5}
1373     \bool_if:nT \l_@@_subscripted_bool
1374     { \dim_compare:nNnTF {\l_@@_operator_width_dim}
1375       < {\l_@@_Partial_const_dim}
1376       { \kern -\l_@@_operator_width_dim }
1377       { \kern -\l_@@_Partial_const_dim }
1378     }
1379   }
1380   {% Unstarred version
1381     \bool_if:nTF \l_@@_subscripted_bool
1382     {% subscripted version
1383       \tl_if_eq:nnTF {#5} {}
1384       {% Handle case of empty variables held constant
1385         \l_@@_Partial_start_tl
1386         \@@_frac:nn{\partial\c_math_superscript_token 2 #2}
1387         {\partial #3\partial #4}\l_@@_Partial_empty_end_tl
1388       }

```

```

1389 { \l_@@_Partial_start_tl
1390   \@@_frac:nn{\partial\c_math_superscript_token 2 #2}
1391     {\partial #3\partial #4}\l_@@_Partial_end_tl
1392     \c_math_subscript_token{#5}
1393   }
1394 }
1395 {% not subscripted
1396   \tl_if_eq:nnTF {#5} {}
1397   {% empty argument
1398     \l_@@_Partial_start_tl
1399     \@@_frac:nn{\partial\c_math_superscript_token 2
1400       #2(\@@_sort_clist:n{#3,#4,#5})}
1401     {\partial #3\partial #4}\l_@@_Partial_empty_end_tl
1402   }
1403   {% Check whether #3 OR #4 are \Nt_i/etc.
1404     \tl_if_in:nnTF {#3} {\Nt}
1405     { \RenewDocumentCommand{\allbut}{0{j} m m}{\allcomponents{##3}}%
1406       \l_@@_Partial_start_tl
1407       \@@_frac:nn{\partial\c_math_superscript_token 2
1408         #2(\@@_sort_clist:n{#4,#5})}
1409       {\partial #3\partial #4}\l_@@_Partial_end_tl
1410     }
1411     { \tl_if_in:nnTF {#4} {\Nt}
1412       { \RenewDocumentCommand{\allbut}{0{j} m m}{\allcomponents{##3}}%
1413         \l_@@_Partial_start_tl
1414         \@@_frac:nn{\partial\c_math_superscript_token 2
1415           #2(\@@_sort_clist:n{#3,#5})}
1416         {\partial #3\partial #4}\l_@@_Partial_end_tl
1417       }
1418       {% Check for x, y, or w
1419         \bool_set_false:N \l_@@_has_x_or_y_bool
1420         \tl_if_in:nnT {#3} {x}
1421         { \l_@@_has_x_or_y_bool }
1422         \tl_if_in:nnT {#3} {y}
1423         { \l_@@_has_x_or_y_bool }
1424         \tl_if_in:nnT {#3} {w}
1425         { \l_@@_has_x_or_y_bool }
1426         \bool_if:NTF \l_@@_has_x_or_y_bool
1427         { \RenewDocumentCommand{\allbutlastand}{0{j} m m}
1428           {\allcomponents{##3}}
1429           \l_@@_Partial_start_tl
1430           \@@_frac:nn{\partial\c_math_superscript_token 2
1431             #2(\@@_sort_clist:n{#4,#5})}
1432           {\partial #3\partial #4}\l_@@_Partial_end_tl
1433         }
1434         {
1435           \l_@@_Partial_start_tl
1436           \@@_frac:nn{\partial\c_math_superscript_token 2
1437             #2(\@@_sort_clist:n{#3,#4,#5})}
1438           {\partial #3\partial #4}\l_@@_Partial_end_tl
1439         }
1440       }
1441     }
1442   }
1443 }
1444 }

```

1445 }

`\PartialMixSecondBigg` The macro `\PartialMixSecondBigg` is analogous to the aforementioned macros `\PartialMixSecondbigg` `\PartialBigg` and `\PartialSecondBigg`. `\PartialMixSecondbigg` is analogous to `\Partialbigg` and `\PartialSecondbigg`.

```
1446 \NewDocumentCommand{\PartialMixSecondBigg}{}
1447 { \tl_set:Nn \l_@@_Partial_start_tl {\Biggl\l_@@_PartialOpen_tl}
1448   \tl_set:Nn \l_@@_Partial_end_tl {\Biggl\l_@@_PartialClose_tl}
1449   \tl_set:Nn \l_@@_Partial_empty_end_tl {\Biggl\l_@@_PartialClose_tl}
1450   \PartialMixSecond
1451 }
1452 \NewDocumentCommand{\PartialMixSecondbigg}{}
1453 { \tl_set:Nn \l_@@_Partial_start_tl {\biggl\l_@@_PartialOpen_tl}
1454   \tl_set:Nn \l_@@_Partial_end_tl {\biggl\l_@@_PartialClose_tl}
1455   \tl_set:Nn \l_@@_Partial_empty_end_tl {\biggl\l_@@_PartialClose_tl}
1456   \PartialMixSecond
1457 }
```

If the user does not load the `amsmath` package, we will not have access to `\Biggl`, `\Biggr`, `\biggl`, and `\biggr`, so we revert them back to the ordinary `\left` and `\right` versions and warn the user.

```
1458 \AtBeginDocument{%
1459   \ifpackageloaded{amsmath}{}{}%
1460     \PackageWarningNoLine{thermodynamics}
1461       {Package~amsmath~not~loaded;~load~to~make~PartialBigg~and~friends~
1462         work~correctly}%
1463     \cs_set_eq:NN \PartialBigg \Partial
1464     \cs_set_eq:NN \Partialbigg \Partial
1465     \cs_set_eq:NN \PartialSecondBigg \PartialSecond
1466     \cs_set_eq:NN \PartialSecondbigg \PartialSecond
1467     \cs_set_eq:NN \PartialMixSecondBigg \PartialMixSecond
1468     \cs_set_eq:NN \PartialMixSecondbigg \PartialMixSecond
1469     \ProvideDocumentCommand{\rvert}{}{}{}
1470     \ProvideDocumentCommand{\lvert}{}{}{}
1471   }%
1472 }
```

`\Partialinline` Inline partial derivatives.

```
\PartialSecondinline 1473 \NewDocumentCommand{\Partialinline}{}
\PartialMixSecondinline 1474 {
1475   \cs_set:Nn \@@_frac:nn { ##1 \l_@@_Partial_middle_tl / ##2 }
1476   \Partial
1477 }
1478 \NewDocumentCommand{\PartialSecondinline}{}
1479 {
1480   \cs_set:Nn \@@_frac:nn { ##1 \l_@@_Partial_middle_tl / ##2 }
1481   \PartialSecond
1482 }
1483 \NewDocumentCommand{\PartialMixSecondinline}{}
1484 {
1485   \cs_set:Nn \@@_frac:nn { ##1 \l_@@_Partial_middle_tl / ##2 }
1486   \PartialMixSecond
1487 }
```

`\Partialinlinetext` Text-only (non-extensible delimiter) versions of `\Partialinline` and friends.

`\PartialSecondinlinetext`
`\PartialMixSecondinlinetext`

```

1488 \NewDocumentCommand{\Partialinlinetext}{}
1489 { \cs_set_eq:NN \l_@@_Partial_start_tl \l_@@_PartialOpen_tl
1490   \cs_set_eq:NN \l_@@_Partial_end_tl \l_@@_PartialClose_tl
1491   \cs_set_eq:NN \l_@@_Partial_empty_end_tl \l_@@_PartialEmptyClose_tl
1492   \cs_set_eq:NN \l_@@_Partial_middle_tl \relax
1493   \Partialinline
1494 }
1495 \NewDocumentCommand{\PartialSecondinlinetext}{}
1496 { \cs_set_eq:NN \l_@@_Partial_start_tl \l_@@_PartialOpen_tl
1497   \cs_set_eq:NN \l_@@_Partial_end_tl \l_@@_PartialClose_tl
1498   \cs_set_eq:NN \l_@@_Partial_empty_end_tl \l_@@_PartialEmptyClose_tl
1499   \cs_set_eq:NN \l_@@_Partial_middle_tl \relax
1500   \PartialSecondinline
1501 }
1502 \NewDocumentCommand{\PartialMixSecondinlinetext}{}
1503 { \cs_set_eq:NN \l_@@_Partial_start_tl \l_@@_PartialOpen_tl
1504   \cs_set_eq:NN \l_@@_Partial_end_tl \l_@@_PartialClose_tl
1505   \cs_set_eq:NN \l_@@_Partial_empty_end_tl \l_@@_PartialEmptyClose_tl
1506   \cs_set_eq:NN \l_@@_Partial_middle_tl \relax
1507   \PartialMixSecondinline
1508 }

```

B.7 Local Override of Delimiters

We define several environments that *locally* override the delimiters on partial derivatives generated with `\Partial` and friends, the subscript notation for partial derivatives, and/or the definitions of range-oriented macros such as `\allNs`. These environments can be nested; the inner-most one will be honored if conflicts occur.

thermoparentheses (*env.*) Inside this environment, partial derivatives will have parentheses around them, regardless of package options.

```

1509 \NewDocumentEnvironment{thermoparentheses}{}
1510 { \cs_set:Nn \l_@@_PartialOpen_tl {(}
1511   \cs_set:Nn \l_@@_PartialClose_tl {)}
1512   \cs_set:Nn \l_@@_PartialEmptyClose_tl {)}
1513 }{}

```

thermobrackets (*env.*) Inside this environment, partial derivatives will have brackets around them, regardless of package options.

```

1514 \NewDocumentEnvironment{thermobrackets}{}
1515 { \tl_set:Nn \l_@@_PartialOpen_tl {[}
1516   \tl_set:Nn \l_@@_PartialClose_tl {]}
1517   \tl_set:Nn \l_@@_PartialEmptyClose_tl {]}
1518 }{}

```

thermobraces (*env.*) Inside this environment, partial derivatives will have braces around them, regardless of package options.

```

1519 \NewDocumentEnvironment{thermobraces}{}
1520 { \tl_set:Nn \l_@@_PartialOpen_tl {\{
1521   \tl_set:Nn \l_@@_PartialClose_tl {\}}
1522   \tl_set:Nn \l_@@_PartialEmptyClose_tl {\}}
1523 }{}

```

thermobar (*env.*) Inside this environment, partial derivatives will have a trailing vertical bar, regardless of package options.


```

1524 \NewDocumentEnvironment{thermobar}{}
1525 { \tl_set:Nn \l_@@_PartialOpen_tl {.}
1526   \tl_set:Nn \l_@@_PartialClose_tl {\rvert}
1527   \tl_set:Nn \l_@@_PartialEmptyClose_tl {.}
1528 }{}

```

thermoplain (*env.*) Inside this environment, partial derivatives will have no decorations around them, regardless of package options.

```

1529 \NewDocumentEnvironment{thermoplain}{}
1530 { \tl_set:Nn \l_@@_PartialOpen_tl {.}
1531   \tl_set:Nn \l_@@_PartialClose_tl {.}
1532   \tl_set:Nn \l_@@_PartialEmptyClose_tl {.}
1533   \bool_set_false:N \l_@@_subscripted_bool
1534 }{}

```

thermoNOsubscripts (*env.*) Inside this environment, subscripts will not be displayed to the right of partial derivatives, regardless of package options.

```

1535 \NewDocumentEnvironment{thermoNOsubscripts}{}
1536   {\bool_set_false:N \l_@@_subscripted_bool}
1537   {}

```

thermosubscripts (*env.*) Inside this environment, subscripts will be displayed to the right of partial derivatives, regardless of package options.

```

1538 \NewDocumentEnvironment{thermosubscripts}{}
1539   {\bool_set_true:N \l_@@_subscripted_bool}
1540   {}

```

thermomolesrange (*env.*) Inside this environment, the macro `\allNs` will expand to n_1, \dots, n_C (or equivalent symbols if `\Nt` and/or `\ncomponents` have been redefined), regardless of package options. Similar expansions will result for `\allXs`, `\allYs`, `\allMs`, and so on.

```

1541 \NewDocumentEnvironment{thermomolesrange}{}
1542   { \@@_set_moles_range }
1543   {}

```

thermointensiveplain (*env.*)

```

1544 % \begin{macrocode}
1545 \NewDocumentEnvironment{thermointensiveplain}{}
1546   { \@@_set_intensive_plain }
1547   {}

```

thermoextensiveplain (*env.*)

```

1548 \NewDocumentEnvironment{thermoextensiveplain}{}
1549   { \@@_set_extensive_plain }
1550   {}

```

thermointensivelowercase (*env.*)

```

1551 \NewDocumentEnvironment{thermointensivelowercase}{}
1552   {% {
1553 % \RenewExpandableDocumentCommand{\MacroFont}}{}{
1554 % \fontencoding\encodingdefault
1555 % \fontfamily\ttdefault
1556 % \fontseries\mddefault
1557 % \fontshape\shapedefault
1558 % \footnotesize}

```

```

1559
1560 \@@_set_intensive_lowercase
1561 \@@_set_lowercase_pms
1562 }
1563 {}

```

thermoextensivesuperscript (*env.*)

```

1564 \NewDocumentEnvironment{thermoextensivesuperscript}{}
1565 {
1566 \@@_set_extensive_superscripts
1567 }
1568 {}

```

B.8 User-Interface Macros to Define Symbols

`\NewSubscriptedSymbol` First, we define a command that serves to create “subscripted” symbols; for example, typing `\cP_i` should yield C_{P_i} rather than C_{P_i} , C_{P_i} , or C_{P_i} . Superscripts are also handled properly and can be in either order.

```

1569 \str_new:N \l_@@_tmp_str
1570 \cs_new:Npn \@@_check_definable:nN #1#2
1571 {
1572 \tl_trim_spaces_apply:nN {#1} \tl_if_single_token:nTF
1573 {
1574 \str_set:Nx \l_@@_tmp_str {\tl_to_str:n {#1}}
1575 \int_compare:nNnT {\str_count:N \l_@@_tmp_str} = 1
1576 { \PackageError{thermodynamics}
1577 {First~argument~of~'\tl_trim_spaces:o {\tl_to_str:n {#2}}}'~
1578 must~be~a~command}
1579 {The~first~argument~of~'\tl_trim_spaces:o {\tl_to_str:n {#2}}}'~
1580 should~be~the~macro~that~will~be~used~to~refer~to~the~symbol.~
1581 The~provided~argument~'\tl_trim_spaces:o {\tl_to_str:n {#1}}}'~
1582 is~a~single~character.
1583 \MessageBreak Perhaps~a~backslash~is~missing?}
1584 }
1585 }
1586 { \PackageError{thermodynamics}
1587 {First~argument~of~'\tl_trim_spaces:o {\tl_to_str:n {#2}}}'~
1588 must~be~a~command}
1589 {The~first~argument~of~'\tl_trim_spaces:o {\tl_to_str:n {#2}}}'~
1590 should~be~the~macro~that~will~be~used~to~refer~to~the~symbol.~
1591 The~provided~argument~'\tl_trim_spaces:o {\tl_to_str:n {#1}}}'~
1592 contains~more~than~one~token.
1593 \MessageBreak Perhaps~a~backslash~is~missing?}
1594 }
1595 }
1596 \tl_new:N \l_@@_super_separator_tl
1597 \tl_new:N \l_@@_sub_separator_tl
1598 \tl_set:Nn \l_@@_super_separator_tl {,}
1599 \tl_set:Nn \l_@@_sub_separator_tl {,}
1600 \NewDocumentCommand{\NewSubscriptedSymbol}{m m m}
1601 {
1602 \@@_check_definable:nN {#1} \NewSubscriptedSymbol
1603 \cs_if_exist:NT #1
1604 { \PackageError{thermodynamics}
1605 {Command~'\tl_trim_spaces:o {\tl_to_str:n {#1}}}'~already~defined}

```

```

1606     {You~have~used~
1607     '\tl_trim_spaces:o {\tl_to_str:n {\NewSubscriptedSymbol}}'~
1608     with~a~command~that~already~has~a~definition}
1609 }
1610 \cs_new:cpn {\cs_to_str:N #1_one:n} ##1
1611 {
1612     {#2}\c_math_superscript_token{##1}
1613     \peek_catcode_remove:NTF \c_math_subscript_token
1614     { \use:c {\cs_to_str:N #1_three:n} }
1615     { \c_math_subscript_token{#3} }
1616 }
1617
1618 \cs_new:cpn {\cs_to_str:N #1_two:n} ##1
1619 { {#2}\c_math_subscript_token{#3\l_@@_sub_separator_tl ##1} }
1620
1621 \cs_new:cpn {\cs_to_str:N #1_three:n} ##1
1622 { \c_math_subscript_token{#3\l_@@_sub_separator_tl ##1} }
1623
1624 \NewDocumentCommand{#1}{}
1625 {% @branch
1626     \peek_catcode_remove:NTF \c_math_superscript_token
1627     { \use:c {\cs_to_str:N #1_one:n} }
1628     { \peek_catcode_remove:NTF \c_math_subscript_token
1629       { \use:c {\cs_to_str:N #1_two:n} }
1630       { {#2}\c_math_subscript_token{#3} }
1631     }
1632 }
1633 }
1634 \NewDocumentCommand{\RenewSubscriptedSymbol}{m m m}
1635 {
1636     \@@_check_definable:nN {#1} \RenewSubscriptedSymbol
1637     \cs_if_exist:NF #1
1638     { \PackageError{thermodynamics}
1639       {Command~'\tl_trim_spaces:o {\tl_to_str:n {#1}}'~not~defined}
1640       {You~have~used~
1641         '\tl_trim_spaces:o {\tl_to_str:n {\RenewSubscriptedSymbol}}'~
1642         with~a~command~that~does~not~have~a~definition}
1643     }
1644     \cs_set:cpn {\cs_to_str:N #1_one:n} ##1
1645     {
1646         {#2}\c_math_superscript_token{##1}
1647         \peek_catcode_remove:NTF \c_math_subscript_token
1648         { \use:c {\cs_to_str:N #1_three:n} }
1649         { \c_math_subscript_token{#3} }
1650     }
1651
1652     \cs_set:cpn {\cs_to_str:N #1_two:n} ##1
1653     { {#2}\c_math_subscript_token{#3\l_@@_sub_separator_tl ##1} }
1654
1655     \cs_set:cpn {\cs_to_str:N #1_three:n} ##1
1656     { \c_math_subscript_token{#3\l_@@_sub_separator_tl ##1} }
1657
1658     \RenewDocumentCommand{#1}{}
1659     {% @branch
1660         \peek_catcode_remove:NTF \c_math_superscript_token
1661         { \use:c {\cs_to_str:N #1_one:n} }

```

```

1662 { \peek_catcode_remove:NTF \c_math_subscript_token
1663 { \use:c {\cs_to_str:N #1_two:n} }
1664 { {#2}\c_math_subscript_token{#3} }
1665 }
1666 }
1667 }

```

\NewSuperscriptedSymbol

```

1668 \NewDocumentCommand{\NewSuperscriptedSymbol}{m m m}
1669 {
1670 \@@_check_definable:nN {#1} \NewSuperscriptedSymbol
1671 \cs_if_exist:NT #1
1672 { \PackageError{thermodynamics}
1673 {Command~'\tl_trim_spaces:o {\tl_to_str:n {#1}}'~already~defined}
1674 {You~have~used~
1675 '\tl_trim_spaces:o {\tl_to_str:n {\NewSuperscriptedSymbol}}'~
1676 with~a~command~that~already~has~a~definition}
1677 }
1678 \cs_new:cpn {\cs_to_str:N #1_one:n} ##1
1679 {
1680 {#2}\c_math_subscript_token{##1}
1681 \peek_catcode_remove:NTF \c_math_superscript_token
1682 { \use:c {\cs_to_str:N #1_three:n} }
1683 { \c_math_superscript_token{#3} }
1684 }
1685
1686 \cs_new:cpn {\cs_to_str:N #1_two:n} ##1
1687 { {#2}\c_math_superscript_token{#3\l_@@_super_separator_tl ##1} }
1688
1689 \cs_new:cpn {\cs_to_str:N #1_three:n} ##1
1690 { {#2}\c_math_superscript_token{#3\l_@@_super_separator_tl ##1} }
1691
1692 \NewDocumentCommand{#1}{}
1693 {
1694 \peek_catcode_remove:NTF \c_math_subscript_token
1695 { \use:c {\cs_to_str:N #1_one:n} }
1696 { \peek_catcode_remove:NTF \c_math_superscript_token
1697 { \use:c {\cs_to_str:N #1_two:n} }
1698 { {#2}\c_math_superscript_token{#3} }
1699 }
1700 }
1701 }
1702 \NewDocumentCommand{\RenewSuperscriptedSymbol}{m m m}
1703 {
1704 \@@_check_definable:nN {#1} \RenewSuperscriptedSymbol
1705 \cs_if_exist:NF #1
1706 { \PackageError{thermodynamics}
1707 {Command~'\tl_trim_spaces:o {\tl_to_str:n {#1}}'~not~defined}
1708 {You~have~used~
1709 '\tl_trim_spaces:o {\tl_to_str:n {\RenewSuperscriptedSymbol}}'~
1710 with~a~command~that~does~not~have~a~definition}
1711 }
1712 \cs_set:cpn {\cs_to_str:N #1_one:n} ##1
1713 {
1714 {#2}\c_math_subscript_token{##1}
1715 \peek_catcode_remove:NTF \c_math_superscript_token

```

```

1716 { \use:c {\cs_to_str:N #1_three:n} }
1717 { \c_math_superscript_token{#3} }
1718 }
1719
1720 \cs_set:cpn {\cs_to_str:N #1_two:n} ##1
1721 { {#2}\c_math_superscript_token{#3\l_@@_super_separator_tl ##1} }
1722
1723 \cs_set:cpn {\cs_to_str:N #1_three:n} ##1
1724 { \c_math_superscript_token{#3\l_@@_super_separator_tl ##1} }
1725
1726 \RenewDocumentCommand{#1}{}
1727 {% @branch
1728   \peek_catcode_remove:NTF \c_math_subscript_token
1729   { \use:c {\cs_to_str:N #1_one:n} }
1730   { \peek_catcode_remove:NTF \c_math_superscript_token
1731     { \use:c {\cs_to_str:N #1_two:n} }
1732     { {#2}\c_math_superscript_token{#3} }
1733   }
1734 }
1735 }
1736 \cs_new:Nn \@@_subscripted_and_superscripted_core:nnnn
1737 {
1738   \cs_set:cpn {\cs_to_str:N #1_one:n} ##1
1739   {
1740     {#2}\c_math_superscript_token{##1}
1741     \peek_catcode_remove:NTF \c_math_subscript_token
1742     { \use:c {\cs_to_str:N #1_three:n} }
1743     { \c_math_subscript_token{#3} }
1744   }
1745   \cs_set:cpn {\cs_to_str:N #1_two:n} ##1
1746   {
1747     {#2}\c_math_subscript_token{#3 ##1}
1748     \peek_catcode_remove:NTF \c_math_superscript_token
1749     { \use:c {\cs_to_str:N #1_four:n} }
1750     { \c_math_superscript_token{#4 \bullet} }
1751   }
1752   \cs_set:cpn {\cs_to_str:N #1_three:n} ##1
1753   { \c_math_subscript_token{#3 ##1} }
1754   \cs_set:cpn {\cs_to_str:N #1_four:n} ##1
1755   { \c_math_superscript_token{#4 ##1} }
1756 }
1757 \NewDocumentCommand{\NewSubscriptedandSuperscriptedSymbol}{m m m m}
1758 {
1759   \@@_check_definable:nN {#1} \NewSubscriptedandSuperscriptedSymbol
1760   \cs_if_exist:NT #1
1761   { \PackageError{thermodynamics}
1762     {Command~'\tl_trim_spaces:o {\tl_to_str:n {#1}}'~already~defined}
1763     {You~have~used~
1764       '\tl_trim_spaces:o
1765         {\tl_to_str:n {\NewSubscriptedandSuperScriptedSymbol}}'~
1766         with~a~command~that~already~has~a~definition}
1767   }
1768   \@@_subscripted_and_superscripted_core:nnnn {#1} {#2} {#3} {#4}
1769   \NewDocumentCommand{#1}{}
1770   {
1771     \peek_catcode_remove:NTF \c_math_superscript_token

```

```

1772 { \use:c {\cs_to_str:N #1_one:n} }
1773 { \peek_catcode_remove:NTF \c_math_subscript_token
1774 { \use:c {\cs_to_str:N #1_two:n} }
1775 { {#2}\c_math_subscript_token{#3}\c_math_superscript_token{#4} }
1776 }
1777 }
1778 }
1779 \NewDocumentCommand{\RenewSubscriptedandSuperscriptedSymbol}{m m m m}
1780 {
1781 \@@_check_definable:nN {#1} \RenewSubscriptedandSuperscriptedSymbol
1782 %^^A \cs_if_exist:NF #1
1783 %^^A { \PackageError{thermodynamics}
1784 %^^A {Command~'\tl_trim_spaces:o {\tl_to_str:n {#1}}'~not~already~defined}
1785 %^^A {You~have~used~
1786 %^^A '\tl_trim_spaces:o
1787 %^^A {\tl_to_str:n {\RenewSubscriptedandSuperScriptedSymbol}}'~
1788 %^^A with~a~command~that~does~not~have~a~definition}
1789 %^^A }
1790 %^^A \@@_subscripted_and_superscripted_core:nnnn {#1} {#2} {#3} {#4}
1791 %^^A \RenewDocumentCommand{#1}{#1}
1792 %^^A {
1793 %^^A \peek_catcode_remove:NTF \c_math_superscript_token
1794 %^^A { \use:c {\cs_to_str:N #1_one:n} }
1795 %^^A { \peek_catcode_remove:NTF \c_math_subscript_token
1796 %^^A { \use:c {\cs_to_str:N #1_two:n} }
1797 %^^A { {#2}\c_math_subscript_token{#3}\c_math_superscript_token{#4} }
1798 %^^A }
1799 %^^A }
1800 }

```

B.9 Symbol Definitions

`\heatcapacitiesymbol` We define symbols for the heat capacities, compressibilities, and so forth.

```

\compressibilitysymbol 1801 \NewExpandableDocumentCommand{\heatcapacitiesymbol}{}{C}
\expansivitysymbol 1802 \NewExpandableDocumentCommand{\compressibilitysymbol}{}{\kappa}
1803 \NewExpandableDocumentCommand{\expansivitysymbol}{}{\alpha}
1804 \NewExpandableDocumentCommand{\Jtsymbol}{}{\mu}

```

`\cV` The heat capacities are molar by default; we also declare extensive and specific heat
`\cP` capacities. The heat capacities themselves are defined to be “smart”: `\cV_i` will
recognize the subscript appropriately and render $C_{V,i}$ rather than C_{Vi} or some other
unintended symbol. Superscripts are also handled appropriately and can be in either
order.

```

1805 \NewSubscriptedSymbol{\cV}{\@@_intensive:n \heatcapacitiesymbol}
1806 { \g_@@_volume_symbol }
1807 \NewSubscriptedSymbol{\cP}{\@@_intensive:n \heatcapacitiesymbol}
1808 { \g_@@_pressure_symbol }

```

`\cVt` We also introduce *extensive* (rather than molar) heat capacity macros.

```

\cPt 1809 \NewSubscriptedSymbol{\cVt}{\@@_extensive:n \heatcapacitiesymbol}
1810 { \g_@@_volume_symbol }
1811 \NewSubscriptedSymbol{\cPt}{\@@_extensive:n \heatcapacitiesymbol}
1812 { \g_@@_pressure_symbol }

```

`\cVs` `\cPs` and `\cVs` are the specific heat capacities.

`\cPs`

1813 \NewSubscriptedSymbol{\cVs}{\@@_specific:n \heatcapacitysymbol}
1814 {\g_@@_volume_symbol}
1815 \NewSubscriptedSymbol{\cPs}{\@@_specific:n \heatcapacitysymbol}
1816 {\g_@@_pressure_symbol}

\kappaT The isothermal and adiabatic compressibilities are defined similarly, but those do not
\kappaS have extensive versions for obvious reasons.

1817 \NewSubscriptedSymbol{\kappaT}{\compressibilitysymbol}
1818 {\g_@@_temperature_symbol}
1819 \NewSubscriptedSymbol{\kappaS}{\compressibilitysymbol}
1820 {\g_@@_entropy_symbol}

\alphaP The macro \alphaP is intended to refer to the isobaric volume expansivity, while
\alphaS \alphaS is the isentropic volume expansivity.

1821 \NewSubscriptedSymbol{\alphaP}{\expansivitysymbol}{\g_@@_pressure_symbol}
1822 \NewSubscriptedSymbol{\alphaS}{\expansivitysymbol}{\g_@@_entropy_symbol}

\muJT The macro \muJT renders the Joule–Thomson coefficient.

1823 \NewSubscriptedSymbol{\muJT}{\JTsymbol}{\text{JT}}

\Psat The \Psat macro (and its clone, the \Pvap macro) should be used for the saturation
\Pvap pressure. Similarly, a \phisat macro typesets the fugacity coefficient at saturation.
\phisat The \fsat macro similarly renders the fugacity at saturation. Other saturation prop-
\fsat erties should use M^\sat or similar, preferably by defining another macro such as
\sat \Gmsat.

1824 \NewDocumentCommand{\sat}{}{\text{sat}}
1825 \NewDocumentCommand{\Psat}{}{P\c_math_superscript_token\sat}
1826 \NewDocumentCommand{\Pvap}{}{\Psat}
1827 \NewDocumentCommand{\phisat}{}{\phi\c_math_superscript_token\sat}
1828 \NewDocumentCommand{\fsat}{}{f\c_math_superscript_token\sat}

\std The \std macro denotes standard properties. \Pstd, \Cstd, \fstd, and \mustd are
\Pstd defined for convenience and for consistency across textbooks.

\fstd 1829 \NewDocumentCommand{\std}{}{\circ}
\Cstd 1830 \NewDocumentCommand{\Pstd}{}{P\c_math_superscript_token \std}
\mustd 1831 \NewDocumentCommand{\Cstd}{}{C\c_math_superscript_token \std}
1832 \NewDocumentCommand{\fstd}{}{f\c_math_superscript_token \std}
1833 \NewDocumentCommand{\mustd}{}{\mu\c_math_superscript_token \std}

\rxn The \rxn macro denotes the extent of reaction.

1834 \NewDocumentCommand{\rxn}{}{\xi}

\fmix The \fmix command is intended to describe fugacities in mixtures. It renders as \hat{f} by
default, and would be used as \fmix_i or the like, producing \hat{f}_i ; some authors like to
use \bar{f}_i or just f_i , and this command creates a consistent way to change between such
options.

1835 \NewDocumentCommand{\fmix}{}{\hat{f}}

\phimix A similar command, \phimix, renders $\hat{\phi}$ by default to represent the fugacity coefficient
in a mixture.

1836 \NewDocumentCommand{\phimix}{}{\hat{\phi}}

`\fpure` The `\fpure` command is intended to describe fugacities in pure substances. It renders as f by default, and would be used as `\fpure` or `\fpure_i` or the like, producing f_i ; some authors like to use $f_{\text{pure } i}$, and others like to use f_i^\bullet ; this command creates a consistent way to change between the these options. A similar command for ϕ is given, `\hipure`, for fugacity coefficients.

```
1837 \NewDocumentCommand{\fpure}{}{f}
1838 \NewDocumentCommand{\hipure}{}{\phi}
```

`\actabs` The `\actabs` and `\actrel` macros are intended to standardize the macro for the
`\actrel` absolute and relative activities, respectively, across textbooks. The defaults are λ and a , respectively.

```
1839 \NewExpandableDocumentCommand{\actabs}{}{\lambda}
1840 \NewExpandableDocumentCommand{\actrel}{}{a}
```

B.9.1 Property Changes

`\Deltamix` Mixing properties, such as ΔG_{mix} , should be accessed using `\Deltamix\Gm` and sim-
`\mixing` ilar constructions—this construct will typeset as $\Delta_{\text{MIX}}G$ using the Thompson package option, for example, and as $\Delta_{\text{mix}}G$ using the Sandler package option.

```
1841 \NewDocumentCommand{\mixing}{}{\text{mix}}
1842 \NewDocumentCommand{\Deltamix}{m}
1843   {\Delta{#1}\c_math_subscript_token\mixing}
```

`\formation` Macros for subscripts/superscripts on property changes,

```
\fusion 1844 \NewDocumentCommand{\formation}{}{f}
\reaction 1845 \NewDocumentCommand{\reaction}{}{\text{rxn}}
\sublimation 1846 \NewDocumentCommand{\sublimation}{}{\text{sub}}
\vaporization 1847 \NewDocumentCommand{\vaporization}{}{\text{vap}}
1848 \NewDocumentCommand{\vaporization}{}{\text{vap}}
```

`\Deltafus` and similar macros for the property changes themselves. Some of these may be
`\Deltasub` redefined by package options that select a particular textbook.

```
\Deltavap 1849 \NewDocumentCommand{\Deltafus}{m}
\Deltarxn 1850   {\Delta #1\c_math_superscript_token\fusion}
1851 \NewDocumentCommand{\Deltasub}{m}
1852   {\Delta #1\c_math_superscript_token\sublimation}
1853 \NewDocumentCommand{\Deltavap}{m}
1854   {\Delta #1\c_math_superscript_token\vaporization}
1855 \NewDocumentCommand{\Deltarxn}{m}
1856 {
1857   \cs_set:Npn \@@_Deltarxn_one ##1
1858   {
1859     \Delta #1\c_math_subscript_token{reaction,##1}
1860   }
1861   \peek_catcode_remove:NTF \c_math_subscript_token
1862   {
1863     \@@_Deltarxn_one
1864   }
1865   {
1866     \Delta #1\c_math_subscript_token{reaction}
1867   }
1868 }
1869 \tl_new:N \l_@@_Deltaf_sym_tl
1870 \NewSubscriptedSymbol{\@@_Deltaf}{\l_@@_Deltaf_sym_tl}{\formation}
```



```

1871 \NewDocumentCommand{\Deltaf}{m}
1872 { \tl_set:Nn \l_@@_Deltaf_sym_tl {\Delta #1}
1873   \@@_Deltaf
1874 }

```

B.9.2 Partial Molar Quantities

`\partialmolar` Partial molar quantities with superscripts appear as $\overline{G}_i^{\text{IG}}$ or $\overline{G}_i^{\text{R}}$ (or, with the `shortpm` option, as $\overline{G}_i^{\text{IG}}$ or $\overline{G}_i^{\text{R}}$), rather than something like $\overline{G}_i^{\text{IG}}$ or $\overline{G}_i^{\text{R}}$; the former looks better but is harder to implement for obvious reasons. Their definitions allow them to be used as symbols, something like `\Gpm_i`, `\Gpm^{\IGM}_i`, `\Gpm{\IGM}_i`, and even `\Gpm_i^{\IGM}`; they can also be treated as commands: `\Gpm{i}` is equivalent to `\Gpm_i` and `\Gpm[\IGM]{i}` is equivalent to `\Gpm_i^{\IGM}`. The macro `\partialmolar` can be used to create an arbitrary partial molar symbol.

```

1875 \tl_new:N \l_@@_pm_symbol_tl
1876 \tl_new:N \l_@@_pm_arg_tl
1877 \NewDocumentCommand{\partialmolar}{m}
1878 {
1879   \tl_set:Nn \l_@@_pm_symbol_tl {#1}
1880   \@@_generic_pm:
1881 }
1882 %% cases to consider:
1883 %% (1) \Mpm{i}
1884 %% (2) \Mpm[S]{i}
1885 %% (3) \Mpm^S_i
1886 %% (4) \Mpm_i^S
1887 %% (5) \Mpm_i
1888 %% note that \Mpm^S with no subscript makes no sense and is thus forbidden
1889 \cs_new:Nn \@@_generic_pm:
1890 {
1891   \peek_catcode_remove:NTF \c_math_subscript_token
1892   {% case 4 or case 5
1893     \@@_pm_case_four_or_five
1894   }
1895   {% look for superscript token
1896     \peek_catcode_remove:NTF \c_math_superscript_token
1897     {% case 3: \Mpm^{#1}_{#2} or \Mpm^{#1}{#2}
1898       \@@_pm_case_three
1899     }
1900     {% Look for optional argument [...]
1901       \peek_charcode:NTF [
1902         {% case 2: \Mpm[S]{i}
1903           \@@_pm_case_two
1904         }
1905         {% case 1: \Mpm{i}
1906           \@@_pm_case_one
1907         }
1908       }
1909     }
1910 }
1911 \cs_new:Npn \@@_pm_case_one #1
1912 {
1913   \bool_if:NTF \l_@@_longpm_bool
1914   { \@@_overline:n {\l_@@_pm_symbol_tl\c_math_subscript_token{#1}} }

```

```

1915 {
1916   \@@_overline:n {\l_@@_pm_symbol_tl}\c_math_subscript_token{#1}
1917 }
1918 }
1919 \cs_new:Npn \@@_pm_case_two [#1]#2
1920 {
1921   \bool_if:NTF \l_@@_longpm_bool
1922   { \@@_overline:n {\l_@@_pm_symbol_tl
1923     \c_math_superscript_token{#1}\c_math_subscript_token{#2}}
1924   }
1925   { \@@_overline:n {\l_@@_pm_symbol_tl
1926     \c_math_superscript_token{#1}\c_math_subscript_token{#2}}
1927   }
1928 }
1929 \cs_new:Npn \@@_pm_case_three #1
1930 {
1931   \tl_set:Nn \l_@@_pm_arg_tl {#1}
1932   \peek_catcode_remove:NTF \c_math_subscript_token
1933   { \@@_pm_case_three_part_two }
1934   { \@@_pm_case_three_part_two }
1935 }
1936 \cs_new:Npn \@@_pm_case_three_part_two #1
1937 {
1938   \bool_if:NTF \l_@@_longpm_bool
1939   { \@@_overline:n {\l_@@_pm_symbol_tl
1940     \c_math_superscript_token{\l_@@_pm_arg_tl}
1941     \c_math_subscript_token{#1}}
1942   }
1943   { \@@_overline:n {\l_@@_pm_symbol_tl
1944     \c_math_superscript_token{\l_@@_pm_arg_tl}
1945     \c_math_subscript_token{#1}}
1946   }
1947 }
1948 \cs_new:Npn \@@_pm_case_four_or_five #1
1949 {
1950   \tl_set:Nn \l_@@_pm_arg_tl {#1}
1951   \peek_catcode_remove:NTF \c_math_superscript_token
1952   { \@@_pm_case_four }
1953   { \@@_pm_case_five }
1954 }
1955 \cs_new:Npn \@@_pm_case_four #1
1956 {
1957   \bool_if:NTF \l_@@_longpm_bool
1958   { \@@_overline:n {\l_@@_pm_symbol_tl\c_math_superscript_token{#1}
1959     \c_math_subscript_token{\l_@@_pm_arg_tl}}
1960   }
1961   { \@@_overline:n {\l_@@_pm_symbol_tl}\c_math_superscript_token{#1}
1962     \c_math_subscript_token{\l_@@_pm_arg_tl}
1963   }
1964 }
1965 \cs_new:Npn \@@_pm_case_five
1966 {
1967   \bool_if:NTF \l_@@_longpm_bool
1968   { \@@_overline:n {\l_@@_pm_symbol_tl
1969     \c_math_subscript_token{\l_@@_pm_arg_tl}}
1970   }

```

```

1971 { \@@_overline:n {\l_@@_pm_symbol_tl}
1972     \c_math_subscript_token{\l_@@_pm_arg_tl}
1973 }
1974 }

```

B.9.3 Thermodynamic Property Definitions

These macros define the user interface to the symbols for energy, volume, and so forth. There are five commands that define thermodynamic properties.

`\NewExtensiveProperty` The command `\NewExtensiveProperty` declares macros for a total, molar, and specific version of the symbol; for example, a second heat-like property could be defined via

```
\NewExtensiveProperty{R}{\mathcal{Q}}
```

The command above would declare the macros `\Rt`, `\Rm`, and `\Rs` that expand to \underline{Q} , Q , and \hat{Q} , respectively, using the default package options.

`\NewPartialMolarProperty` The command `\NewPartialMolarProperty` declares a macro for the partial molar quantity. For example,

```
\NewPartialMolarProperty{I}{\Psi}
```

would create the command `\Ipm`, which would typeset a partial molar command with the base symbol Ψ , yielding $\bar{\Psi}_i$.

`\NewThermodynamicProperty` Declaring a new potential is handled by the `\NewThermodynamicProperty` macro, which takes two arguments. The first is the base of the name, and the second is the base of the symbol. This declares four new commands for the extensive, molar, specific, and partial molar properties. These commands consist of the first argument followed by `t`, `m`, `s`, and `pm`, respectively. For example, one might define the entropy via

```
\NewThermodynamicProperty{S}{S}
```

and it would define the macros `\St`, `\Sm`, `\Ss`, and `\Spm` that yield, respectively, \underline{S} , S , \hat{S} , and \bar{S}_i (assuming the subscript to the partial molar quantity was i). It would also declare residual and excess properties for that base symbol. Note that the actual definition of the entropy and the other standard properties is slightly more complicated so as to allow for different symbols to be used in different textbooks.

```

1975 \NewDocumentCommand{\NewThermodynamicProperty}{m m}
1976 {
1977   \NewExtensiveProperty{#1}{#2}
1978   \NewPartialMolarProperty{#1}{#2}
1979   \NewResidualProperty{#1}{#2}
1980   \NewExcessProperty{#1}{#2}
1981 }
1982 \ProvideDocumentCommand{\NewExtensiveProperty}{m m}
1983 {
1984   % Extensive property
1985   \exp_after:wN \NewDocumentCommand \exp_after:wN
1986     {\cs:w #1t\cs_end:}{\@@_extensive:n {#2}}
1987   % Molar property
1988   \exp_after:wN \NewDocumentCommand \exp_after:wN
1989     {\cs:w #1m\cs_end:}{\@@_intensive:n {#2}}
1990   % Specific property

```

```

1991 \exp_after:wN \NewDocumentCommand \exp_after:wN
1992   {\cs:w #1s\cs_end:}\{\@@_specific:n {#2}}
1993 }
1994 \NewDocumentCommand{\NewPartialMolarProperty}{m m}
1995 {
1996   % Partial molar property
1997   \exp_after:wN \NewDocumentCommand \exp_after:wN
1998     {\cs:w #1pm\cs_end:}\{\partialmolar{#2}}
1999 }
2000 \ProvideDocumentCommand{\NewExcessProperty}{m m}
2001 {
2002   \exp_after:wN \NewSuperscriptedSymbol \exp_after:wN
2003     {\cs:w #1 Et\cs_end:}\{\@@_extensive:n{#2}}{\excess}
2004   \exp_after:wN \NewSuperscriptedSymbol \exp_after:wN
2005     {\cs:w #1 E\cs_end:}\{\@@_intensive:n{#2}}{\excess}
2006   \exp_after:wN \NewSuperscriptedSymbol \exp_after:wN
2007     {\cs:w #1 Es\cs_end:}\{\@@_specific:n{#2}}{\excess}
2008
2009   % Excess partial molar property
2010   \exp_after:wN \NewDocumentCommand \exp_after:wN
2011     {\cs:w #1Epm\cs_end:}\{\partialmolar{#2}
2012       \c_math_superscript_token\excess}
2013 }
2014 \ProvideDocumentCommand{\NewResidualProperty}{m m}
2015 {
2016   \exp_after:wN \NewSuperscriptedSymbol \exp_after:wN
2017     {\cs:w #1 Rt\cs_end:}\{\@@_extensive:n{#2}}{\residual}
2018   \exp_after:wN \NewSuperscriptedSymbol \exp_after:wN
2019     {\cs:w #1 R\cs_end:}\{\@@_intensive:n{#2}}{\residual}
2020   \exp_after:wN \NewSuperscriptedSymbol \exp_after:wN
2021     {\cs:w #1 Rs\cs_end:}\{\@@_specific:n{#2}}{\residual}
2022
2023   % Residual partial molar property
2024   \exp_after:wN \NewDocumentCommand \exp_after:wN
2025     {\cs:w #1Rpm\cs_end:}\{\partialmolar{#2}
2026       \c_math_superscript_token\residual}
2027 }

```

\Nt We define \Nt as the number of moles, as that changes between books a lot, but it
\Et does not have extensive, molar, and specific equivalents. Heat and work are defined,
\Em but lack partial molar properties. Area does not have excess or residual properties.

```

\Es 2028 \NewDocumentCommand{\Nt}\{\g_@@_mole_symbol}
\Epm 2029 \NewThermodynamicProperty{E}\{\g_@@_total_energy_symbol}
2030 \NewThermodynamicProperty{U}\{\g_@@_internal_energy_symbol}
2031 \NewThermodynamicProperty{F}\{\g_@@_Helmholtz_symbol}
2032 \NewThermodynamicProperty{G}\{\g_@@_Gibbs_symbol}
2033 \NewThermodynamicProperty{H}\{\g_@@_enthalpy_symbol}
2034 \NewThermodynamicProperty{L}\{\g_@@_Landau_symbol}
2035 \NewThermodynamicProperty{V}\{\g_@@_volume_symbol}
2036 \NewThermodynamicProperty{S}\{\g_@@_entropy_symbol}
2037 \NewExtensiveProperty{A}\{\g_@@_area_symbol}
2038 \NewPartialMolarProperty{A}\{\g_@@_area_symbol}
2039 \NewExtensiveProperty{Q}\{\g_@@_heat_symbol}
2040 \NewExtensiveProperty{W}\{\g_@@_work_symbol}

```

\cVpm Partial molar heat capacities are *hard*, but the following implementation seems to
\cPpm

work flawlessly... so far.

```

2041 \NewPartialMolarProperty{cP}{\cP}
2042 \NewPartialMolarProperty{cV}{\cV}
2043 \NewSubscriptedSymbol{\cPpmsymbol}{\@@_overline_copy:n{\heatcapacitysymbol}}
2044     {\g_@@_pressure_symbol}
2045 \NewSubscriptedSymbol{\cVpmsymbol}{\@@_overline_copy:n{\heatcapacitysymbol}}
2046     {\g_@@_volume_symbol}

```

B.9.4 Electrochemistry

`\Epot` The `\Epot` command can be used for the electric potential.

```

2047 \NewExpandableDocumentCommand{\Epot}{}{E}

```

B.9.5 Residual and Excess Properties

`\residual` Macros are defined for residual properties (departure from ideal gases) and excess properties (departure from ideal solutions). We begin with two macros to use for defining generic residual and excess properties that are not already defined.

```

2048 \NewDocumentCommand{\residual}{}{R}
2049 \NewDocumentCommand{\excess}{}{E}

```

`\prodall` The `\sumall` macro and its cousin, `\sumallbutlast`, simplify the typesetting of `\sumall` commonly-used sums; the command `\prodall` does the same thing for products, `\sumallbutlast` viz.,

```

\[\sumall_i x_i = 1 \quad \sumallbutlast_i x_i = 1 - x_{\ncomponents}
\quad K = \exp\left(\frac{-\Delta G_{\text{rxn}}}{RT}\right)
= \prodall_i a_i^{\nu_i} \quad \]

```

gives

$$\sum_{i=1}^C x_i = 1 \quad \sum_{i=1}^{C-1} x_i = 1 - x_C \quad K = \exp\left(\frac{-\Delta G_{\text{rxn}}}{RT}\right) = \prod_{i=1}^C a_i^{\nu_i}$$

```

2050 \NewDocumentCommand{\sumall}{m m}
2051 { \sum\c_math_subscript_token{#2=1}
2052     \c_math_superscript_token{\ncomponents} }
2053 \NewDocumentCommand{\sumallbutlast}{m m}
2054 { \sum\c_math_subscript_token{#2=1}
2055     \c_math_superscript_token{\ncomponents-1} }
2056 \NewDocumentCommand{\prodall}{m m}
2057 { \prod\c_math_subscript_token{#2=1}
2058     \c_math_superscript_token{\ncomponents} }

```

`\IG` The `\IG`, `\IGM`, and `\IS` macros (meaning “ideal gas,” “ideal gas mixture,” and “ideal `\IGM` solution,” respectively) should be used to make clean transitions between textbooks—`\IS` some use “IM” or “ID” rather than “IS,” for example.

```

2059 \NewDocumentCommand{\IG}{}{\text{IG}}
2060 \NewDocumentCommand{\IGM}{}{\text{IGM}}
2061 \NewDocumentCommand{\IS}{}{\text{IS}}

```

`\Henryrat` **B.9.6 Henry’s Law Constants and Activity Coefficients**

`\Henrymol` The Henry’s Law constants for the rational basis ($y_iP = x_ih_i$) and the molal basis ($y_iP = C_i\mathcal{H}_i$) are given by the macros `\Henryrat` and `\Henrymol`, respectively. Using them this way consistently allows for easy switching back and forth.

```
2062 \NewDocumentCommand{\Henryrat}{}{h}
```

```
2063 \NewDocumentCommand{\Henrymol}{}{\mathcal{H}}
```

`\gammait` The ordinary activity coefficient is typically denoted γ or γ , so the base symbol will still be `\gamma`. The macro `\gammait` is defined just in case `\gamma` is redefined by a package option. However, symbols for the Henry’s Law activity coefficients are far from universal, so I have defined macros to make their use consistent. The defaults render `\gammarat` as γ^* and `\gammamol` as γ^\square . `\gammamol` will use `\square` from packages if it is defined; if not, it “fakes it” with the definition below. The symbol for the chemical potential is `\mu`, but `\mu` is defined so the symbol is still available, should `\mu` be redefined later.

```
2064 \cs_gset_eq:NN \gammait \gamma
```

```
2065 \cs_gset_eq:NN \mu \mu
```

```
2066 \AtBeginDocument{%
```

```
2067   \ProvideDocumentCommand{\square}{}{%
```

```
2068     \text{\leavevmode
```

```
2069       \hbox to.65em{%
```

```
2070         \hfil\vrule
```

```
2071         \vbox to.53em{\hrule width.45em\vfil\hrule}%
```

```
2072         \vrule\hfil}%
```

```
2073     }%
```

```
2074   }%
```

```
2075 }
```

```
2076 \NewDocumentCommand{\gammarat}{}{\gamma\c_math_superscript_token\ast}
```

```
2077 \NewDocumentCommand{\gammamol}{}{\gamma\c_math_superscript_token\square}
```

B.10 Jacobians

`thermovmatrix (env.)` This is a non-user-interface wrapper environment used to detect, in effect, whether `amsmath` has been loaded. If so, it uses its `vmatrix` environment for Jacobians; if not, it fakes it with `array` (which does not look nearly as good).

```
2078 \NewDocumentEnvironment{thermovmatrix}{}{
```

```
2079 { \cs_if_exist:NTF \vmatrix
```

```
2080 { \begin{vmatrix} }
```

```
2081 { \left|\begin{array}{c c c c c c c c c c} }
```

```
2082 }
```

```
2083 { \cs_if_exist:NTF \endvmatrix
```

```
2084 { \end{vmatrix} }
```

```
2085 { \end{array}\right| }
```

```
2086 }
```

`Jacobian` The `\Jacobian` command typesets the Leibnitz notation for the Jacobian determinant.

```
2087 \NewDocumentCommand{\Jacobian}{m m}
```

```
2088 { \@@_frac:nn{\partial(\#1)}{\partial{(\#2)}} }
```

`Jacobiandet` Similarly, the `\Jacobiandet` macro typesets the actual determinant that the Jacobian notation represents.

```
2089 \NewDocumentCommand{\Jacobiandet}{0{} 0{} m m}
```

```
2090 {
```

```

2091 \@@_Jacobian_set_ncomponents:nn {#3} {#4}
2092 \begin{thermovmatrix}
2093 \@@_Jacobianmatrix:nnnn {#1} {#2} {#3} {#4}
2094 \end{thermovmatrix}
2095 }
2096 \seq_new:N \l_@@_row_seq
2097 \seq_new:N \l_@@_matrix_seq
2098 \clist_new:N \l_@@_other_vars_clist
2099 \clist_new:N \l_@@_other_vars_copy_clist
2100 \tl_new:N \l_@@_Jacobian_x_tl
2101 \tl_new:N \l_@@_Jacobian_n_tl
2102 \tl_new:N \l_@@_Jacobian_temp_tl
2103 \bool_new:N \l_@@_found_dots_bool
2104 \cs_new:Nn \@@_Jacobian_set_ncomponents:nn
2105 {
2106 % If any entry is \dots, we assume the Jacobian is of the form
2107 %  $d(f_1, \dots, f_n)/d(x_1, \dots, x_n)$  where  $f$  is some function
2108 % (any symbol) and  $x$  is some variable (any symbol).
2109 \tl_if_in:nnTF {#1} {\dots}
2110 {% Has dots
2111 \bool_set_true:N \l_@@_found_dots_bool
2112 % look for what "x" is
2113 \tl_set:Nn \l_@@_Jacobian_x_tl {\tl_head:n {#2}}
2114 % look for what "n" is and set \ncomponents to it
2115 \tl_set:Nx \l_@@_Jacobian_n_tl {\tl_item:nn {#2} {-1}}
2116 \RenewExpandableDocumentCommand{\ncomponents}{}{\l_@@_Jacobian_n_tl}
2117 }
2118 {% Does not have dots; proceed accordingly
2119 \bool_set_false:N \l_@@_found_dots_bool
2120 }
2121 }
2122 \cs_new_protected:Nn \@@_Jacobianmatrix:nnnn
2123 {
2124 \seq_clear:N \l_@@_matrix_seq
2125 \clist_set:Nn \l_@@_other_vars_clist {#4}
2126 \clist_set_eq:NN \l_@@_other_vars_copy_clist \l_@@_other_vars_clist
2127
2128 \clist_map_inline:nn {#3}
2129 {
2130 \seq_clear:N \l_@@_row_seq
2131 \tl_if_in:nnTF {##1} {\dots}
2132 {% The current row has "dots" => row is \vdots && \vdots
2133 \seq_put_right:Nn \l_@@_matrix_seq
2134 { \vdots \c_alignment_token \c_alignment_token \vdots }
2135 }
2136 {% Ordinary row
2137 \clist_map_inline:nn {#4}
2138 {
2139 \tl_if_in:nnTF {####1} {\dots}
2140 {% this column has "dots" in it
2141 \seq_put_right:Nn \l_@@_row_seq \dots
2142 }
2143 {% Normal column
2144 \clist_set_eq:NN \l_@@_other_vars_clist
2145 \l_@@_other_vars_copy_clist
2146 \clist_remove_all:Nn \l_@@_other_vars_clist {####1}

```

```

2147 \bool_if:NTF \l_@@_found_dots_bool
2148 { \tl_set:Nn \l_@@_Jacobian_temp_tl {\tl_item:nn {####1} {-1}}
2149 \seq_put_right:Nx \l_@@_row_seq
2150 {
2151 #1\Partial{##1}{####1}
2152 {\allbut{\l_@@_Jacobian_temp_tl}{\l_@@_Jacobian_x_tl}}
2153 }
2154 }
2155 { \seq_put_right:Nx \l_@@_row_seq
2156 {
2157 #1\Partial{##1}{####1}
2158 {\clist_use:Nn \l_@@_other_vars_clist ,}
2159 }
2160 }
2161 }
2162 }
2163 \seq_put_right:Nx \l_@@_matrix_seq
2164 {
2165 \seq_use:Nn \l_@@_row_seq { \c_alignment_token }
2166 }
2167 }
2168 }
2169 \tl_if_empty:nTF {#2}
2170 {
2171 \tl_if_eq:nnTF {#1} {\displaystyle}
2172 { \seq_use:Nn \l_@@_matrix_seq { \[2.75ex] } }
2173 { \seq_use:Nn \l_@@_matrix_seq { \[1.25ex] } }
2174 }
2175 {
2176 \seq_use:Nn \l_@@_matrix_seq { \[#2] }
2177 }
2178 }

```

Change History

v1.00		optional argument that changes N_i to N_j , say, when using TesterModell or other options that denote moles of all components that way. Similar updates to <code>\allNs</code> and friends.	33
	General: Initial public release	1	
v1.01			
	General: Added <code>\Partialinline</code> and friends to facilitate in-line (non-display-mode) partial derivatives, with corresponding changes to <code>\Partial</code> and friends for ease of implementation. Also added <code>\Partialinlinetext</code> and friends for non-expanding delimiters.	55	
	Changed options with two E or two A variables to use calligraphic letters for the less-common of the two.	28	
	<code>\allcomponents</code> : Updated <code>\allcomponents</code> to include an		
			<code>\Partial</code> : Changed <code>\adjust@width</code> 0.1\operator@width (from 2pt). . .
			Changed <code>\adjust@width</code> to 2pt (up from 1pt).
			51
		v2.00	
		General: Added <code>\muJT</code> to represent Joule–Thomson coefficients (which have different notation across textbooks).	6
		Added a sorting routine to make function arguments be in a consistent order using the	

nosubscripts option.	50	option.	34
Revision to use L ^A T _E X3		Improved consistency of package	
(expl3/xparse) syntax layer.	1	options for textbooks and added a	
Jacobian: Reimplemented \Jacobian		table of symbols organized by	
to handle an arbitrary number of		textbook option.	21
variables and implemented		\actrel: Added \actabs and \actrel	
\Jacobianet to handle the matrix		to make absolute and relative	
representation of the Jacobian.	70	activity symbols consistent.	64
\NewSubscriptedSymbol: Deleted		\cPpm: Simplified implementations of	
\DeclareSubscriptedSymbol in		partial molar heat capacities that	
favor of xparse-based		take advantage of	
\NewSubscriptedSymbol and		\NewPartialMolarProperty and	
\RenewSubscriptedSymbol.	58	\NewSubscriptedProperty. This	
\NewSuperscriptedSymbol: Created		fixes the issue of the subscripts	
\NewSuperscriptedSymbol to		being “persistent” after \cPpm is	
handle superscripted excess and		used in a line. Also declared	
residual properties without		alternative symbols for use with	
intervention.	60	the shortpm option.	69
\Partial: Changed length added to		\Deltarxn: Changed implementation	
\l_@@_Partial_const_dim from		of \Deltarxn to handle subscripts.	64
−0.15 to −0.20.	51	\Epot: Added \Epot to standardize	
thermoextensiveplain: Added		the electric potential across	
environment to invoke the		textbooks for electrochemistry.	69
“extensive-plain” option locally.	57	\muit: Added \muit and \gammat to	
thermoextensivesuperscript:		preserve symbols in case the base	
Added environment to invoke the		macros are redefined by a package.	70
“extensive-superscript” option		\mustd: Added \Cstd and \mustd to	
locally.	58	standardize standard molalities	
thermointensivelowercase: Added		and chemical potentials across	
environment to invoke the		textbooks.	63
“intensive-lowercase” option		thermoshortpm: Added “longpm”	
locally.	57	and “shortpm” options to control	
thermointensivplain: Added		how partial molar properties are	
environment to invoke the		rendered along with the	
“intensive-plain” option locally.	57	environments thermolongpm and	
thermomolesrange: Added		thermoshortpm to change them	
environment to invoke the		locally.	31
“moles-range” definitions of		\xrxn: Added \xrxn macro to	
\allNs and friends locally.	57	standardize the extent of reaction	
v2.01		macro across textbooks.	63
\NewSuperscriptedSymbol: Fixed		v2.03	
two typos in		General: Changed contact	
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